



UCLA

Amphidynamic Materials — Part II

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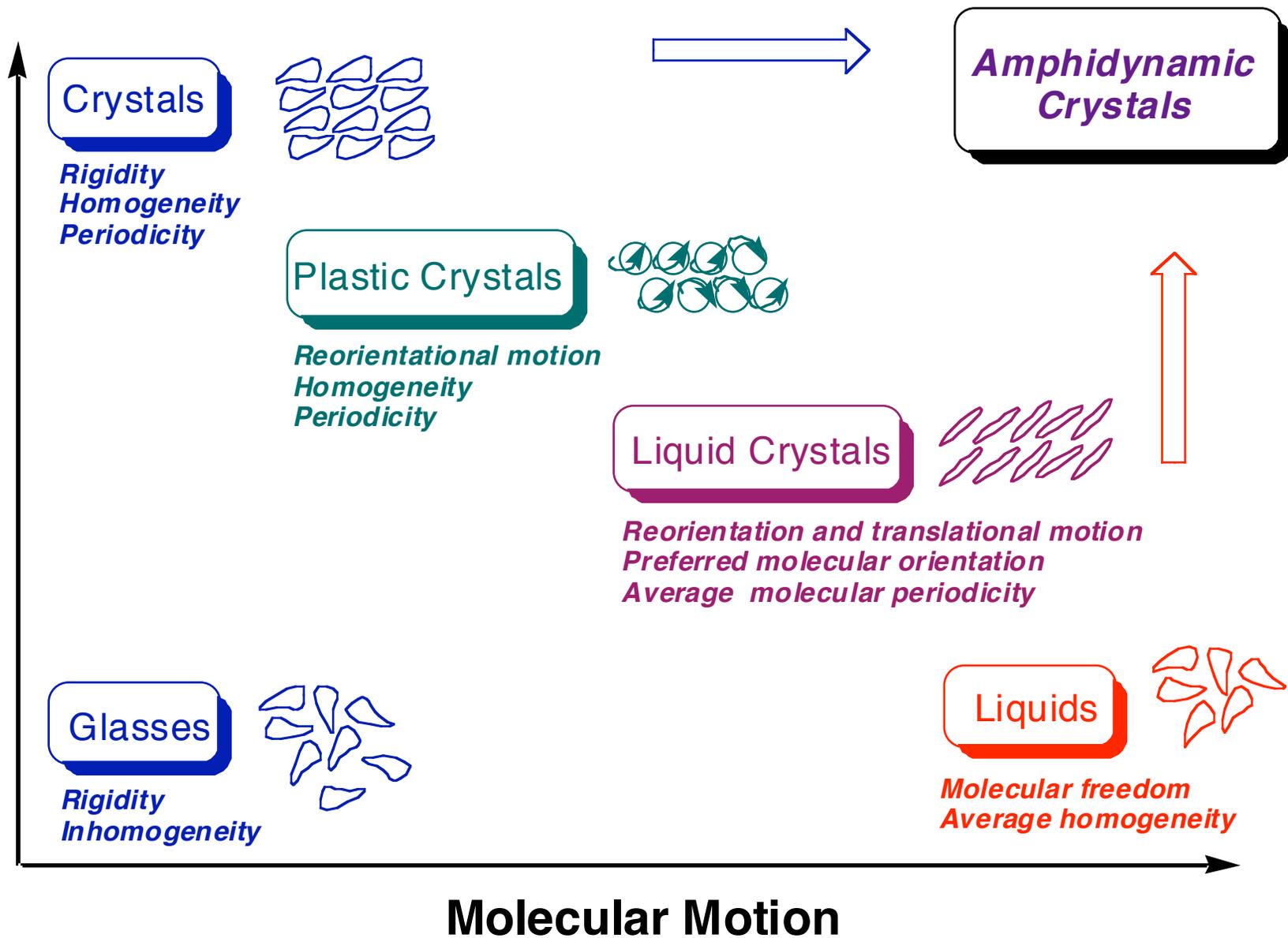
CINEVSTAV-IPN

Prof. Norberto Farfan

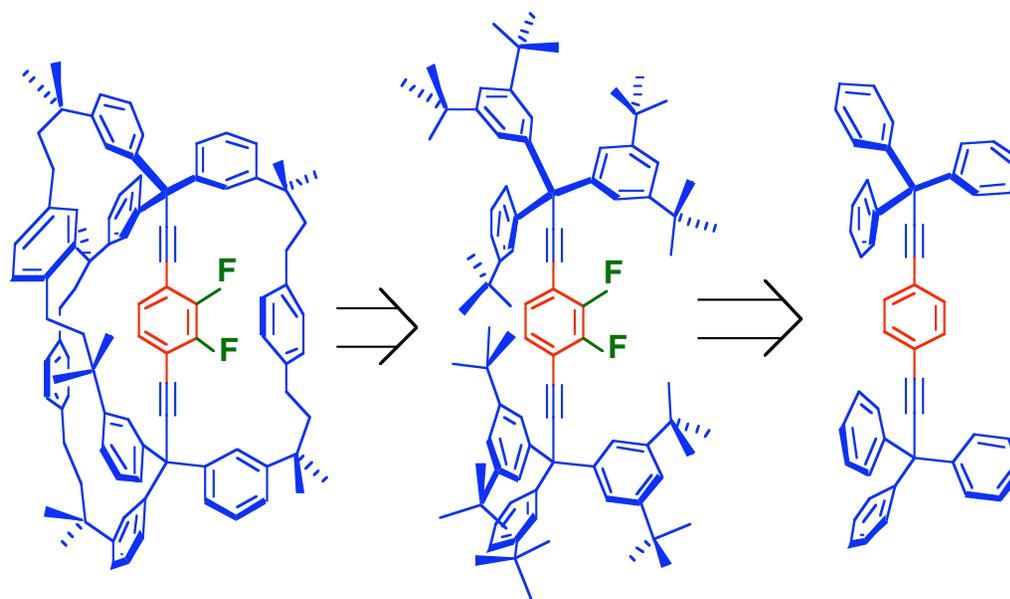
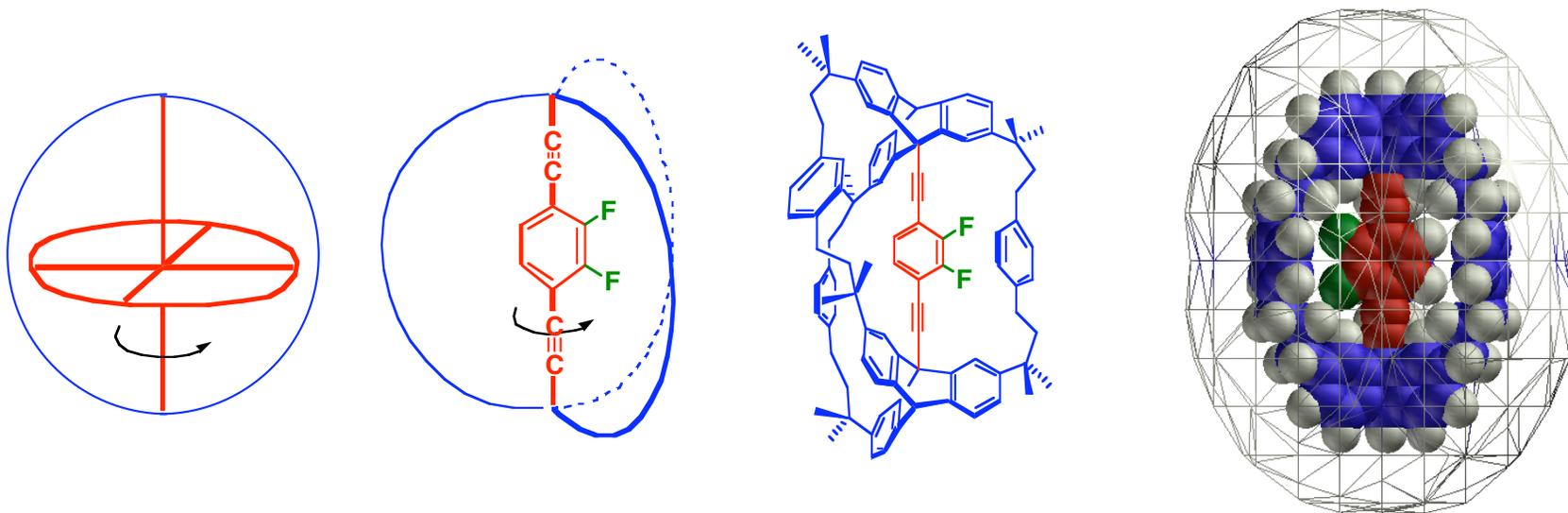
Pprof. Rosa Santillan

*National Science
Foundation*

Condensed Phase Matter and Molecular Dynamics



Molecular Compasses and Gyroscopes



Gyroscopic Motion in Crystals

Solid State NMR

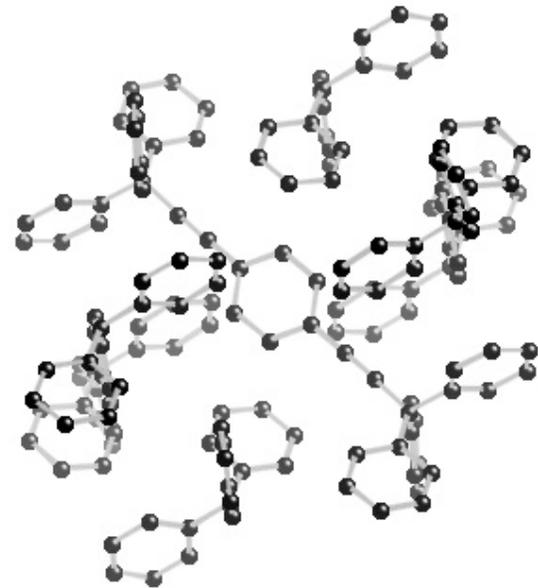
X-Ray Diffraction

Dielectric Spectroscopy

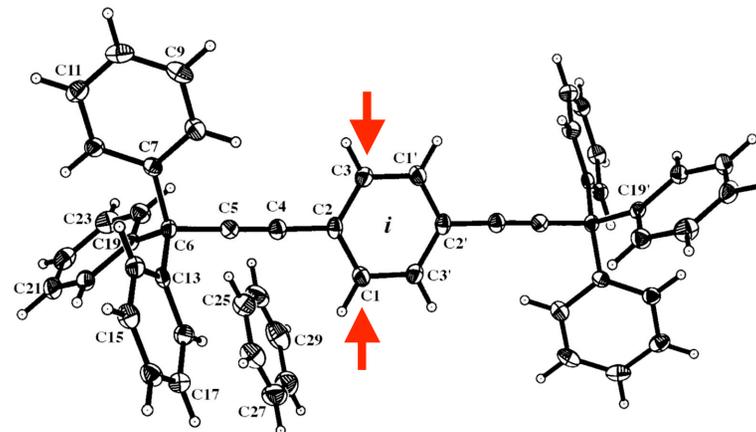
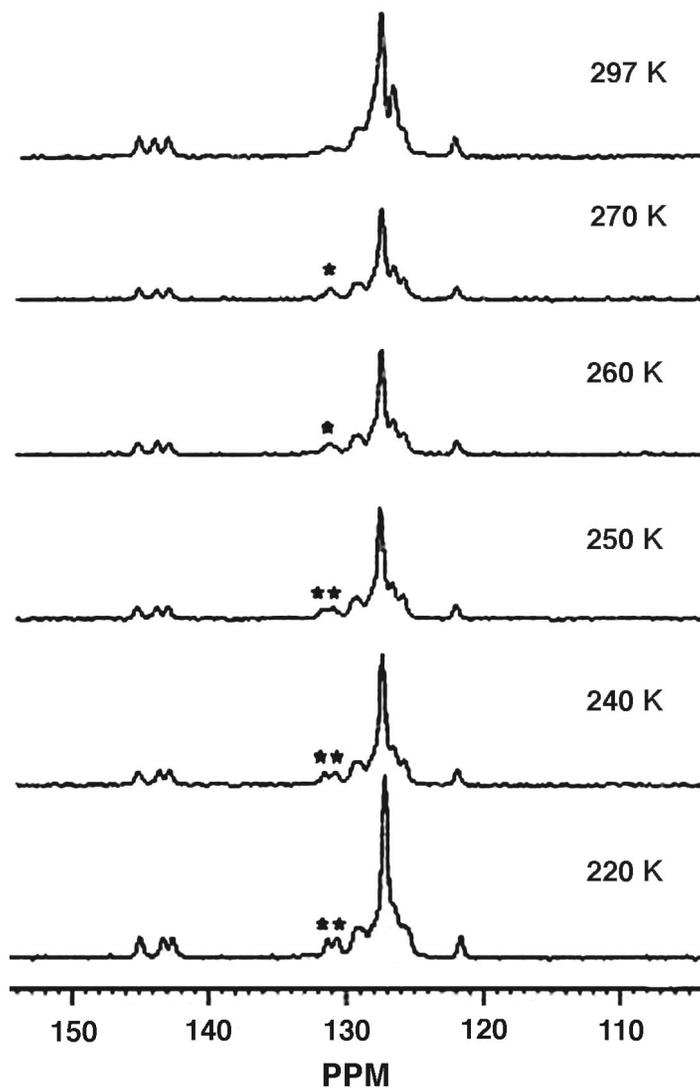
Computer Modeling

Fluorescence Anisotropy Decay

Inelastic Neutron Scattering



Dynamic NMR: VT ^{13}C CP-MAS

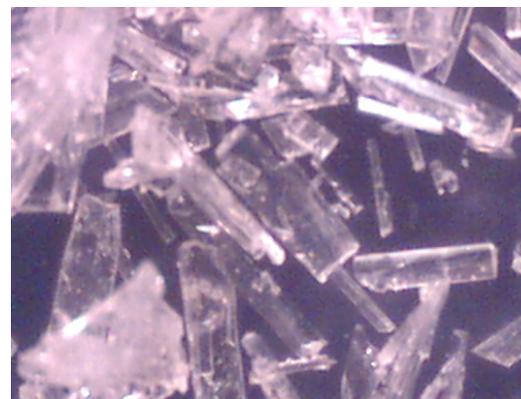


$\square\square = 60\text{Hz}$

$T_C = -16\text{ }^\circ\text{C}$

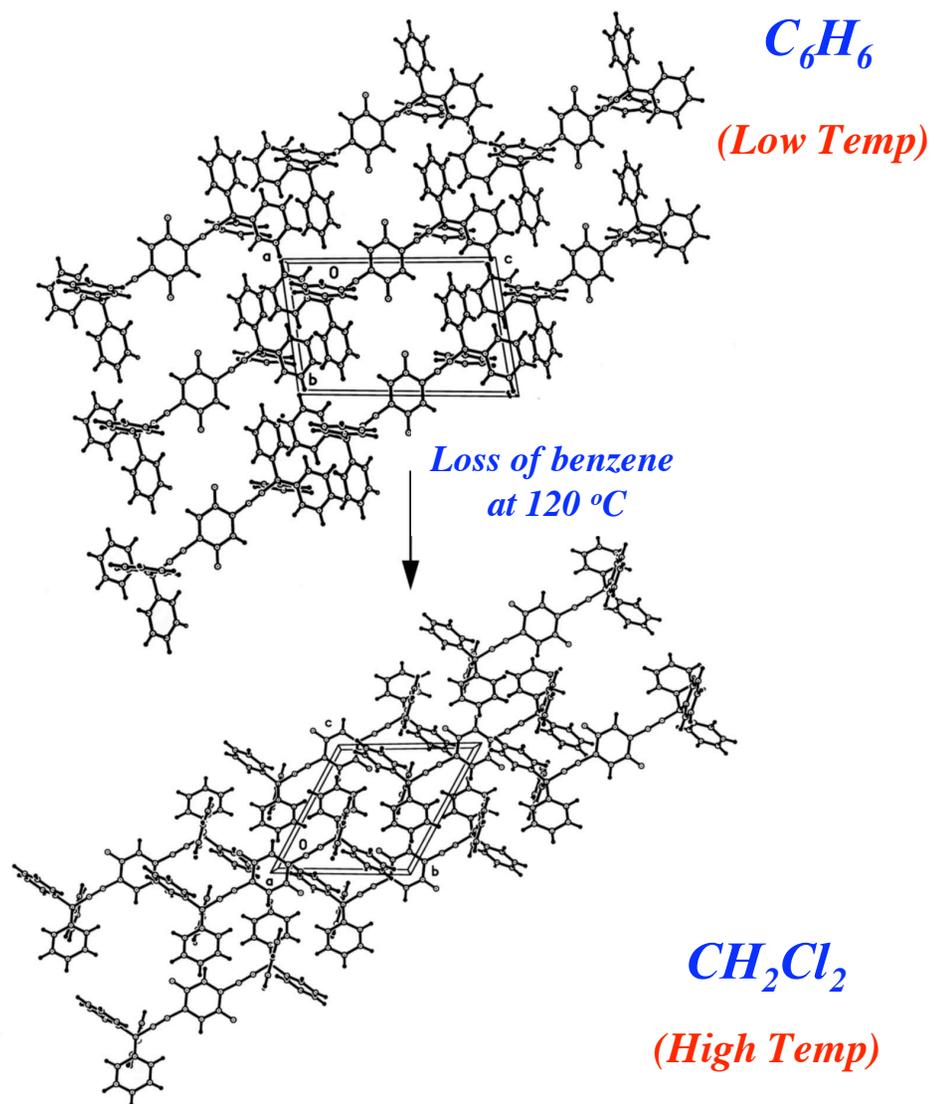
$k_{\text{rot}} = 267\text{ s}^{-1}$

$E_a = 12.8\text{ kcal/mol}$



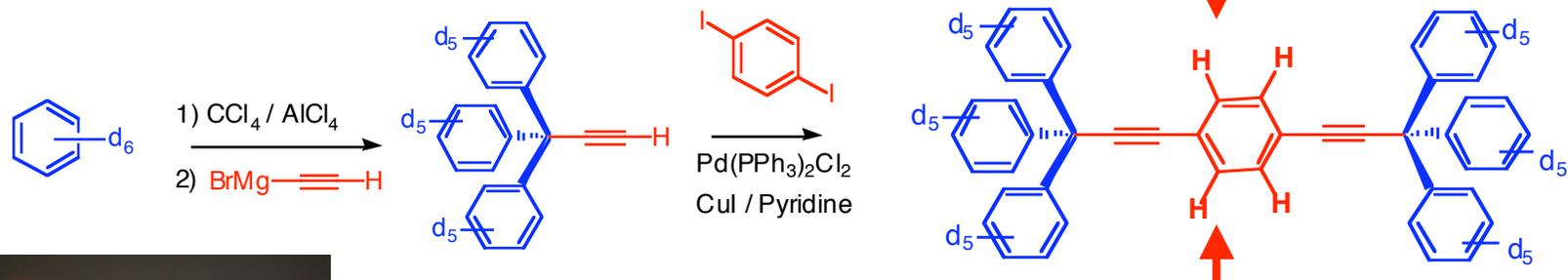
at T_C : $k_{\text{exch}} = k_{\text{Rot}} = \square\square(2)^{1/2}$
 $E_a = RT_C [22.96 + \text{Ln} (T_C / \square\square)]$

Two Pseudopolymorphs



Solid State Dynamics

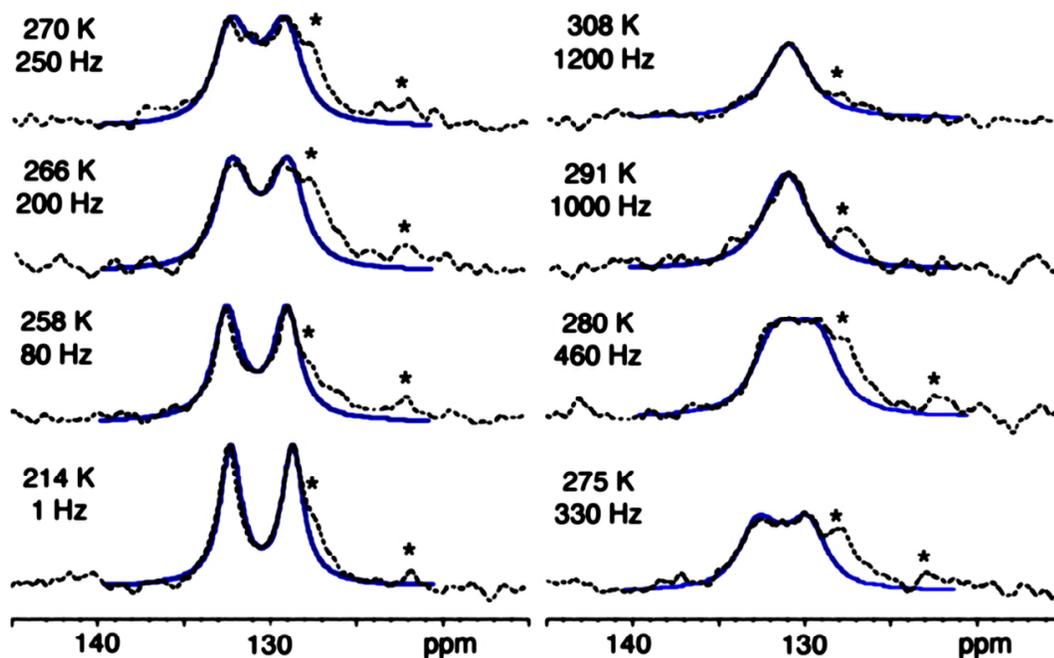
d_{30} -Trityl- h_4 -phenylene



$$k_{280} = 611.6 \text{ Hz}$$

$$\Delta_a = 12.8 \text{ kcal/mol}$$

*Faster rotation needs heating...
and a different method of measure!*

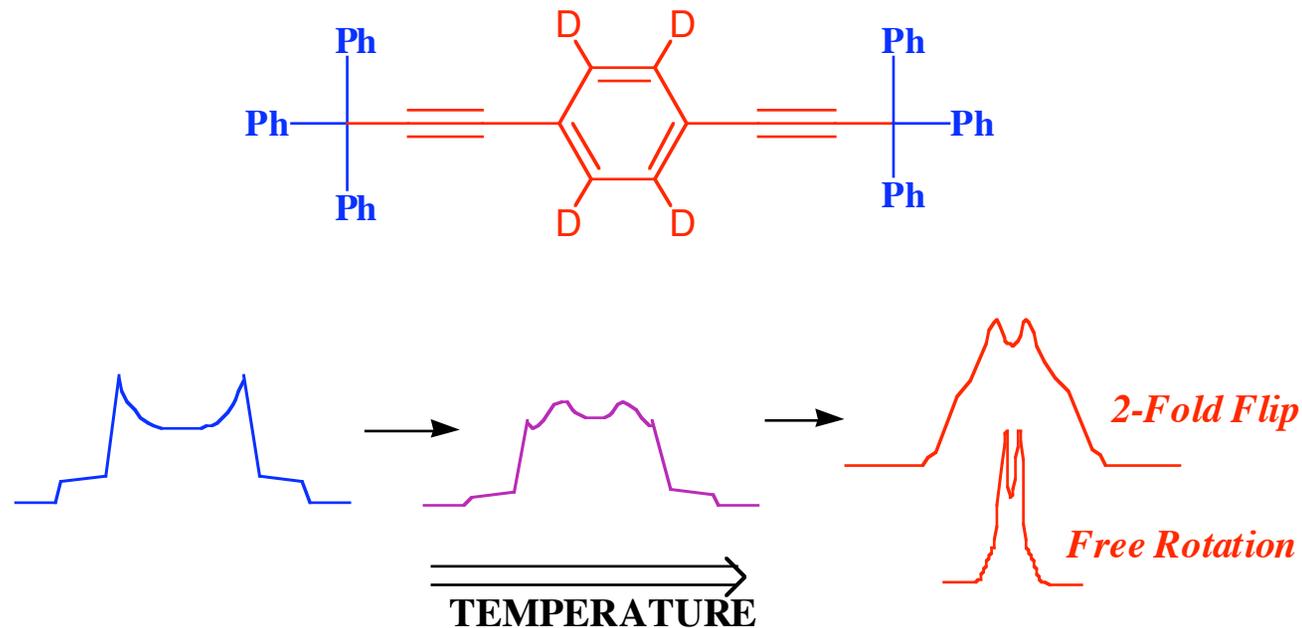


Steve Karlen

Solid State Dynamics

^2H NMR Line Shape Analysis ($10^4 - 10^8 \text{ s}^{-1}$)

$$2\nu_Q = (3/4)(e^2q_{zz}Q/h) [(3\cos^2\theta - 1) + \eta \sin^2\theta \cos^2\phi]$$



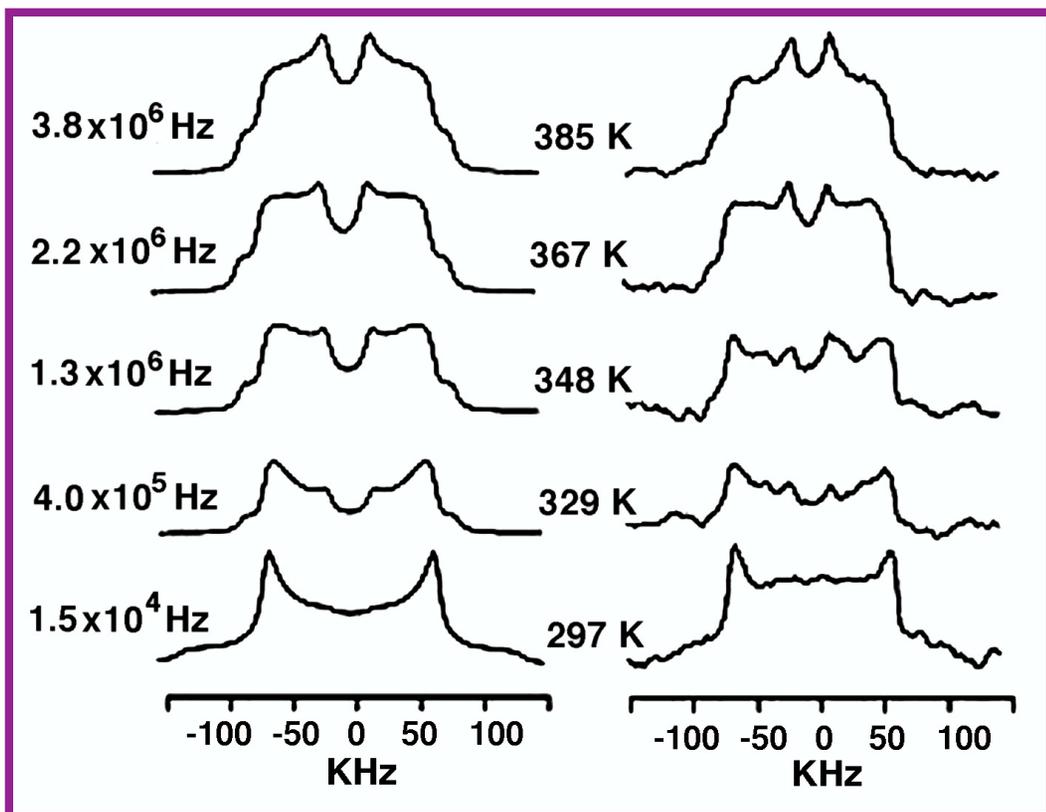
Dynamic averaging of the orientation-dependent interaction between the ^2H nuclear spin and the electric field gradient at the nuclear position.

Solid State Dynamics by ^2H NMR

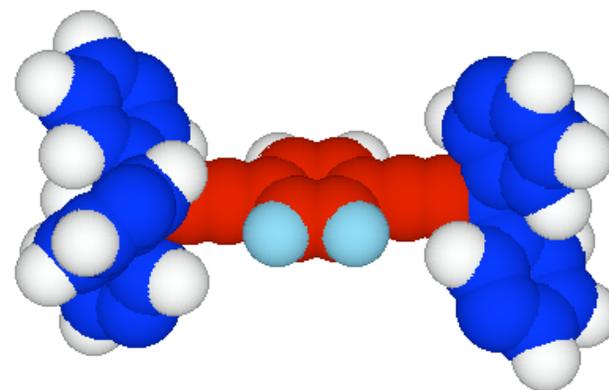
^2H NMR

Calculated

Experimental



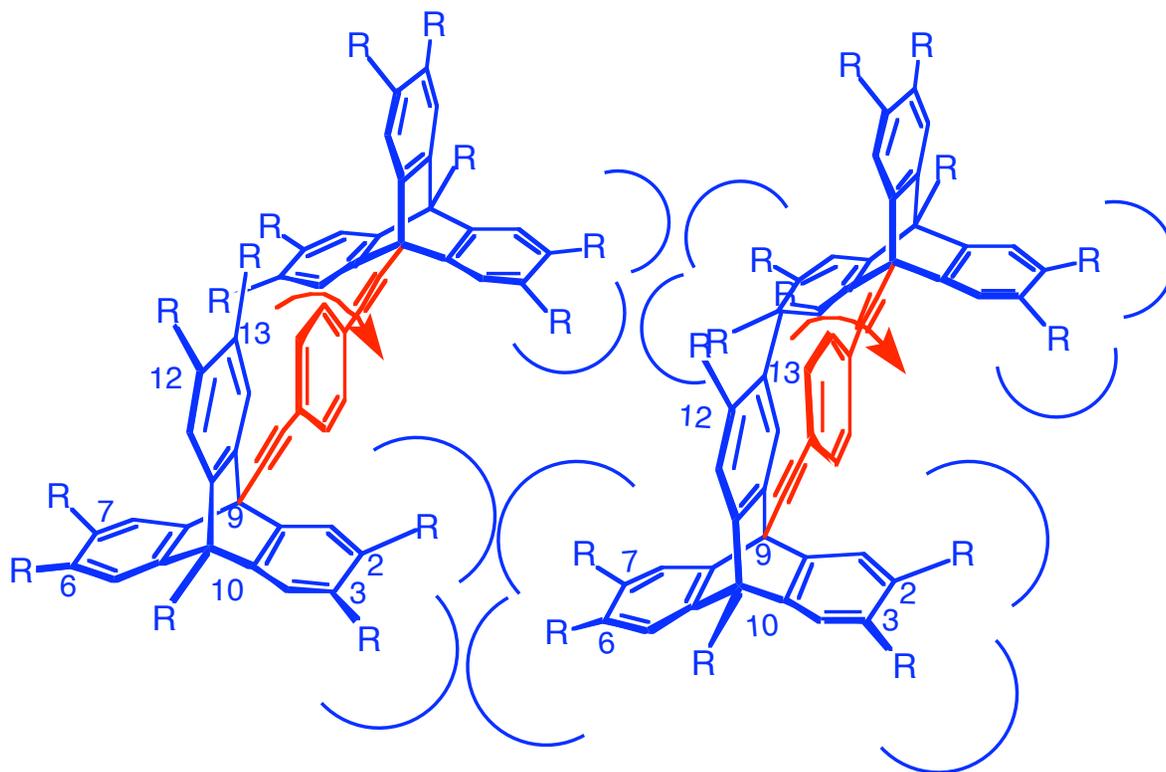
**Two-fold flip
(X-ray structure)**



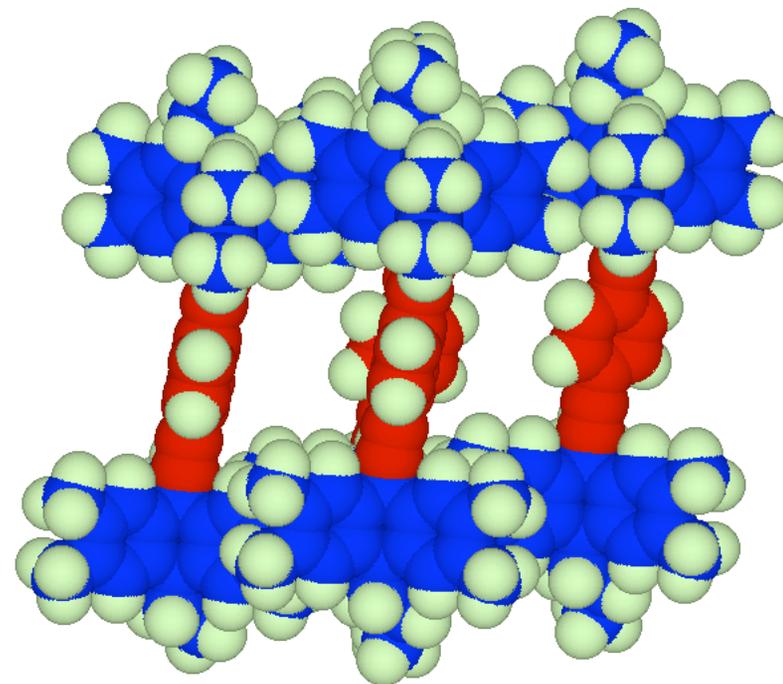
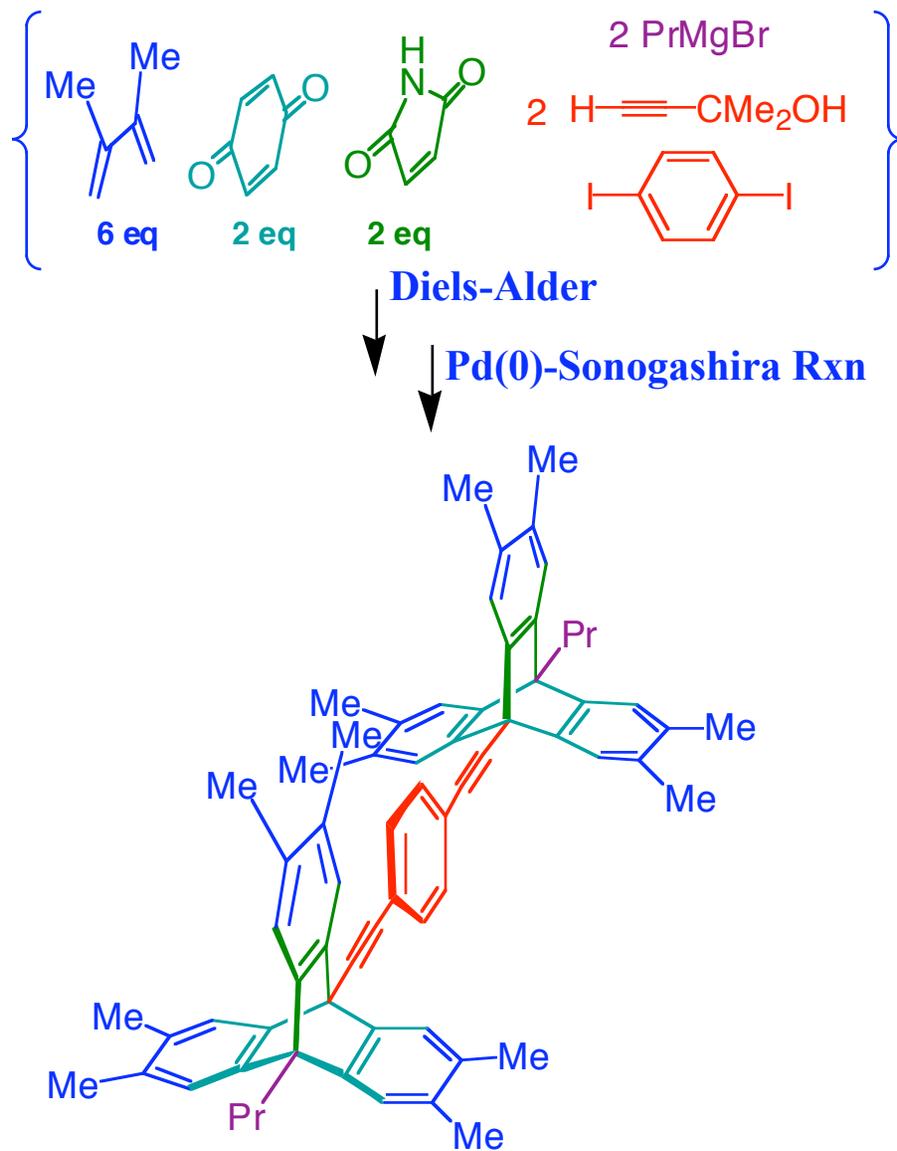
*1.3 MHz Rotation by
75 °C !*

**Two-fold flip
 $E_a = 14.6 \pm 1.5$ kcal/mol**

A Robust Frame for Robust Crystals



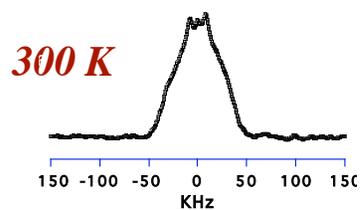
Robust and Shielding Triptycyl Frames



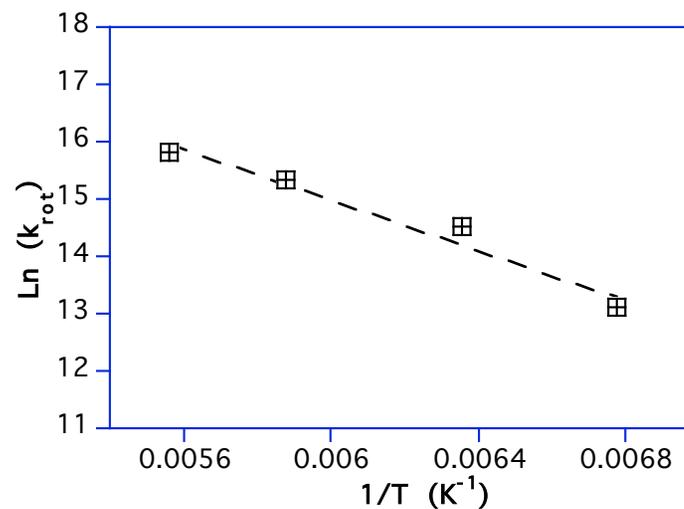
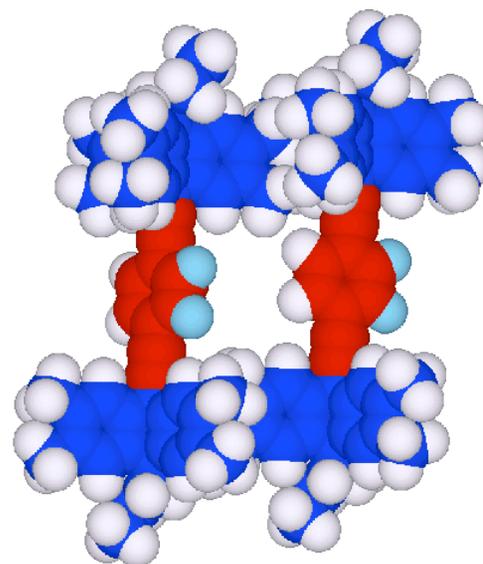
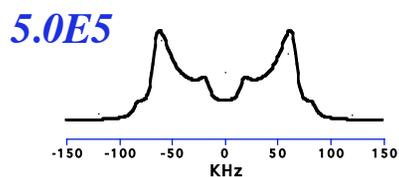
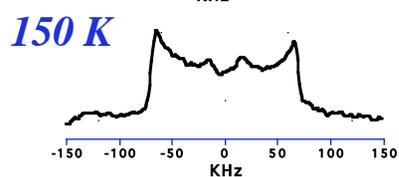
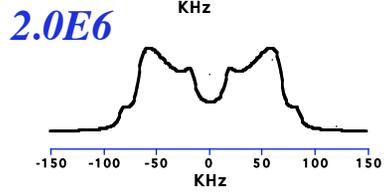
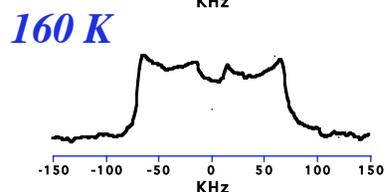
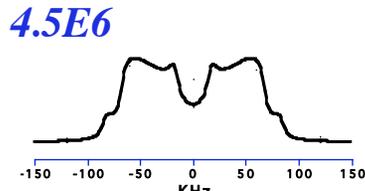
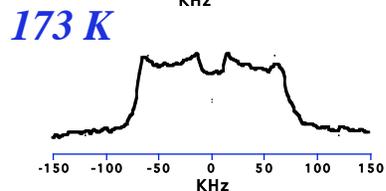
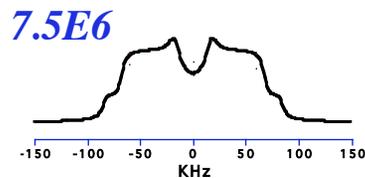
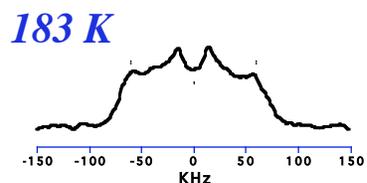
Cl-Benzene Removed

Godinez et al. *J. Org. Chem.* 2004, 69, 1652.

^2H NMR Line Shape Analysis



Too narrow to fit a 2-fold flip, but no continuous rotation

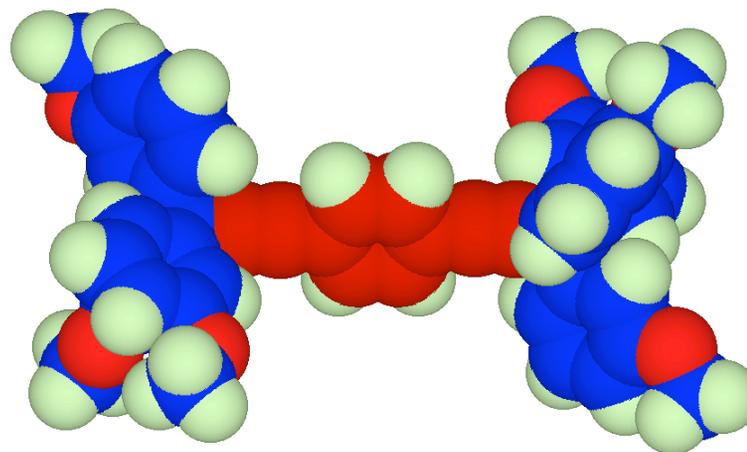
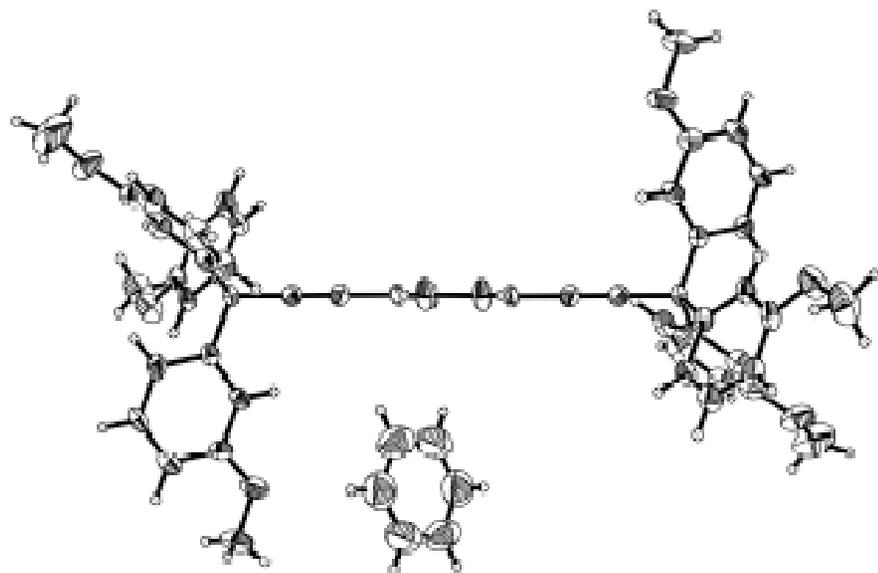
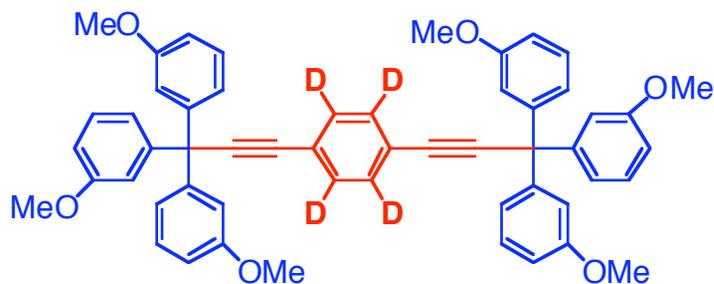


$A = 1.45 \times 10^{12}$; $E_a = 4.3 \text{ kcal/mol}$

Rotational Dynamics by X-Ray Diffraction

Dunitz, J.; Maverick, E. F.; Trueblood, K. N.

Angew. Chem. Int. Ed. 1988, 27, 880-895.



ORTEP:
Thermal Ellipsoids (ADP)

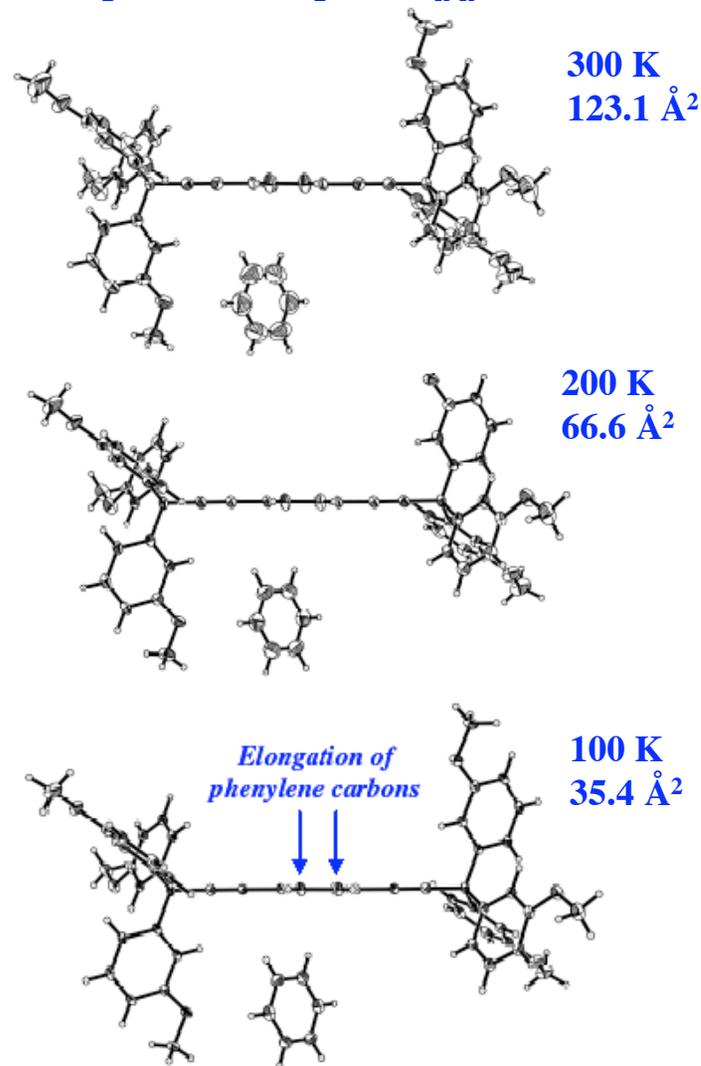
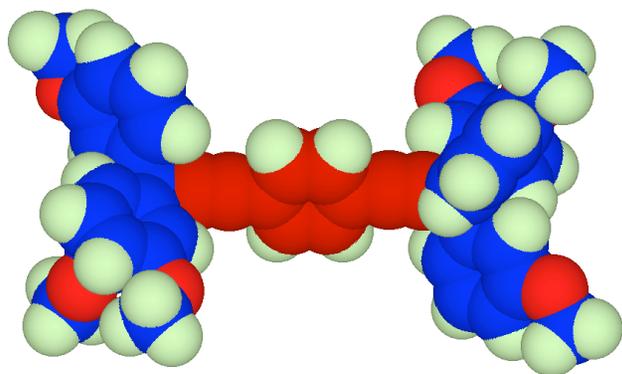
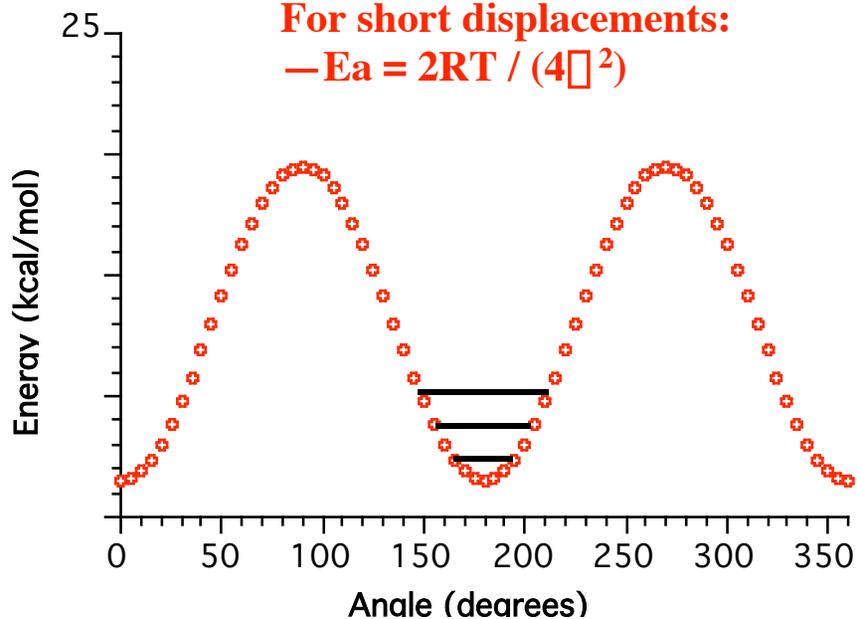
Rotational Dynamics by X-Ray Diffraction

Thermal Ellipsoids (ADP)

$$-V(x) = E_a (1 - \cos 2\theta) / 2$$

For short displacements:

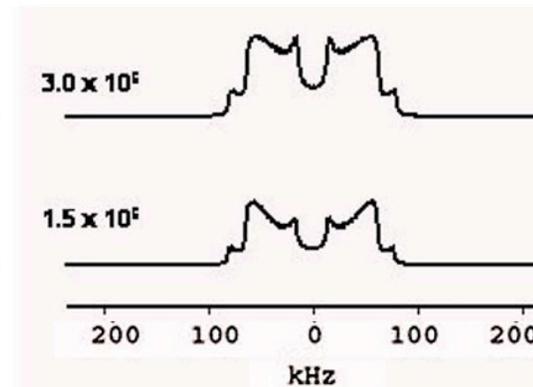
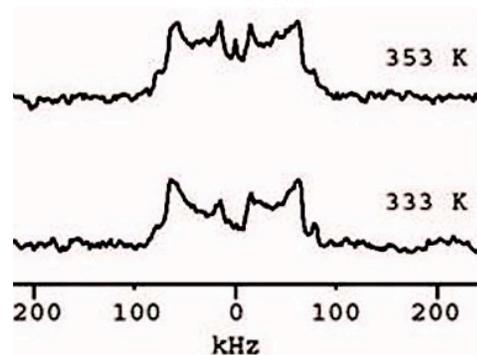
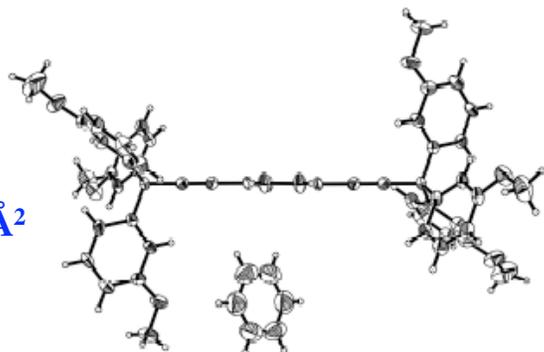
$$-E_a = 2RT / (4\theta^2)$$



$$E_a = 9.8 \text{ kcal/mol}$$

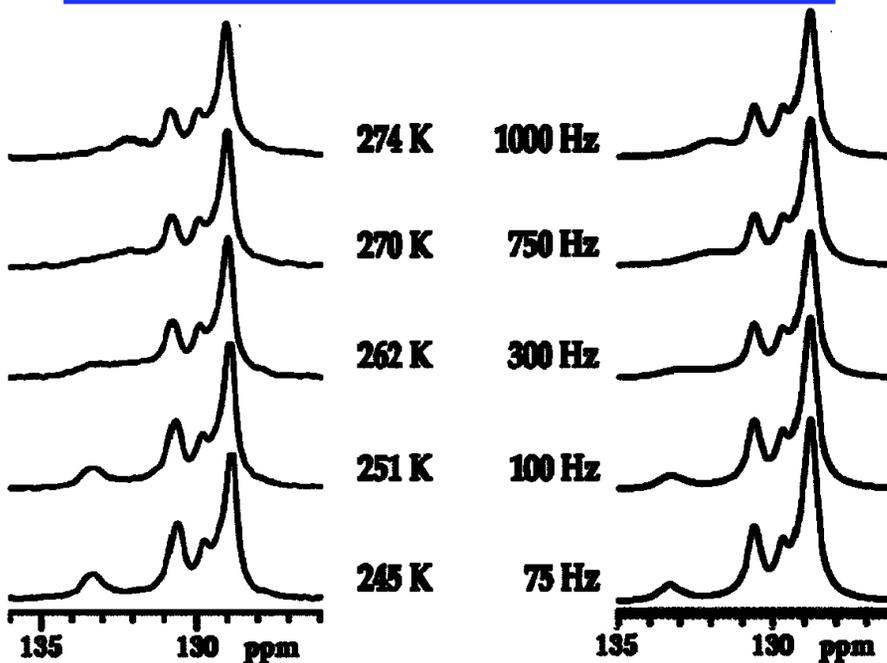
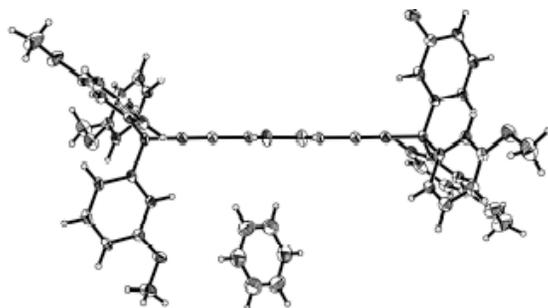
X-Ray Diffraction vs VT-NMR

300 K
123.1 Å²

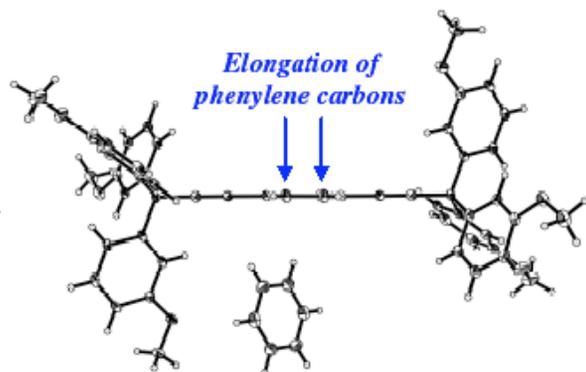


$E_a = 11.5 \text{ kcal/mol}; A = 3 \times 10^{13} \text{ s}^{-1}$

200 K
66.6 Å²



100 K
35.4 Å²

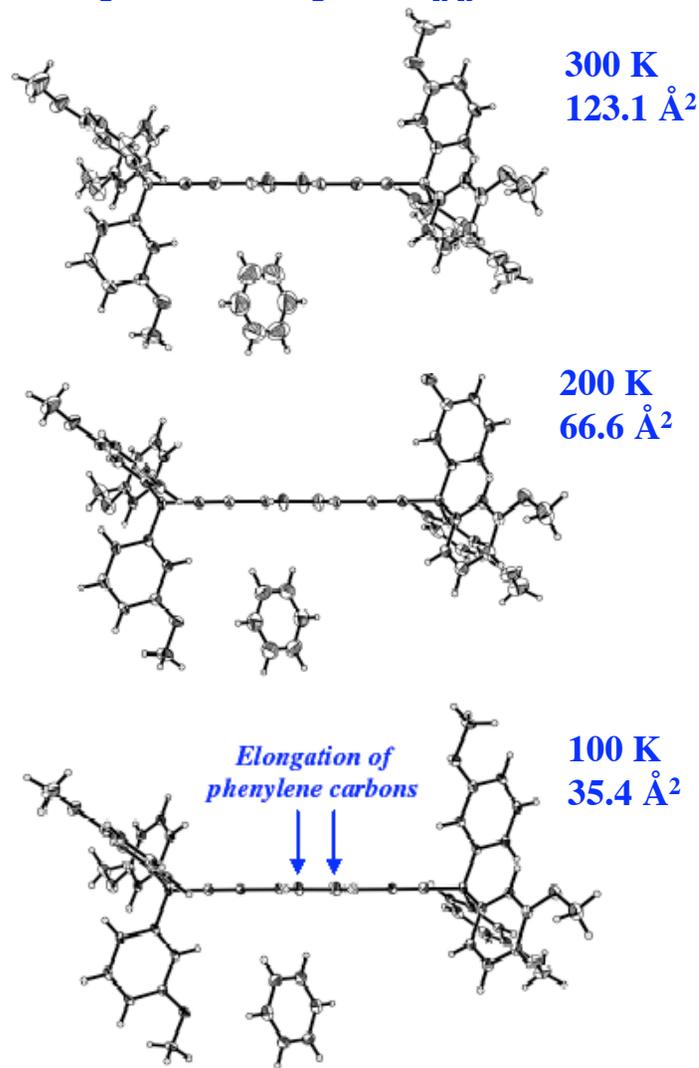
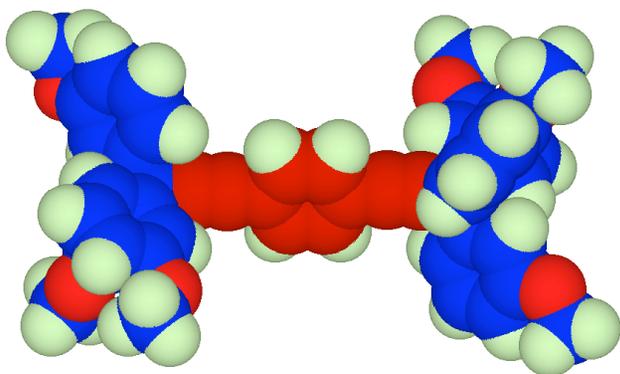
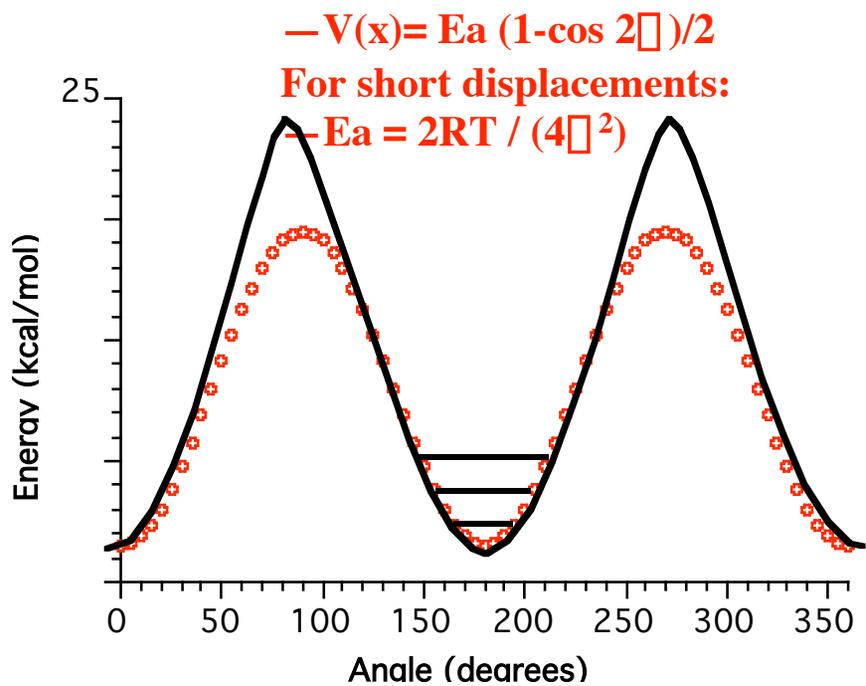


$E_a = 9.8 \text{ kcal/mol}$

$E_a = 11.7 \text{ kcal/mol}; A = 1.7 \times 10^{12} \text{ s}^{-1}$

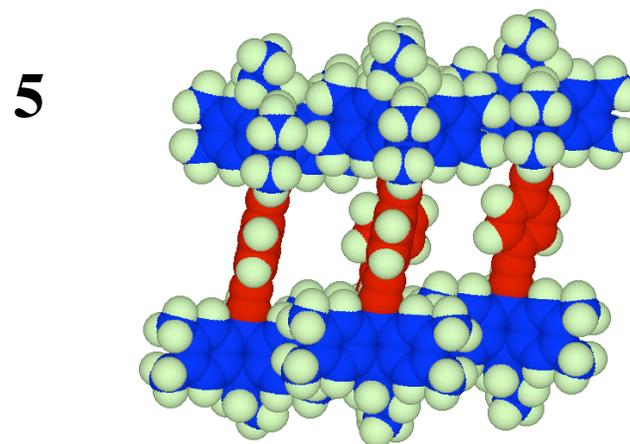
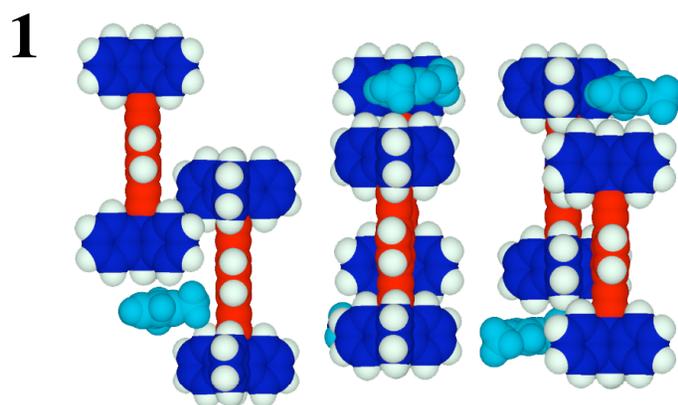
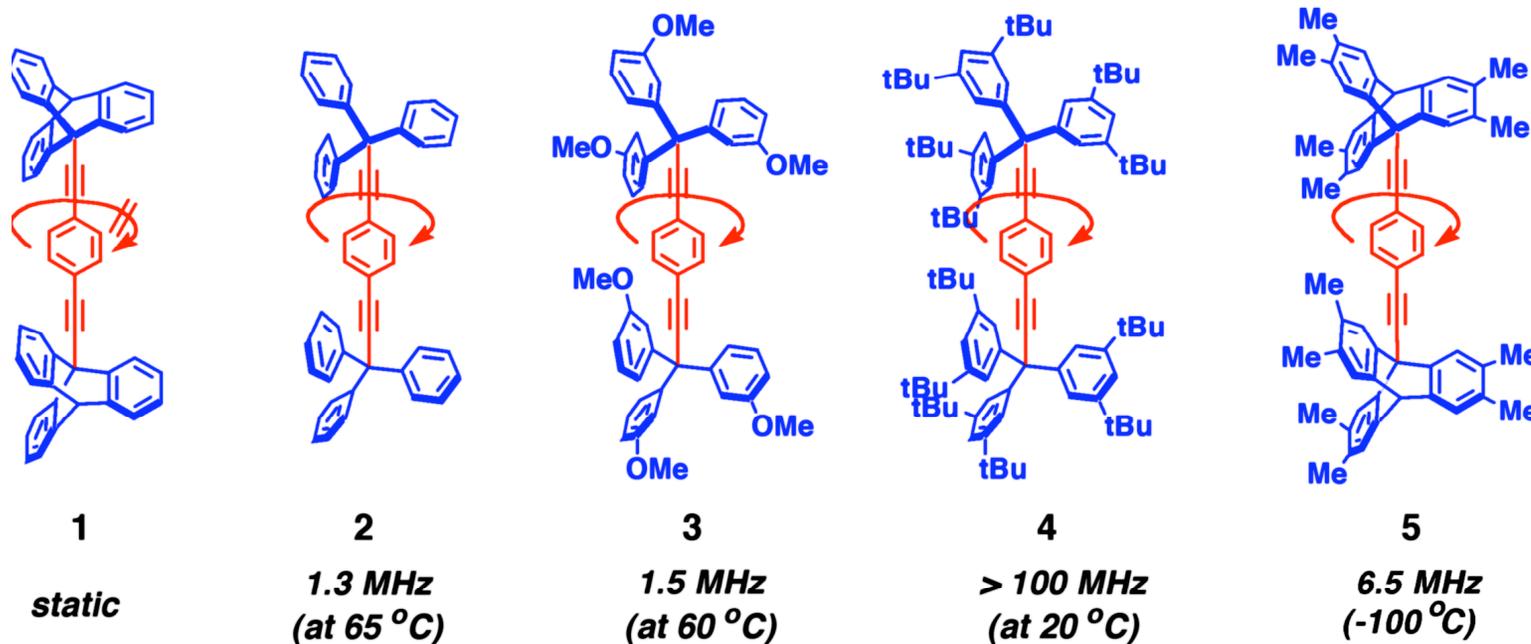
Rotational Dynamics by X-Ray Diffraction

Thermal Ellipsoids (ADP)



$E_a = 9.8 \text{ kcal/mol}$

Stator Effects on Rotary Motion (ca. $0-10^{10} \text{ s}^{-1}$)

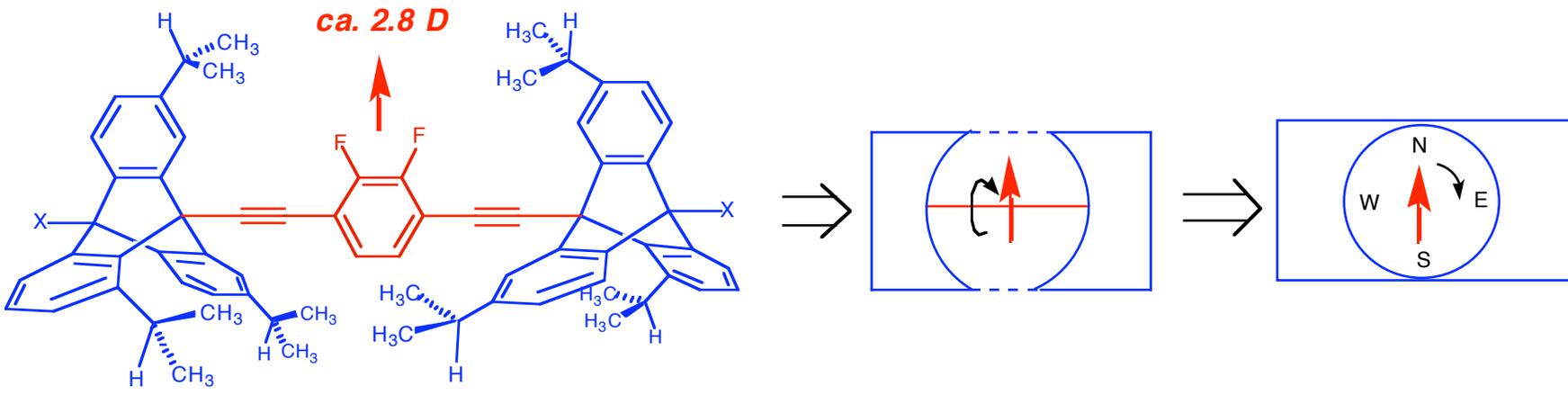


Godinez, et al., *JACS*. 2002, 124, 4701-4707.

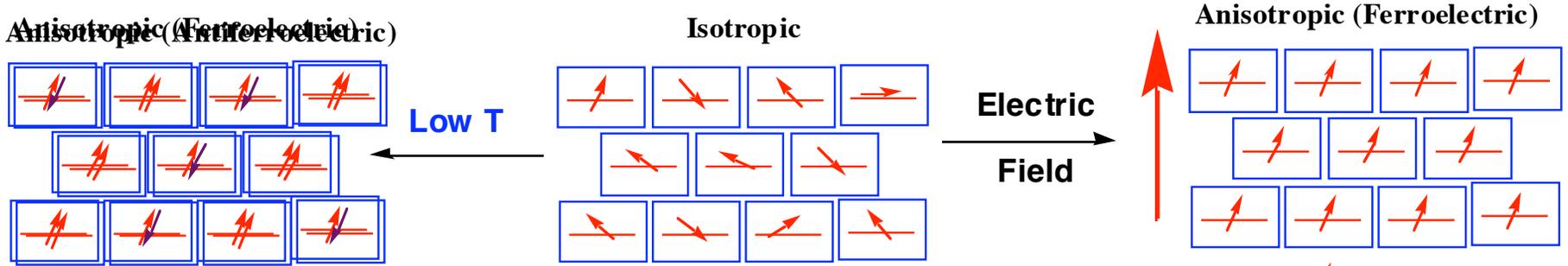
Godinez, et al., *JOC*. 2004, 69, 1652.

Feature article: Garcia-Garibay, M. *Proc. Natl. Acad. Sci, USA*, 2005, 102, 10771-10776.

Molecular Compasses



Photonic Materials

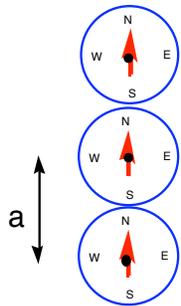


- Dichroism
- Birefringence
- NLO

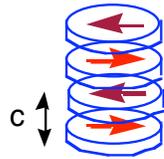
- Electric Dipoles
- Frictionless => Switches
- Hindered => Polar Materials, Electrets

Reorienting Dipole Lattices

1D-Chain

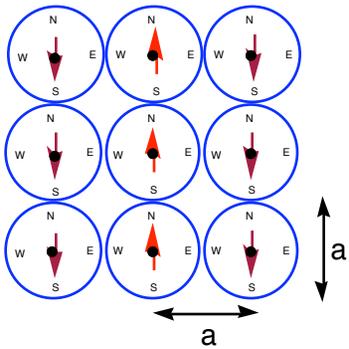


1D-Stack



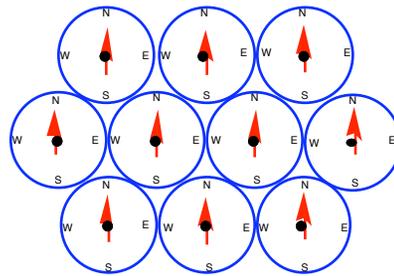
Square

$a = b, \alpha = 90^\circ$

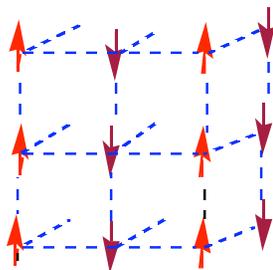


Hexagonal

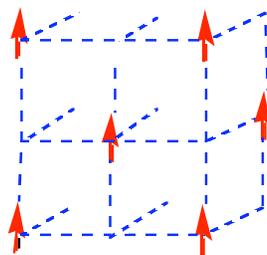
$a = b, \alpha = 60^\circ$



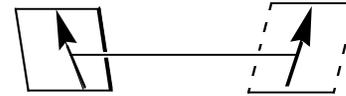
3D-Cubic



3D-FCC



$$E_{dd} = \frac{1}{4\pi\epsilon_0} \frac{q_i q_j}{r_{ij}^3} [\cos\alpha - 3\cos\alpha_1 \cos\alpha_2]$$



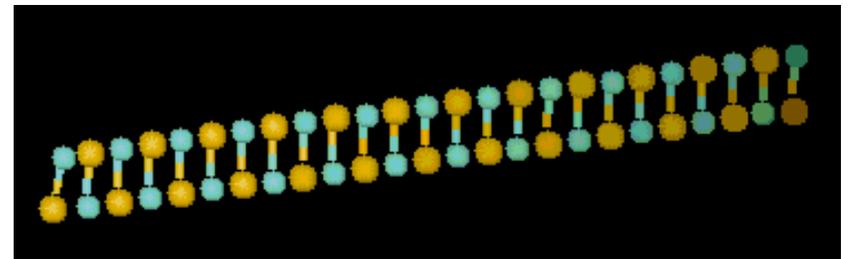
ϵ = dielectric constant

α = dihedral angle between dipoles

α_1, α_2 = angle between dipole vector and line

r_{ij} = distance between dipoles

M. Ratner



$$f_0 = \frac{1}{2\epsilon} \sqrt{\frac{k_B T_0}{I}} \quad \text{interaction energy} \quad \approx 100 \text{ GHz}$$

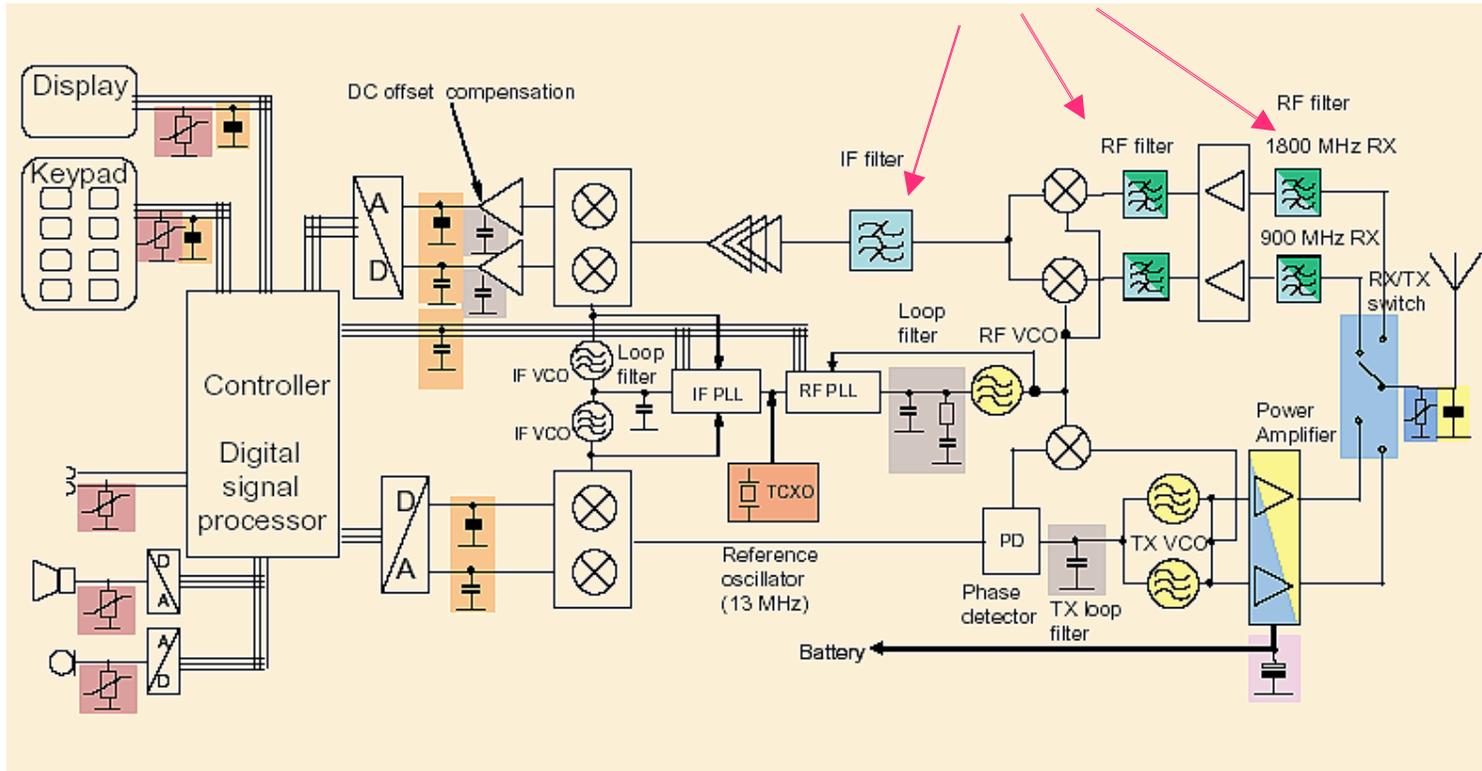
moment of inertia

$$v_R \approx f_0 a \approx 100 \text{ GHz} \cdot 10 \text{ \AA} = 100 \text{ m/s}$$

lattice constant

Surface Acoustic Wave Filters and Delay Lines

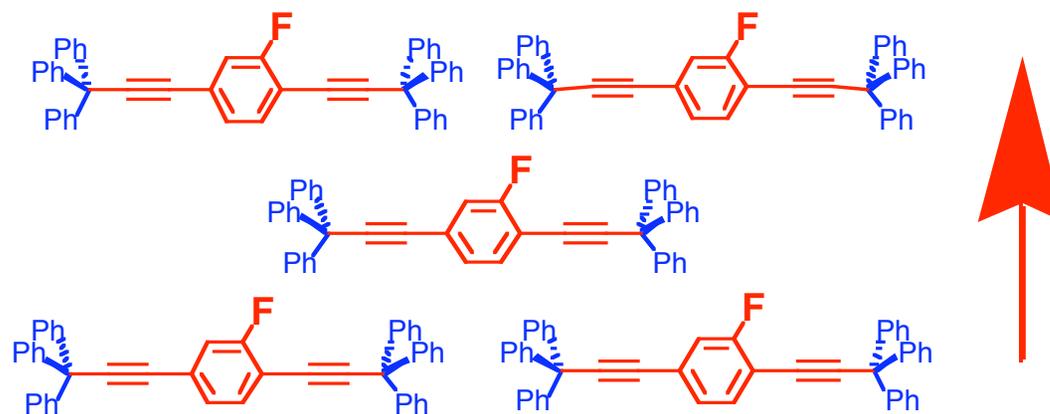
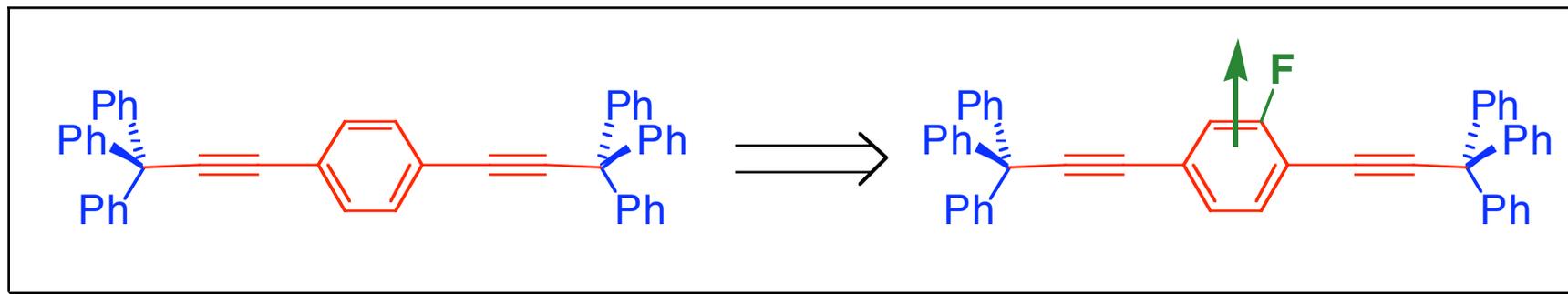
SAW filters



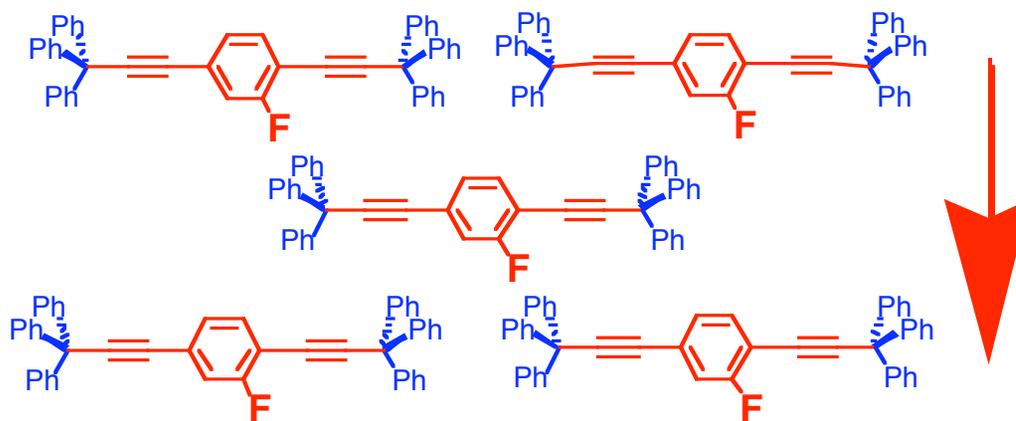
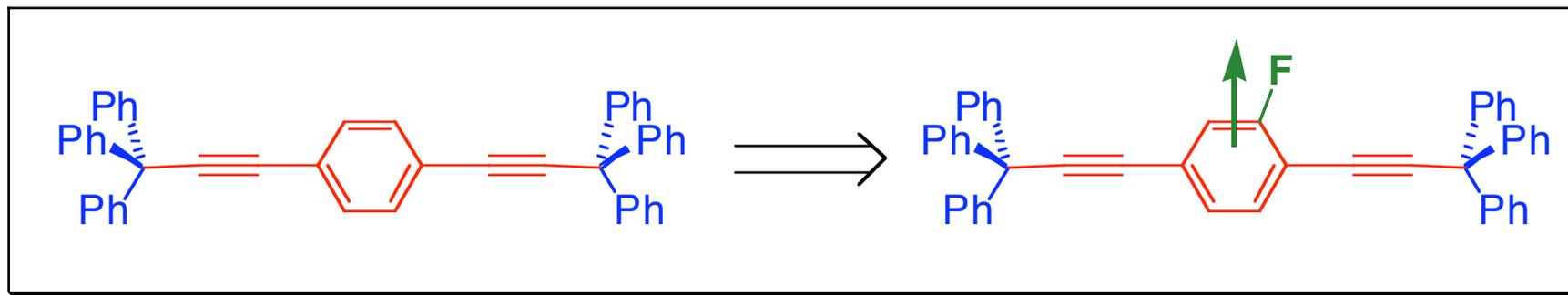
Schematic of a GSM cell phone handset (EPCOS)

More than 1 billion SAW devices are sold annually

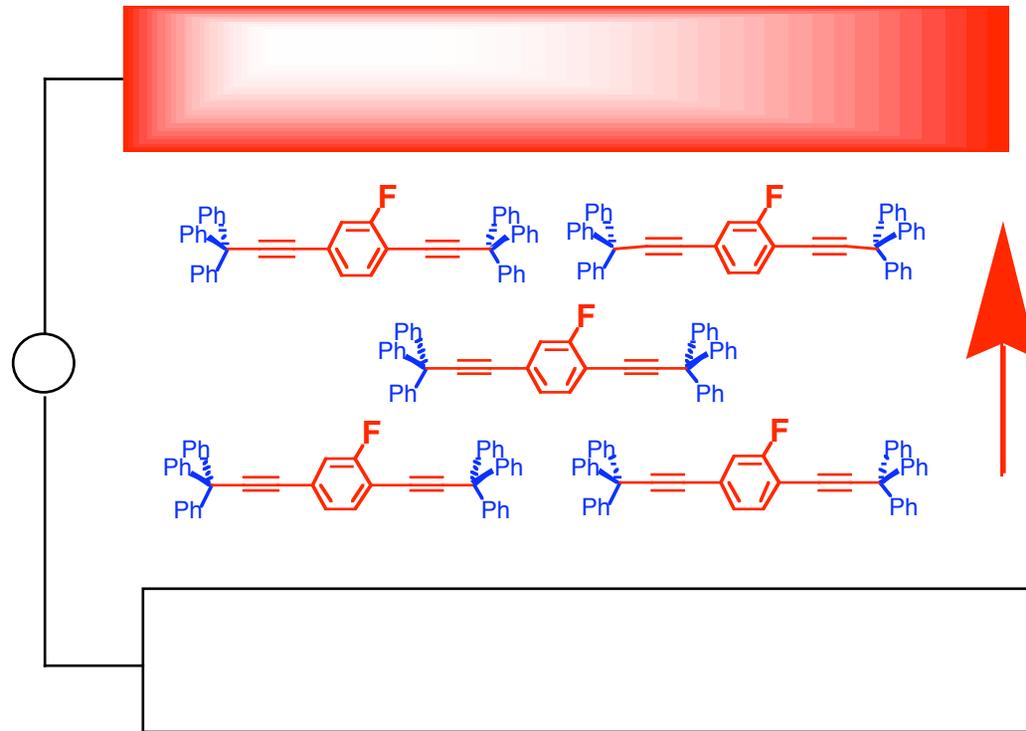
Rotational Dynamics: Two-fold flip



Rotational Dynamics: Two-fold flip

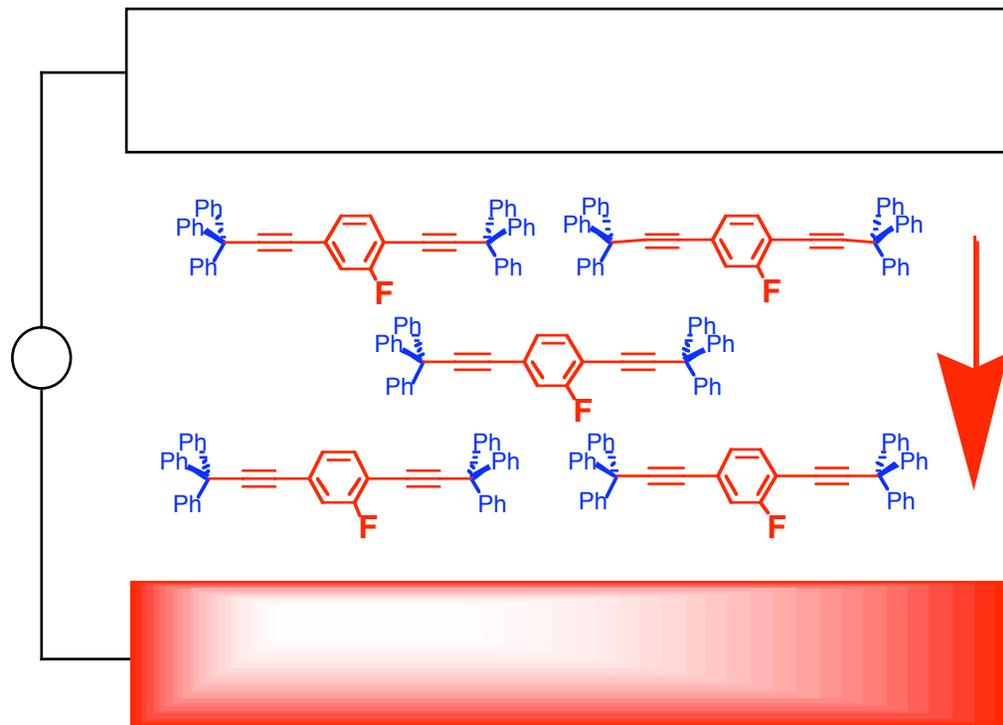


Dielectric Measurements



(Electric Energy is Absorbed when AC Frequency Matches that of the Internal Dipole Dynamics)

Dielectric Measurements

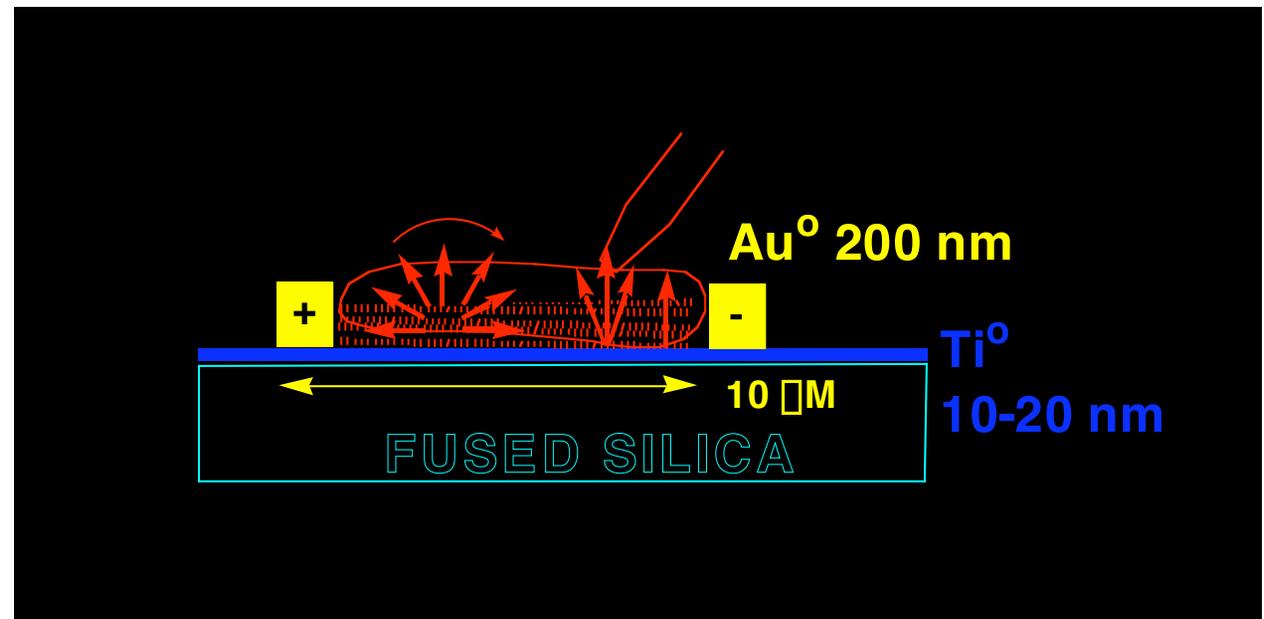


(Electric Energy is Absorbed when AC Frequency Matches that of the Internal Dipole Dynamics)

Dielectric Measurements



*Interdigitated Electrodes
Patterned Photolithographically
by Wet Etching*



F-Rotor: Frequency-Dependent Dielectric Measurements

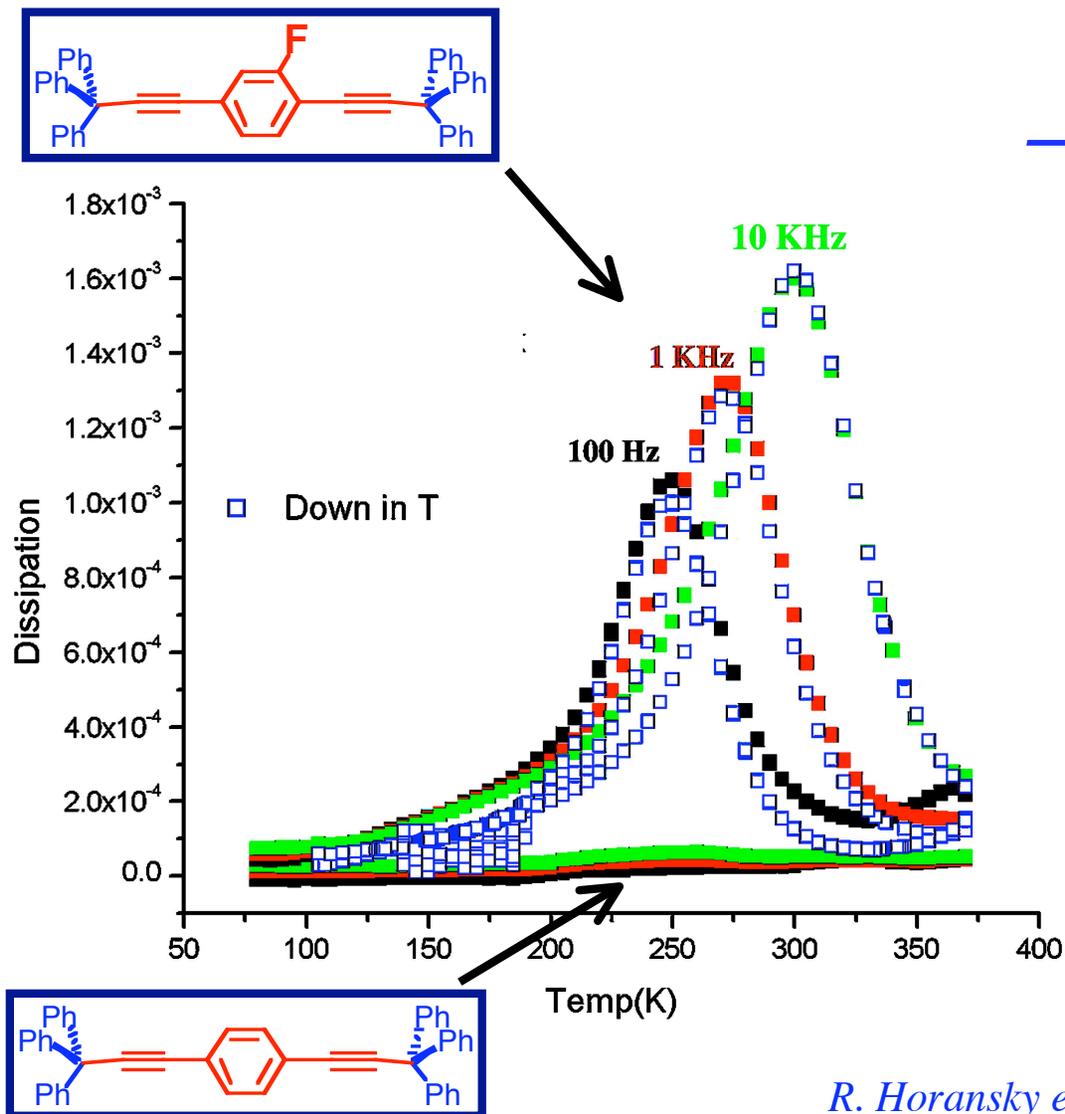
— Polycrystalline samples

a) Non-Polar Rotor Gives Baseline

b) Reversible on Heating and cooling cycles

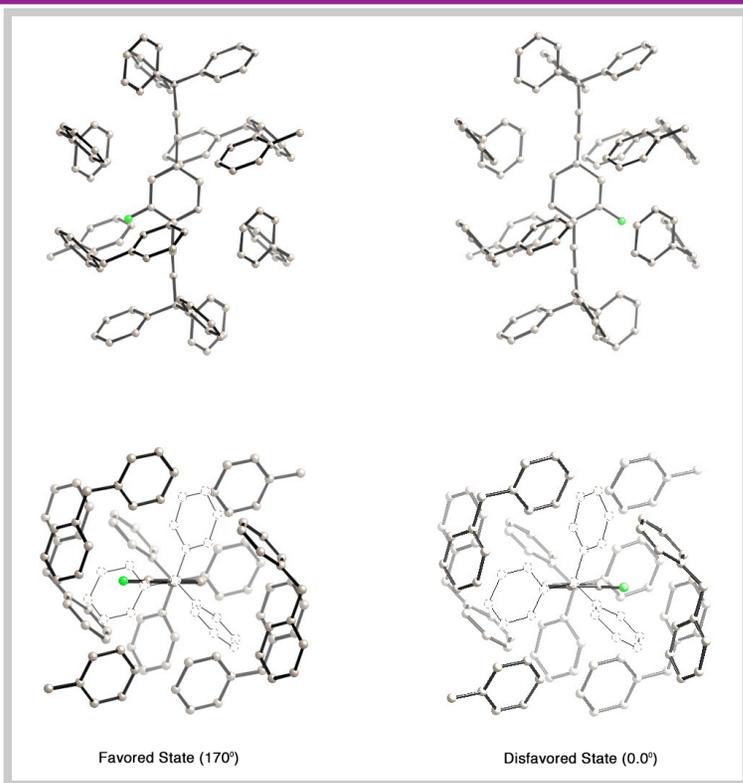
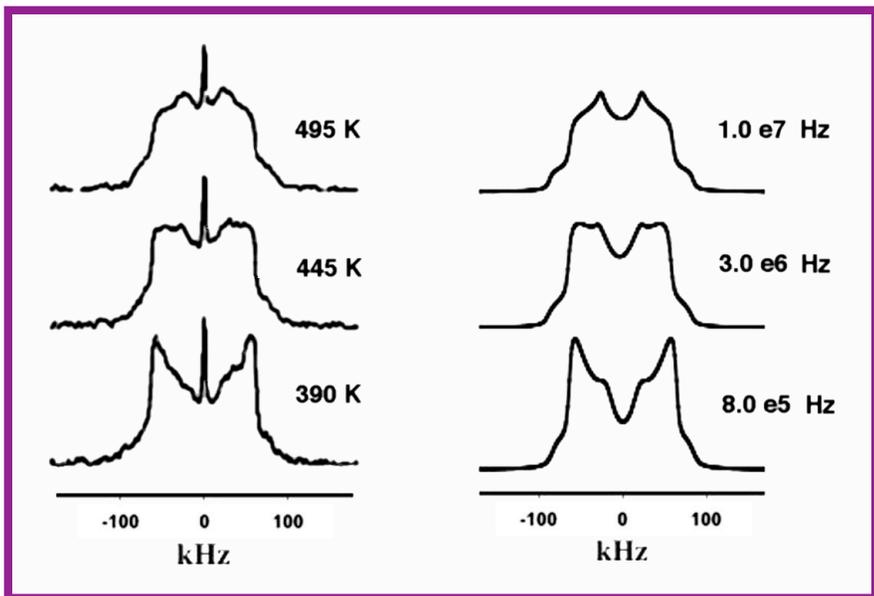
c) Peak Position Depends on AC Freq.

d) Signal Intensity Increases With Temp.

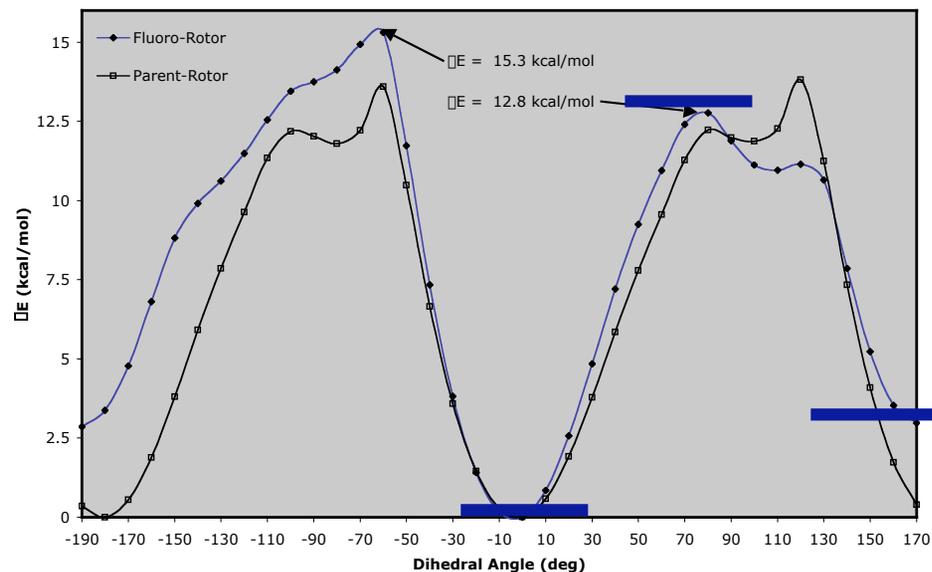


R. Horansky et al. Phys. Rev. B., 2005, 72_014302.

F-Rotor: Force-Field Estimate of Rotational Potential



Molecular Mechanics Dihedral Drives
Local Model



DIELECTRIC

$E_a = 13.7 \text{ kcal/mol}$

$\Delta E = 1.4 \text{ kcal/mol}$

^2H NMR

$E_a = 12.4 \text{ kcal/mol}$

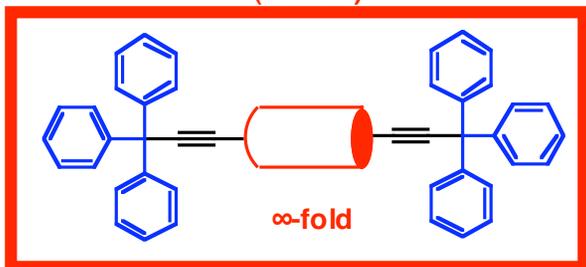
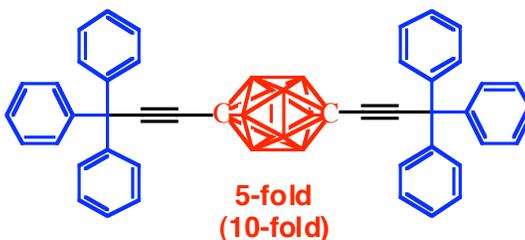
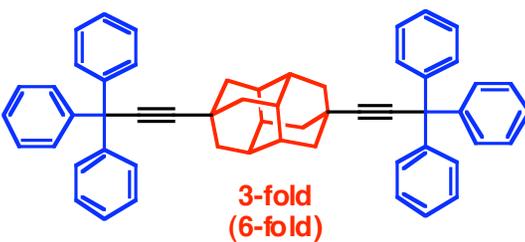
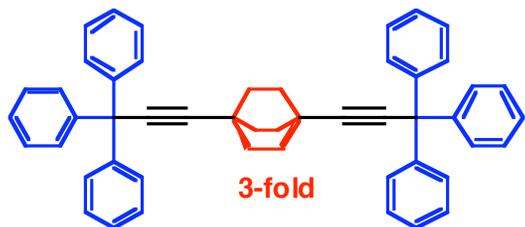
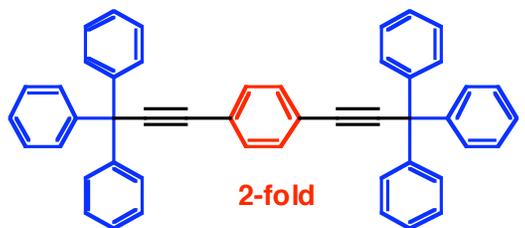
$\Delta E = 0.8 \text{ kcal/mol}$

Force-Field Potential

$E_a = 12.8 \text{ kcal/mol}$

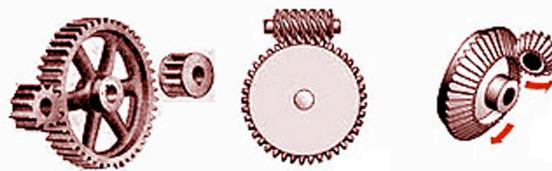
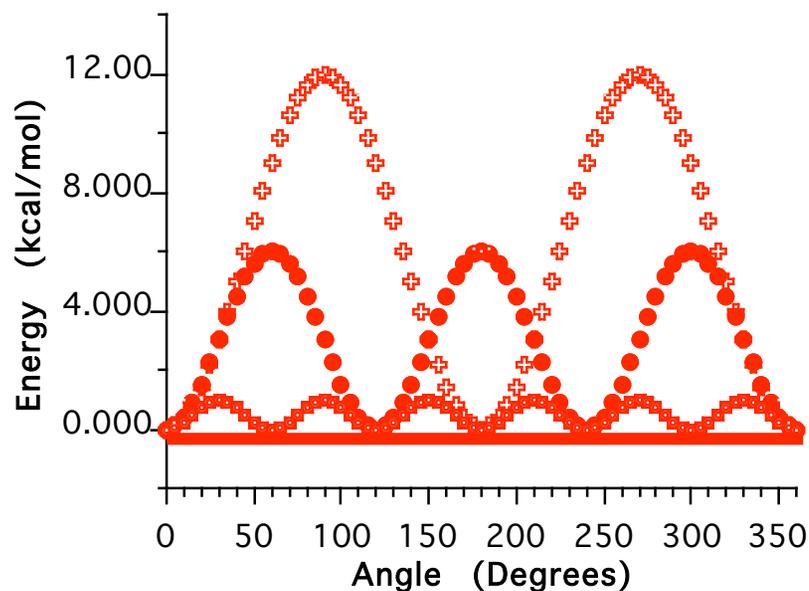
$\Delta E = 2.5 \text{ kcal/mol}$

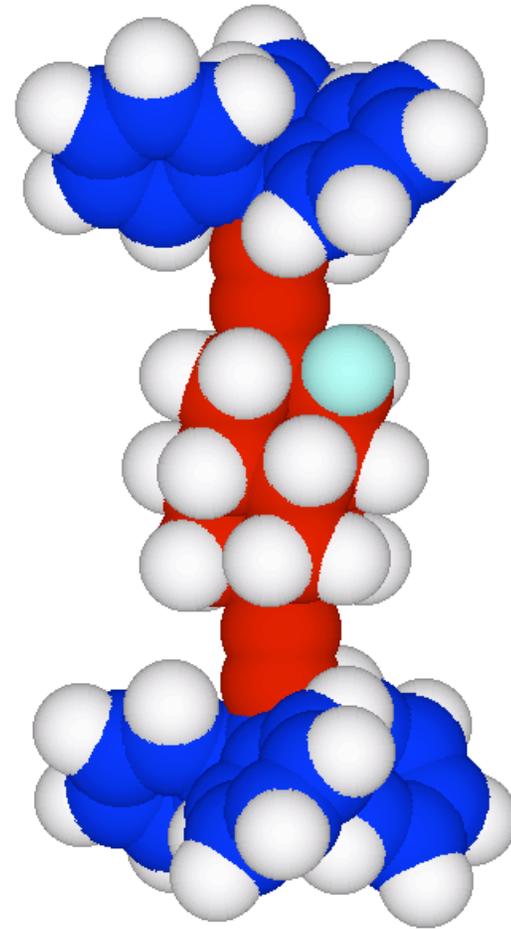
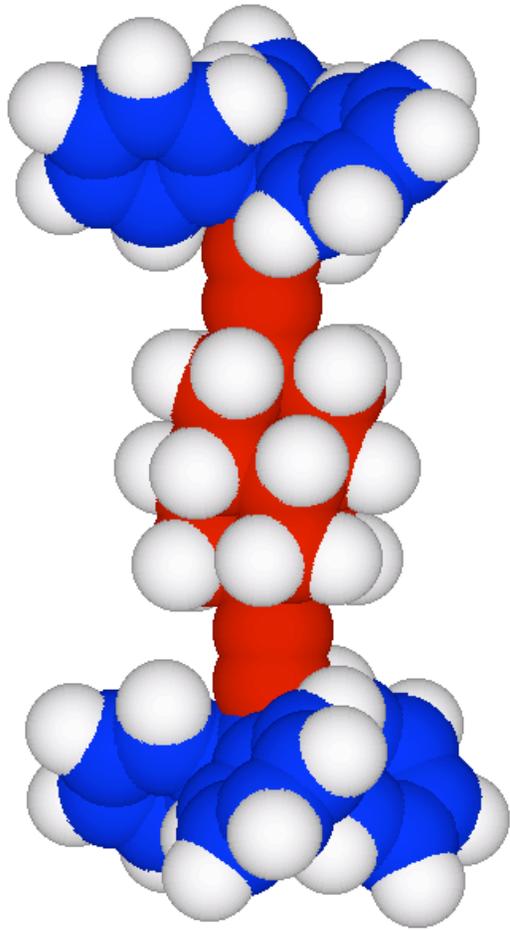
Rotator Effects on Rotary Motion ?



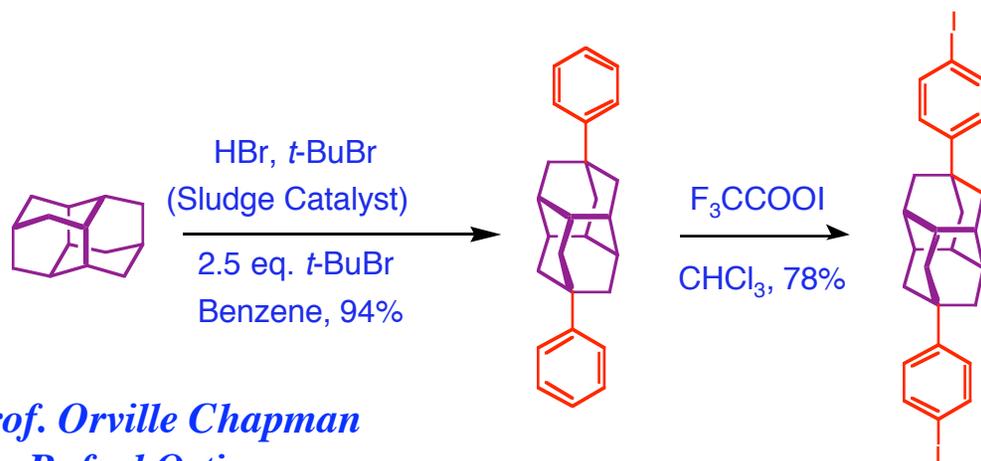
High Symmetry Order Rotors

- Low rotational barriers
- High radial resolution (polarity)
- Cogitation (gearing)



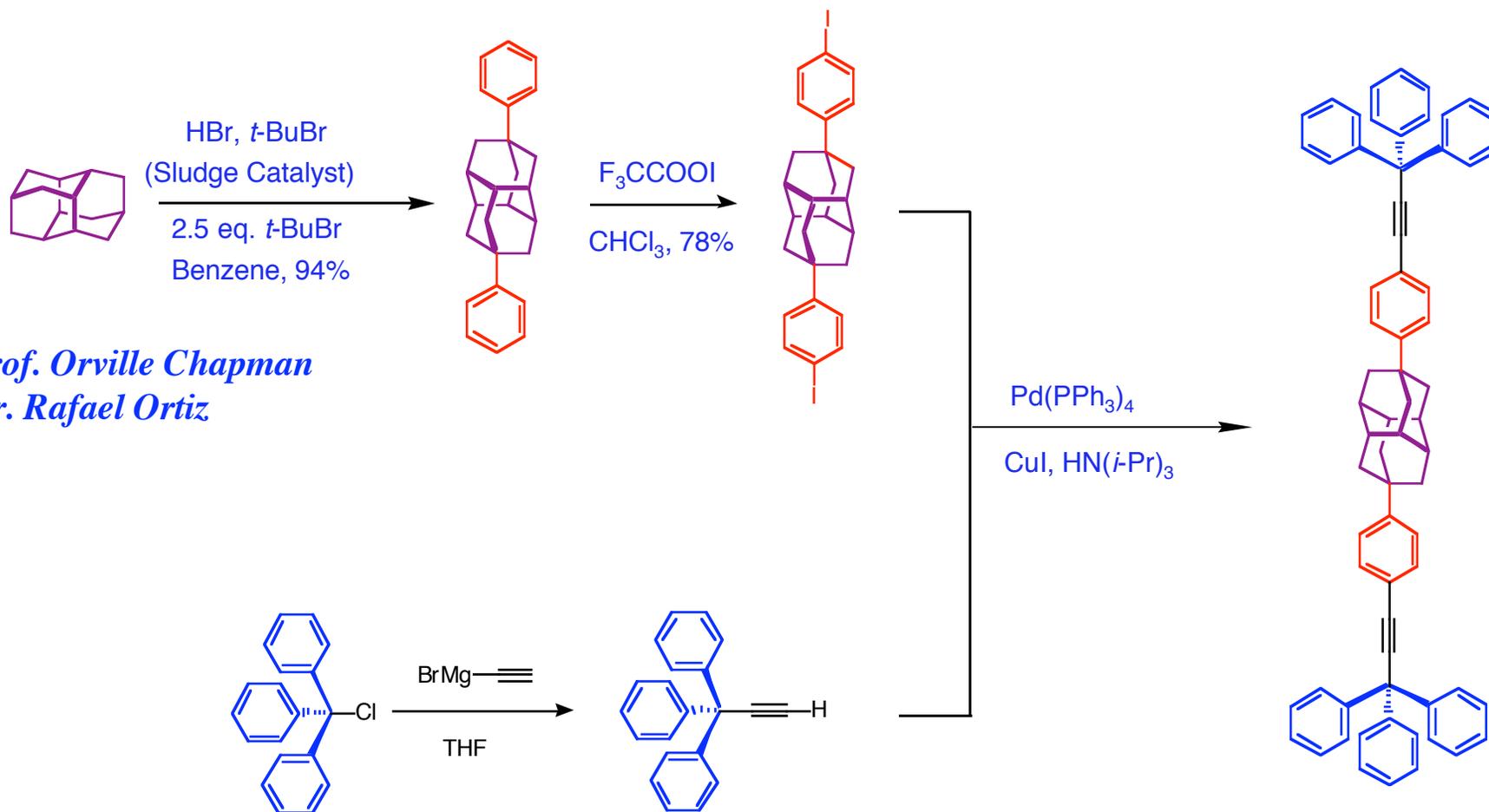


The right place, the right time....



Prof. Orville Chapman
Dr. Rafael Ortiz

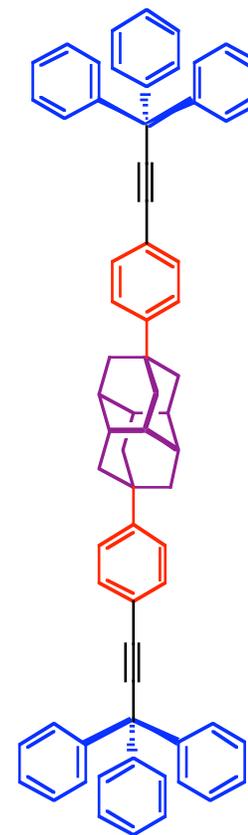
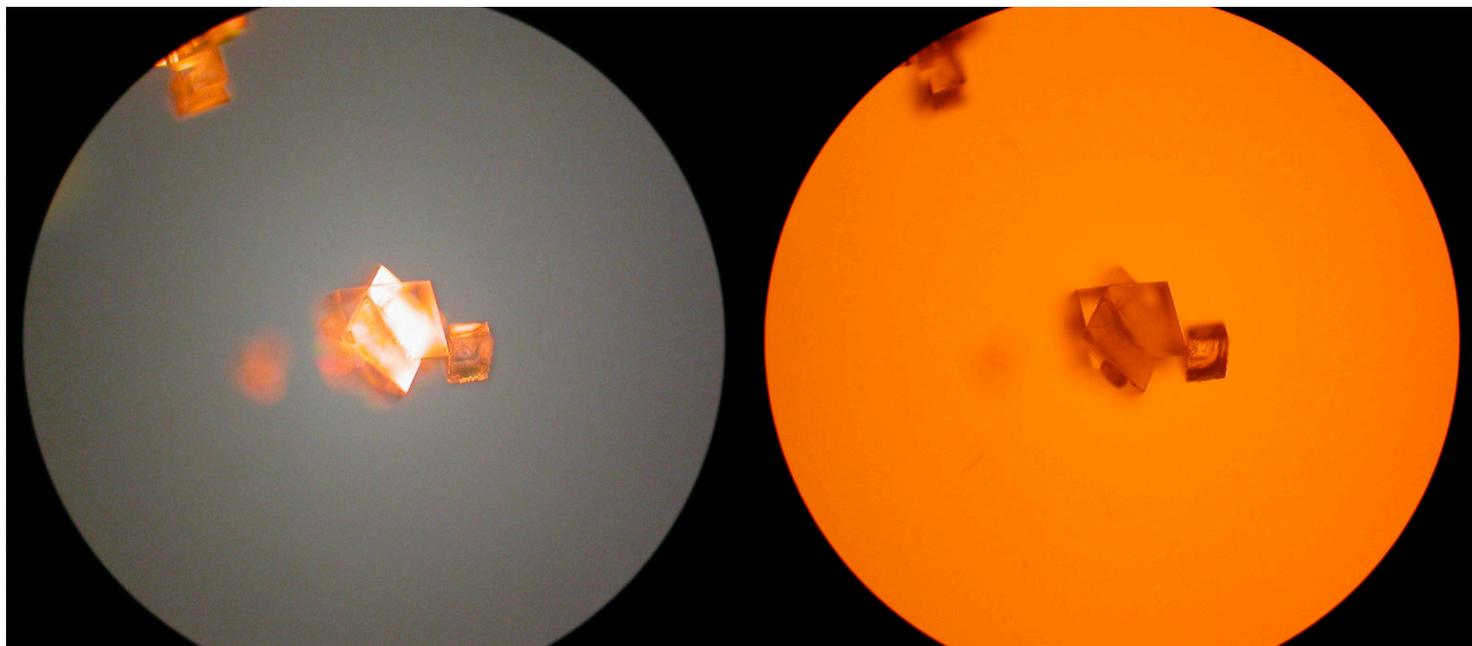
The right place, the right time....



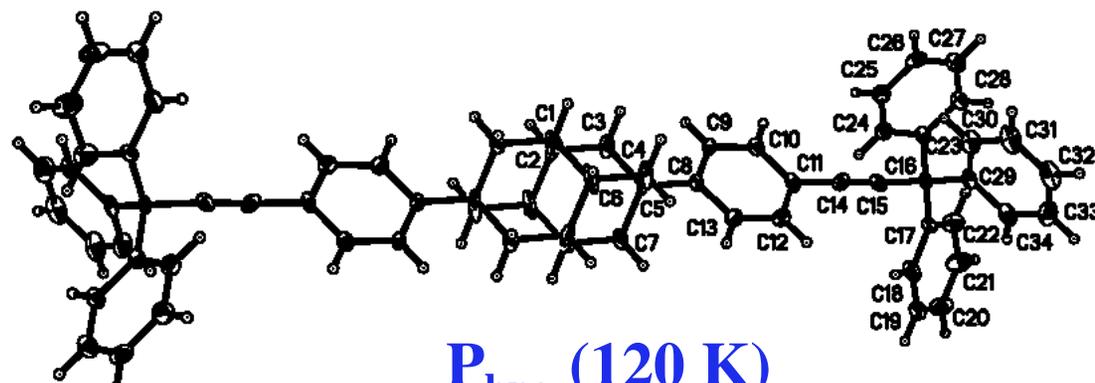
*Prof. Orville Chapman
Dr. Rafael Ortiz*

—Compare Rotatory Dynamics of Phenylene and Diamantene!

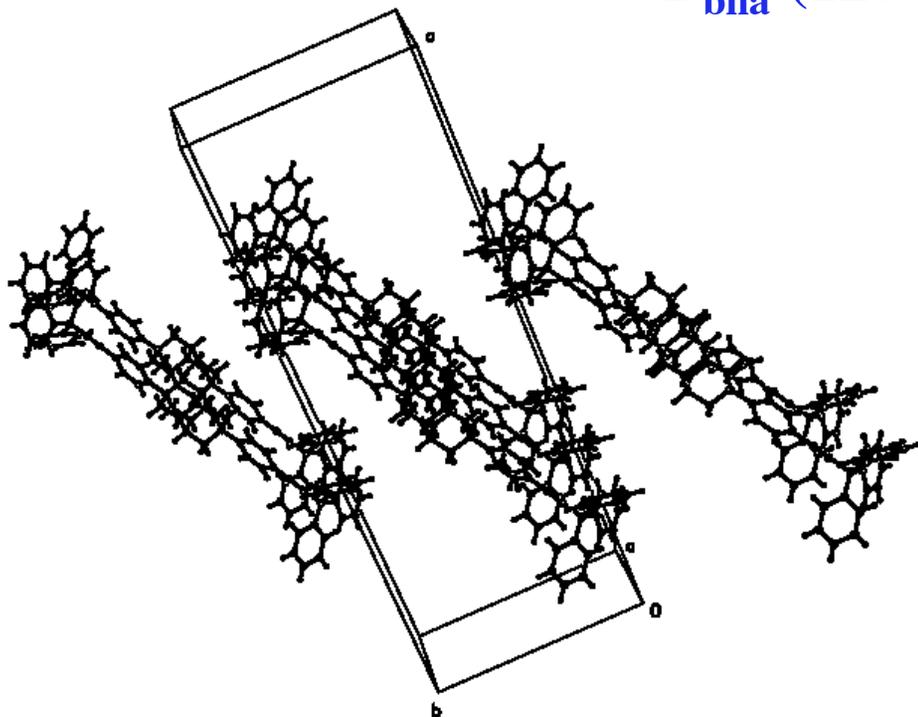
*Bis(phenylene)diamantane Rotator system:
X-Ray structure*



*Bis(phenylene)diamantane Rotator system:
X-Ray structure*

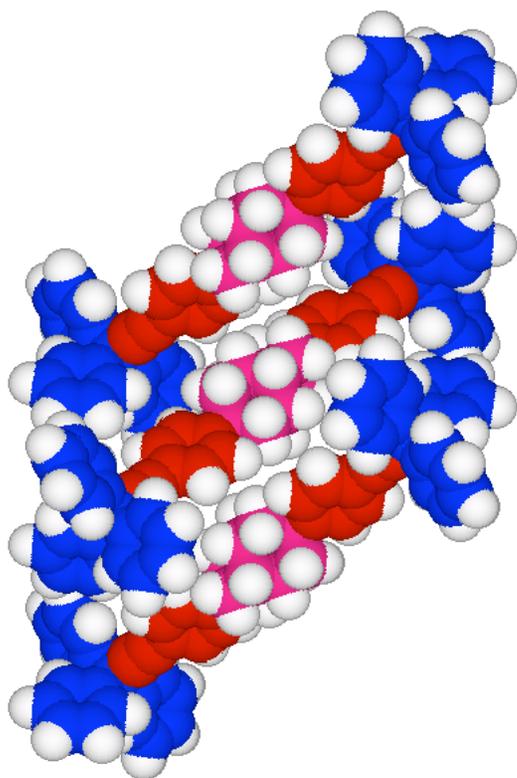
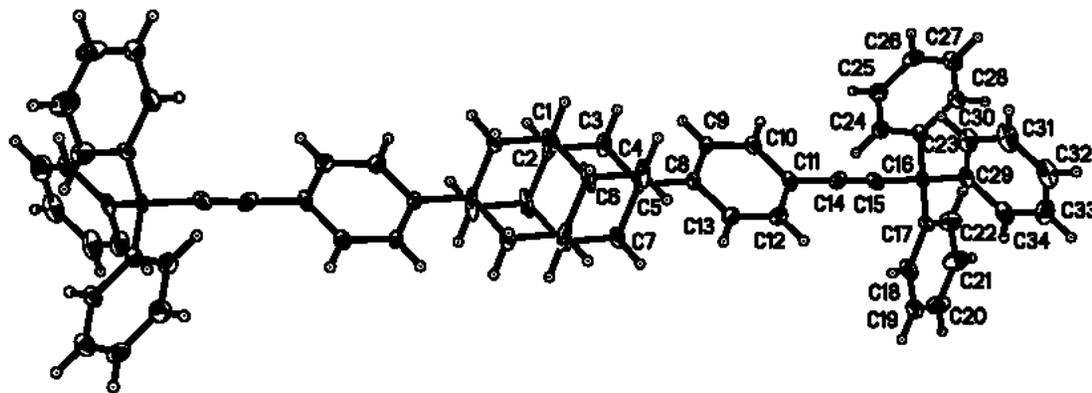


P_{bna} (120 K)



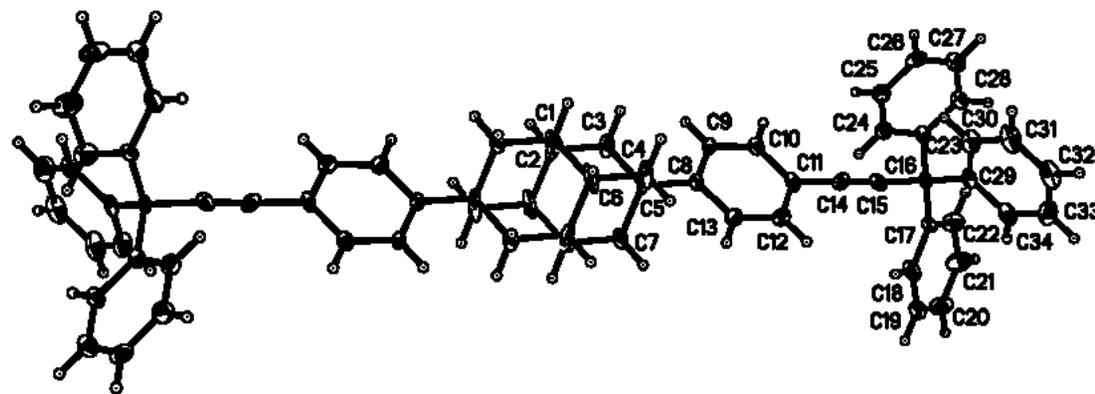
Log-stack packing

Diamantane Rotor

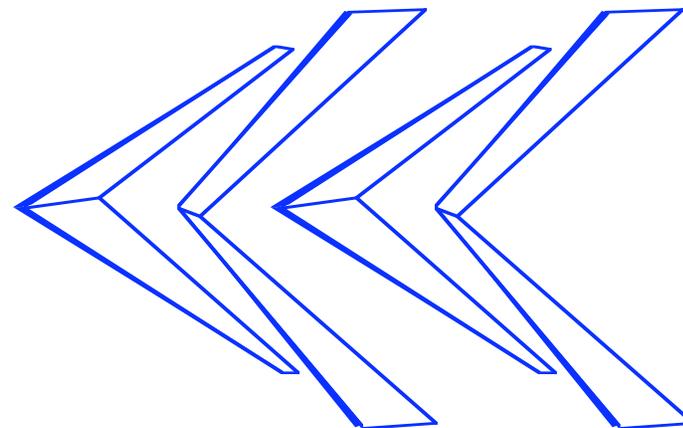
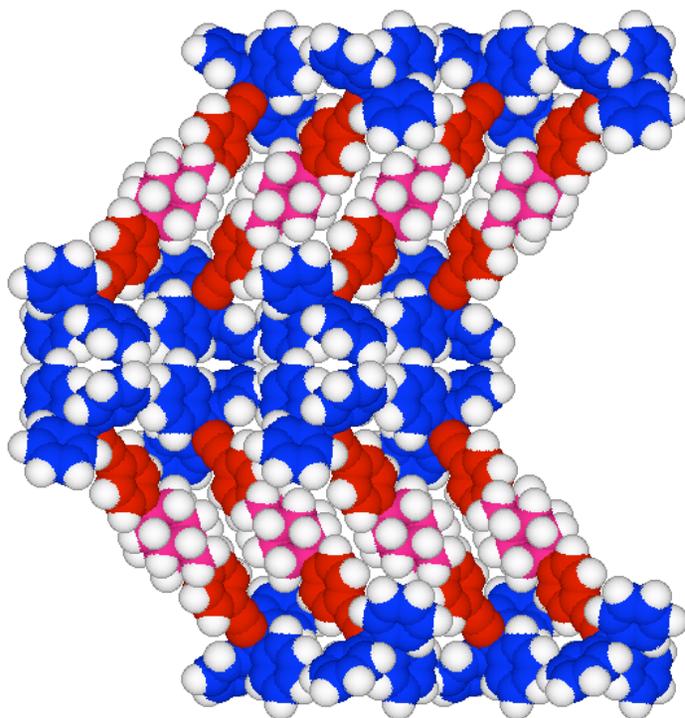


Log-stack packing

Diadamantane Rotor

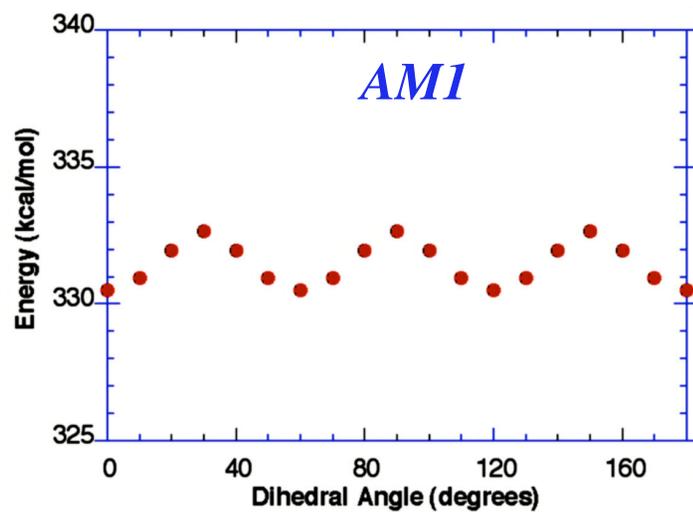
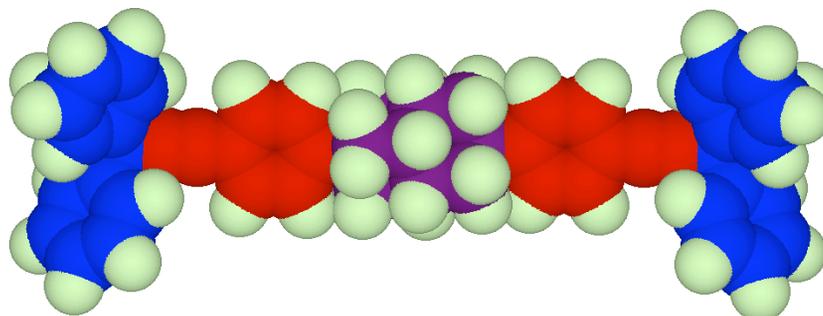
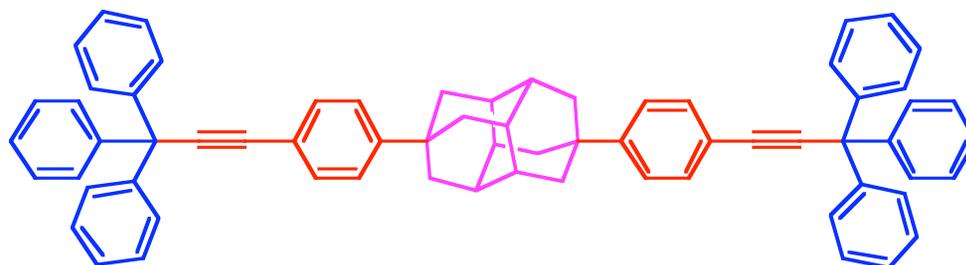


*four-fold
phenyl
embrace*

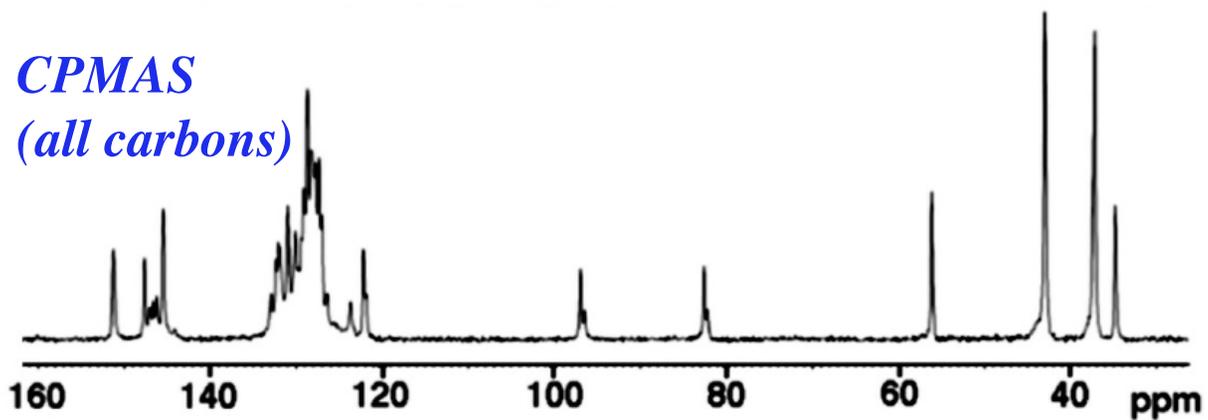
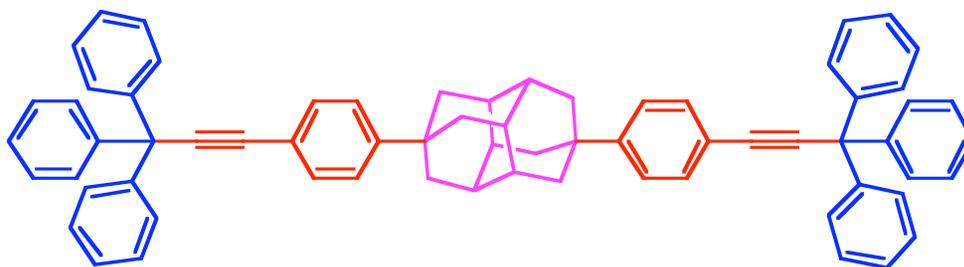


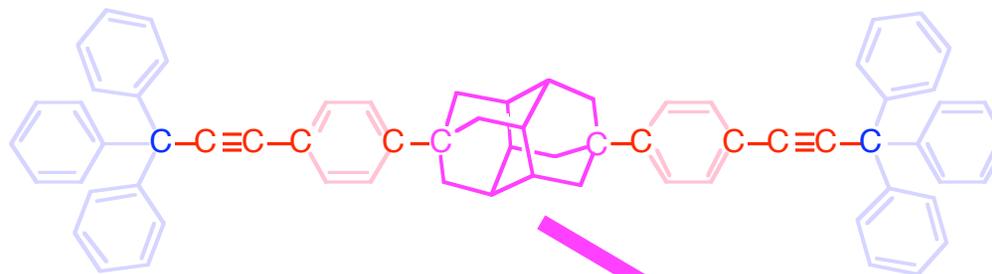
In-and-out Chevrons

*Bis(phenylene)diamantane Rotator system:
Intrinsic bis(phenylene)-diamantene rotational barrier*

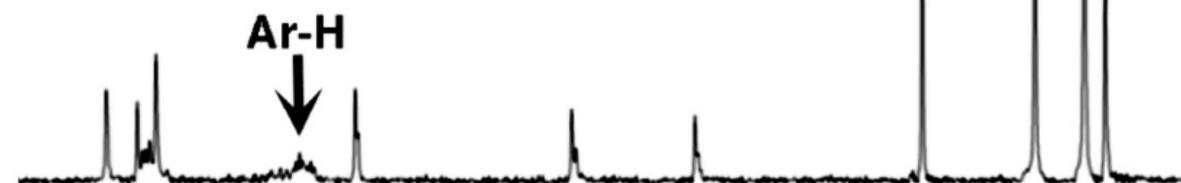


A Simple Insight into Rapid Diamantene Dynamics

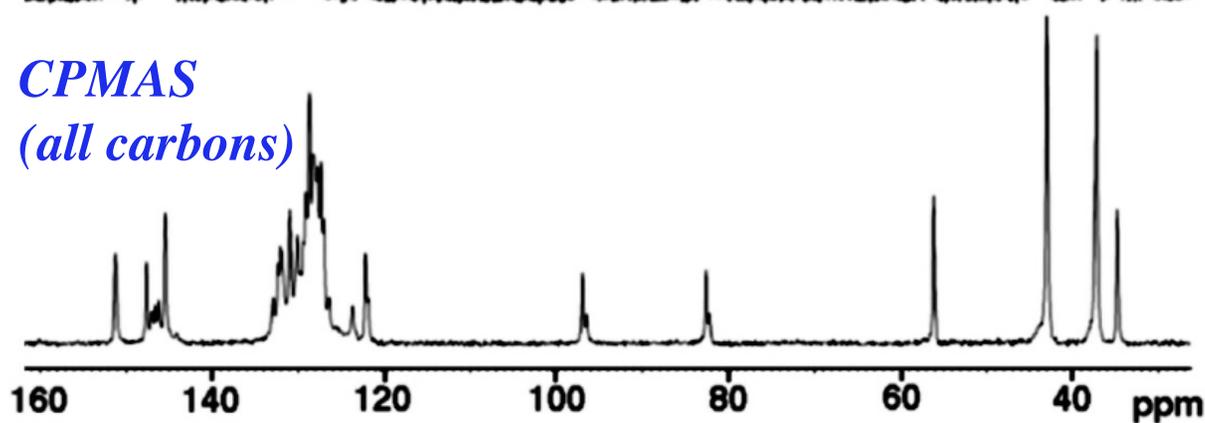




Dipolar dephasing
(only quaternary and highly mobile carbons)



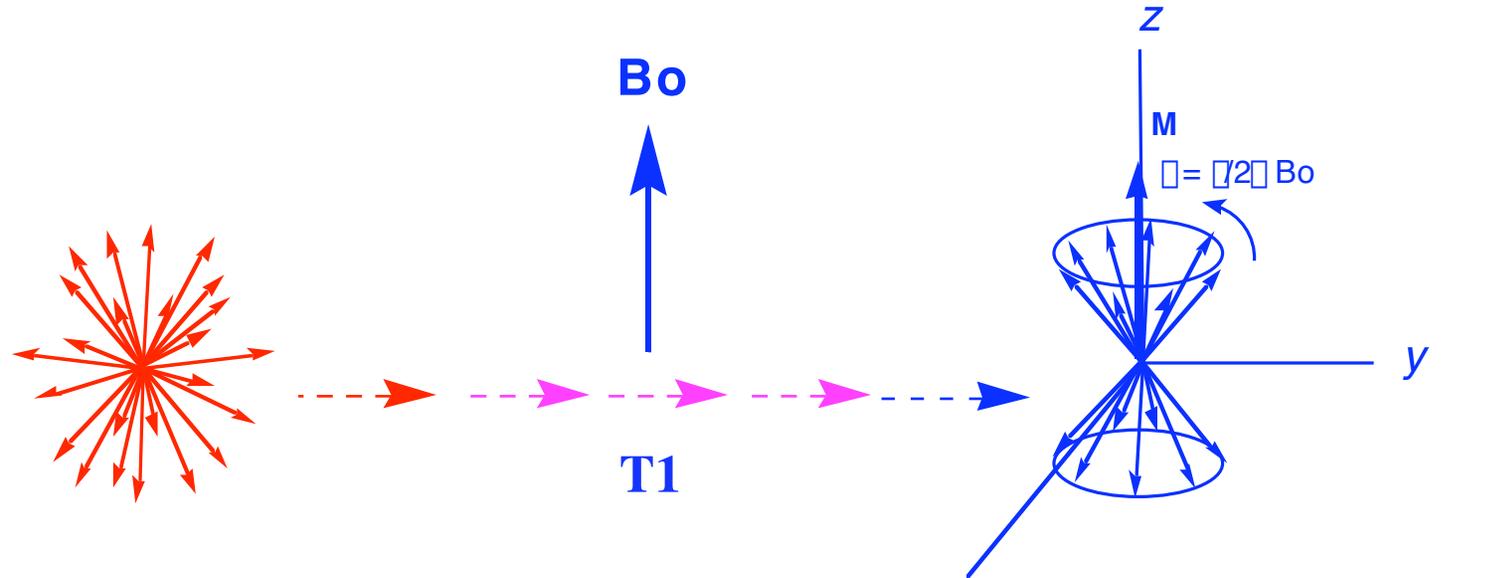
CPMAS
(all carbons)



For a more quantitative method...

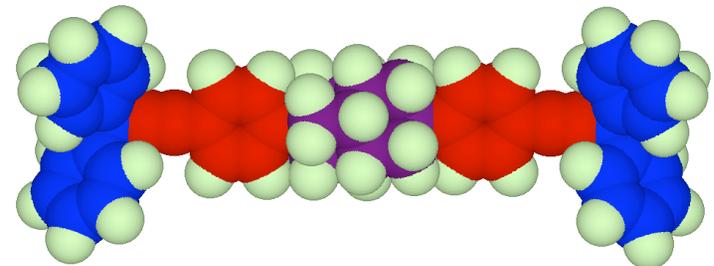
Spin Lattice Relaxation (T_1)

No Field => No Magnetization



SPIN LATTICE RELAXATION T_1 , is the time that it takes for M to reach its equilibrium value at B_0

- Spontaneous transitions would take thousands of years!*
- T_1 's are dominated by STIMULATED transitions, i.e., random local fields that oscillate at the Larmor frequency (ω_L) generated by the mechanical motion of nearby dipoles.*



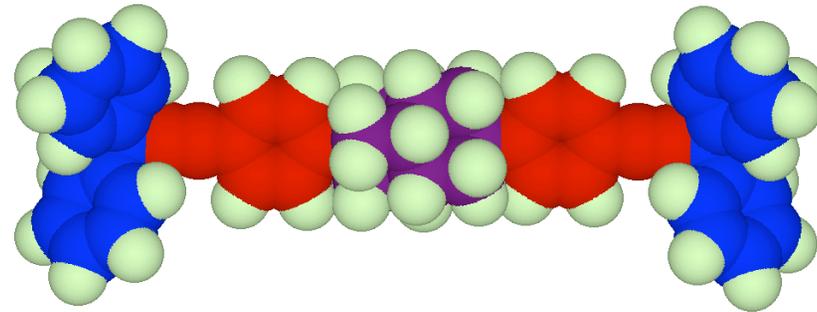
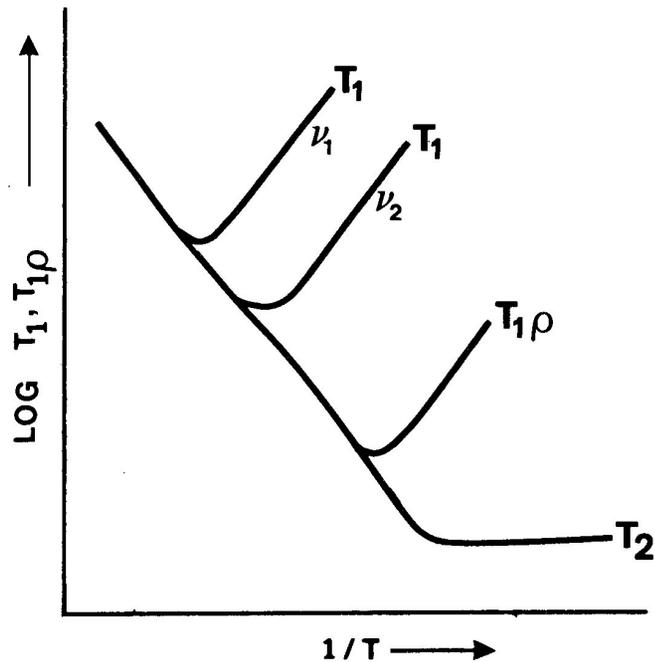
Spin-Lattice Relaxation

Kubo and Tomita relaxation expression:

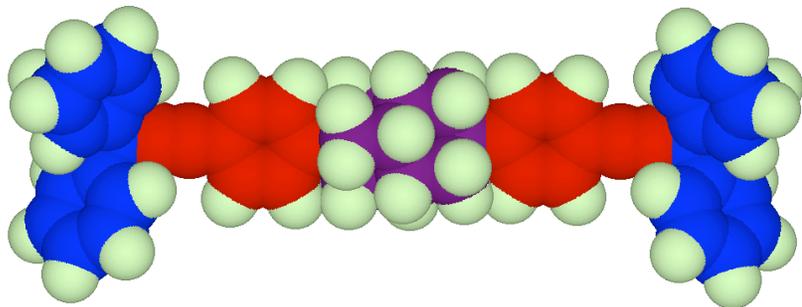
$$T_1^{-1} = C [\nu_c (1 + \nu_c^2 T_c^2)^{-1} + 4 \nu_c (1 + 4 \nu_c^2 T_c^2)^{-1}]$$

$$\nu_c = \nu_0 \exp (E_a/kT)$$

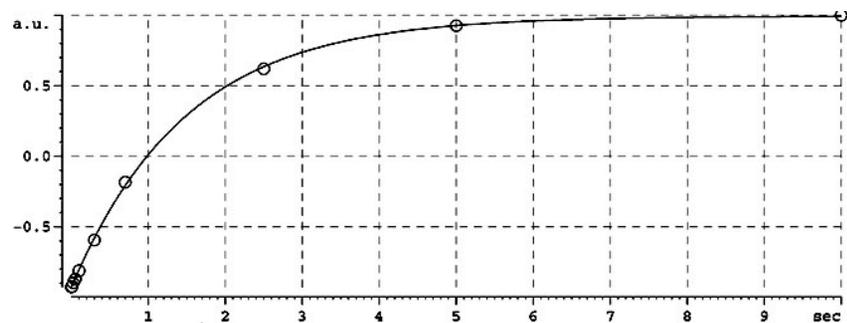
Slope = $-E_a/R$ and $+E_a/R$
 Minimum = $\nu_0 T_c$



- T_1 = Spin Lattice Relaxation
- T_{1r} = Spin Lattice Relaxation
in the Rotating Frame
- T_2 = Spin-Spin Relaxation

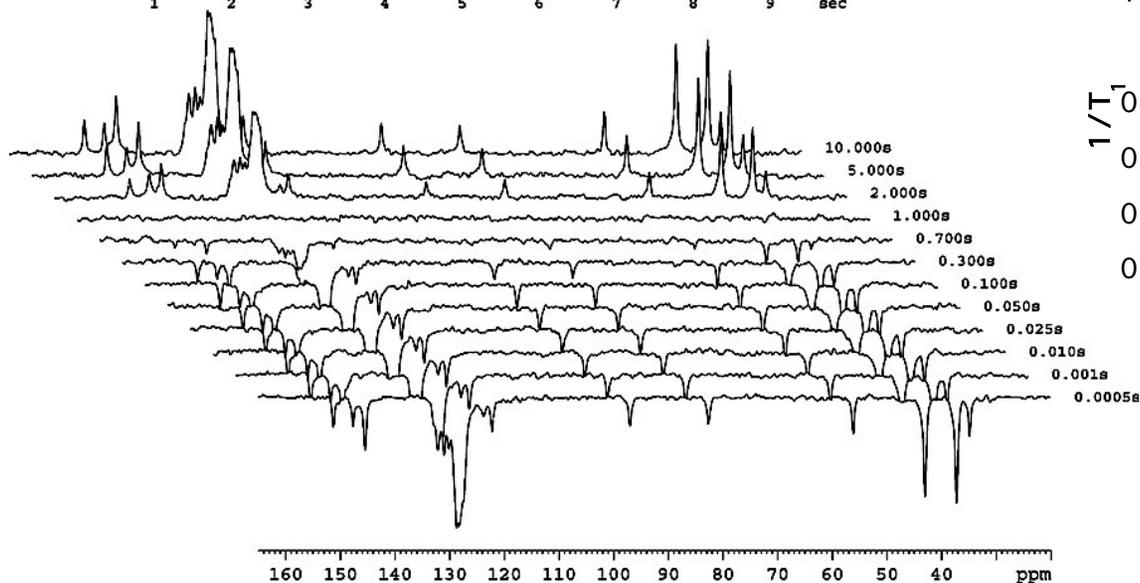


Gyroscopic Rotation By Spin-Lattice Relaxation

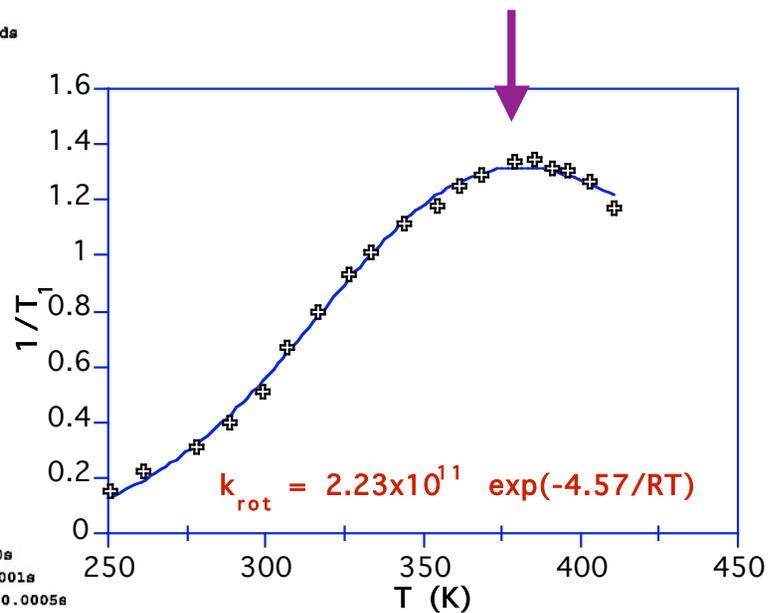


SDK1086
Diphenyl Diamantane TR
Recrystallized form DCM
T = 296 K

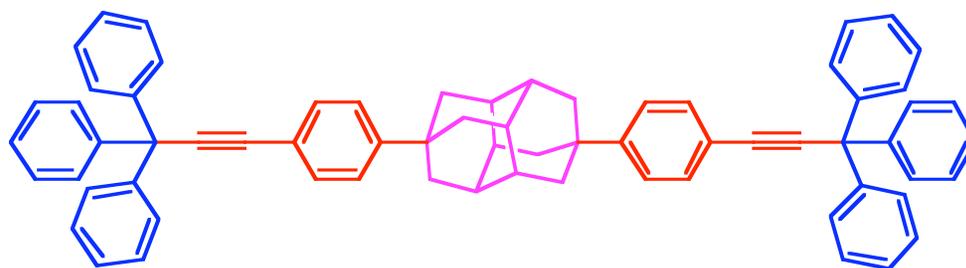
T1 = 1.5 seconds



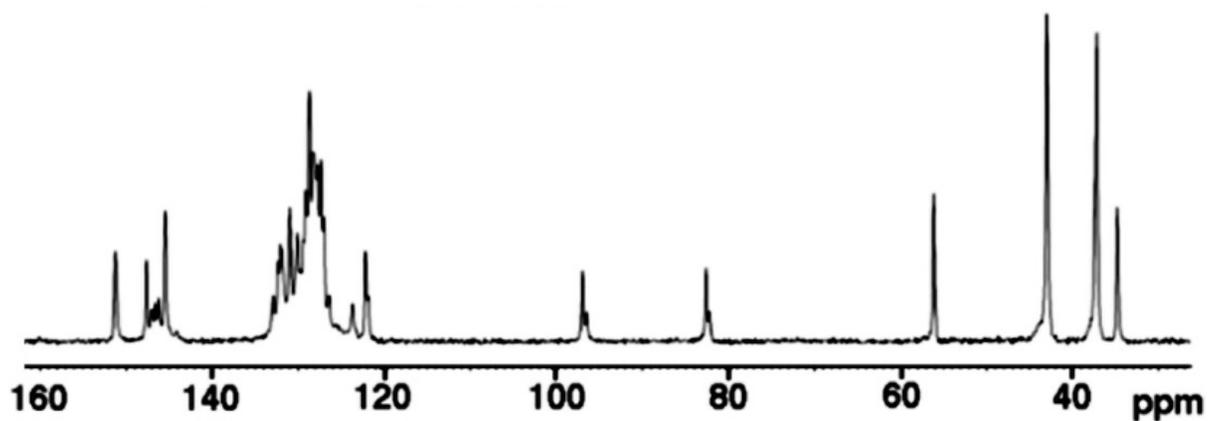
300 MHz at 100 °C

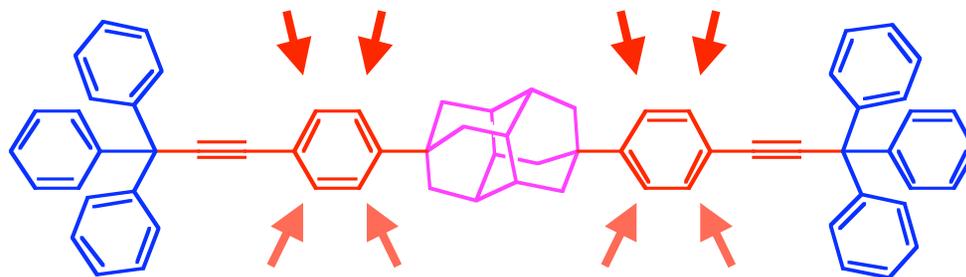


*Only ~1.6 kcal/mol above
the intrinsic barrier!*

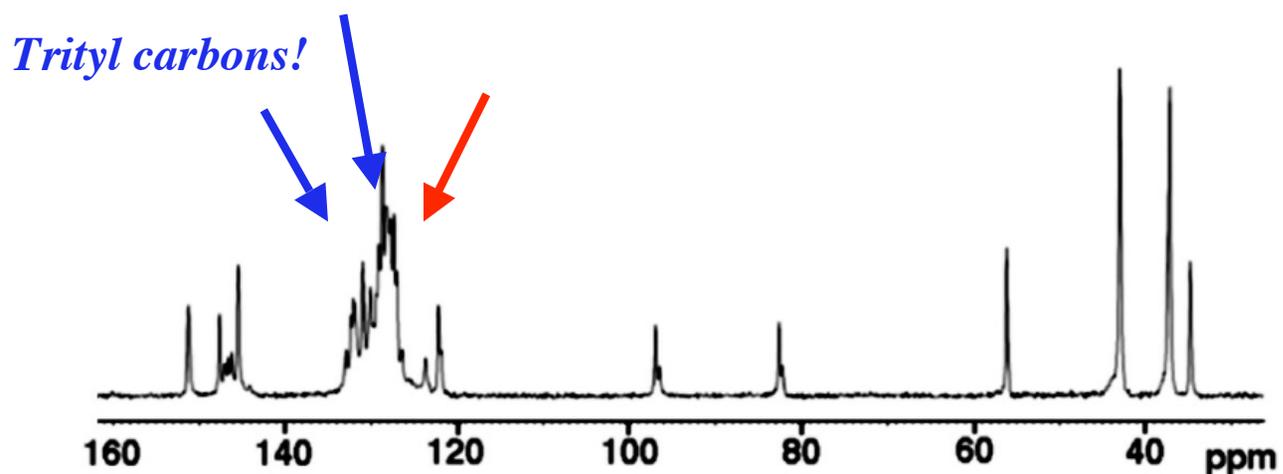


Can we use ^{13}C CPMAS VT-NMR coalescence methods to measure the rotation of the two phenylenes?



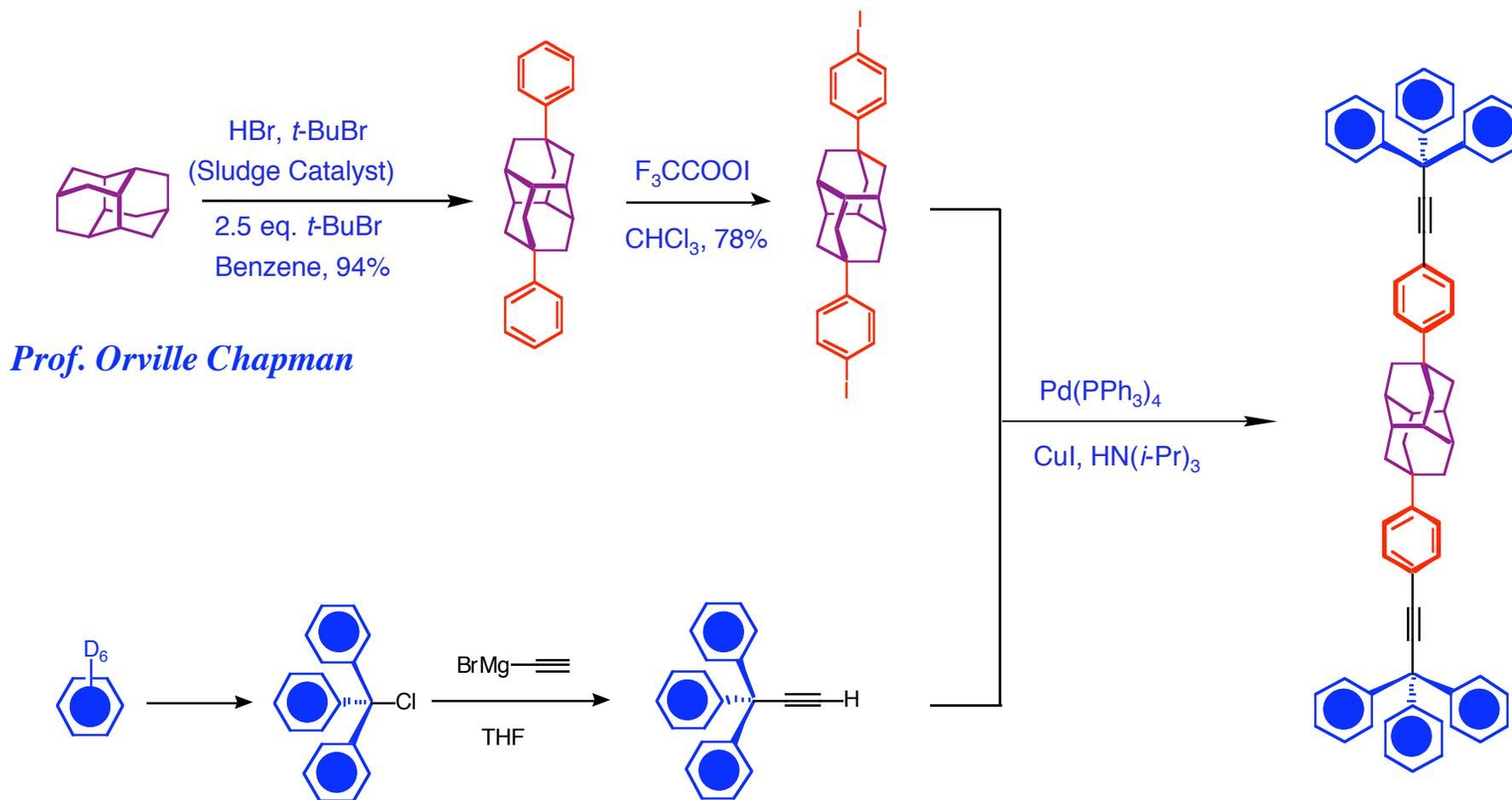


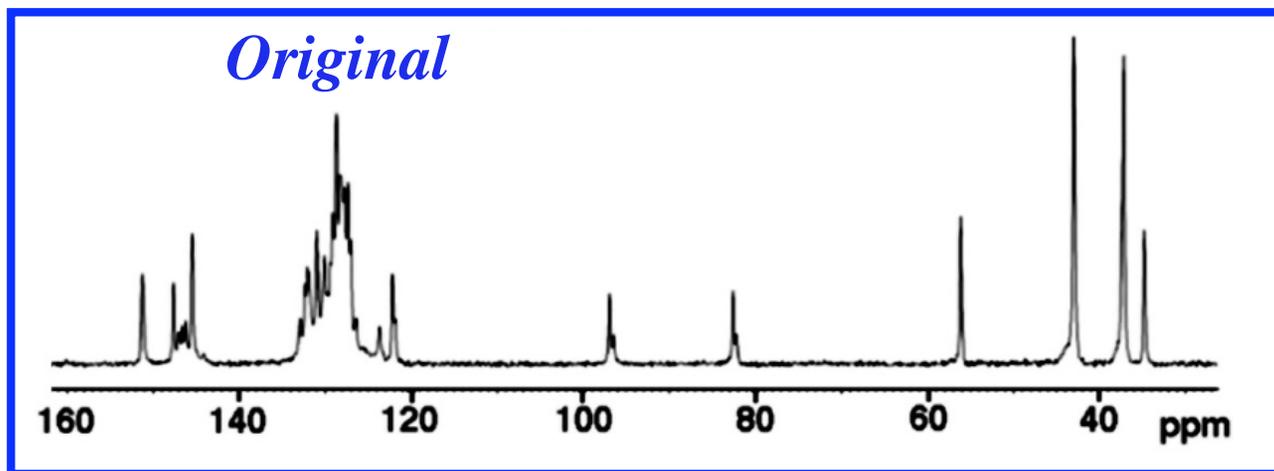
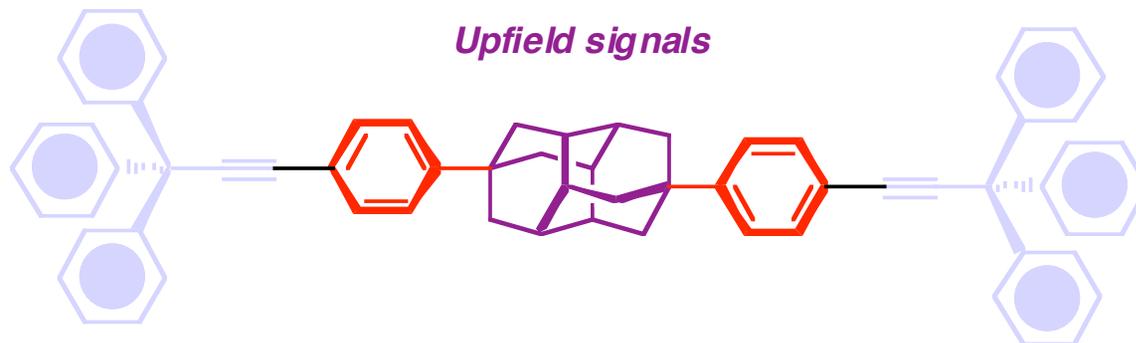
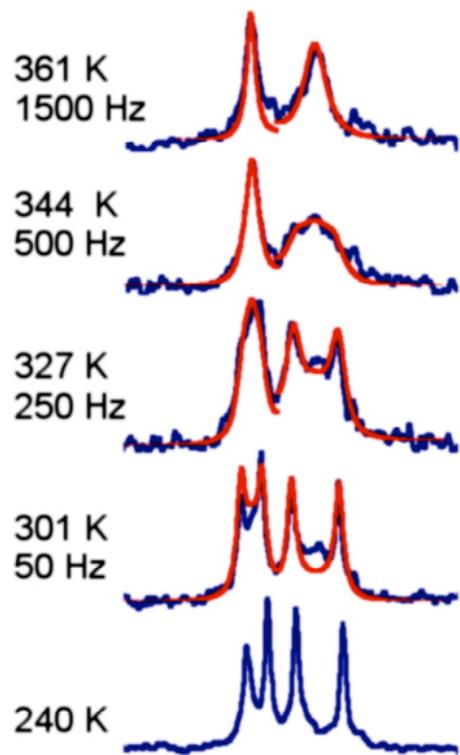
Can we use ^{13}C CPMAS VT-NMR coalescence methods to measure the rotation of the two phenylenes?

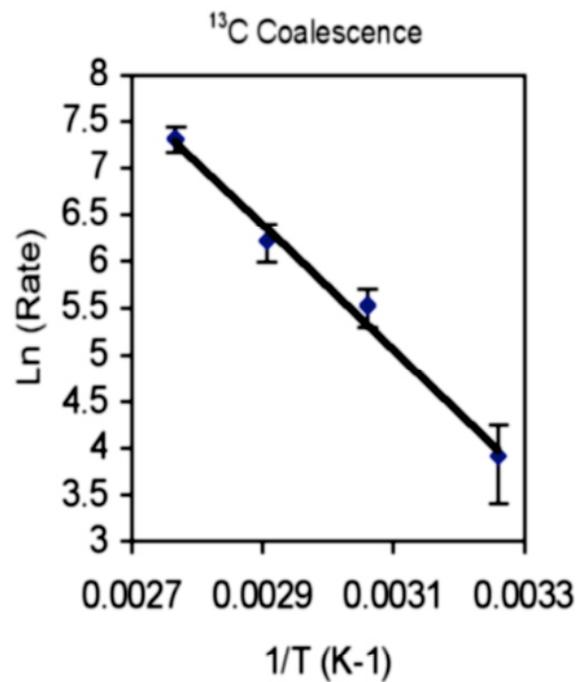
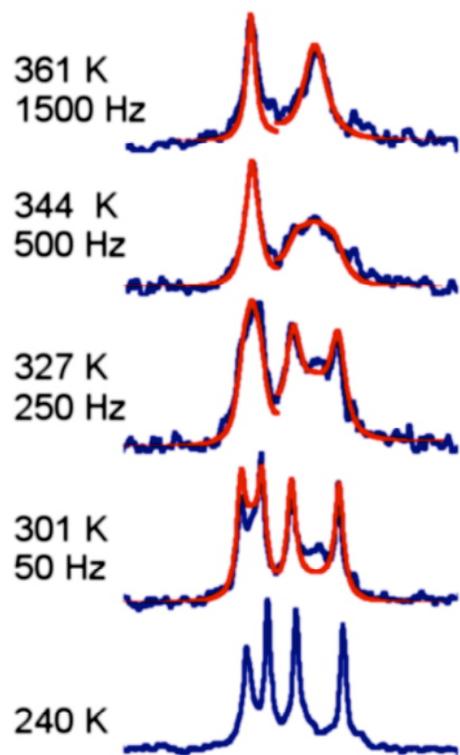


Organic Synthesis to the Rescue....

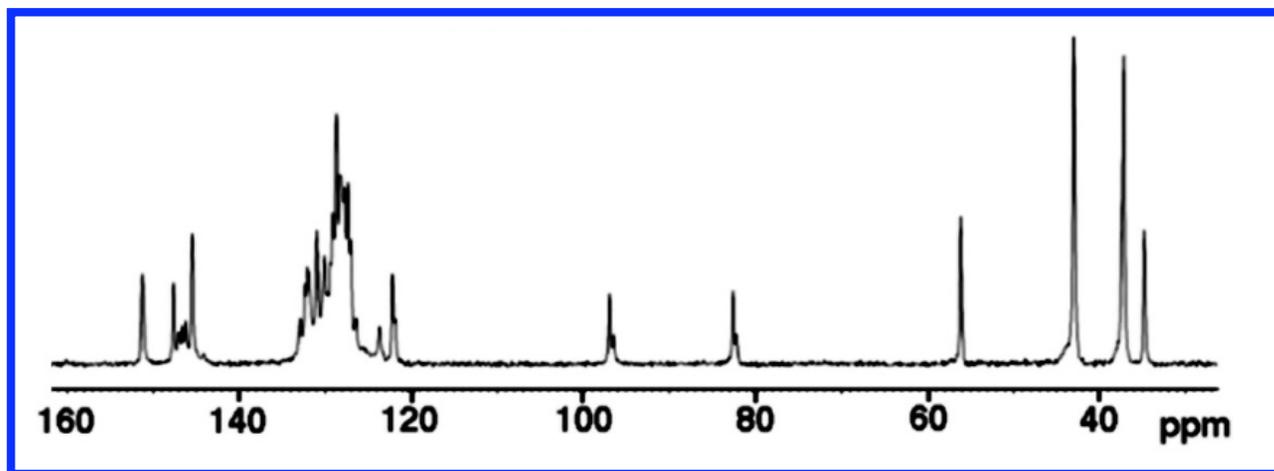
—Deuterate the trityl group... Detect only the ¹H-substituted PHENYLENES

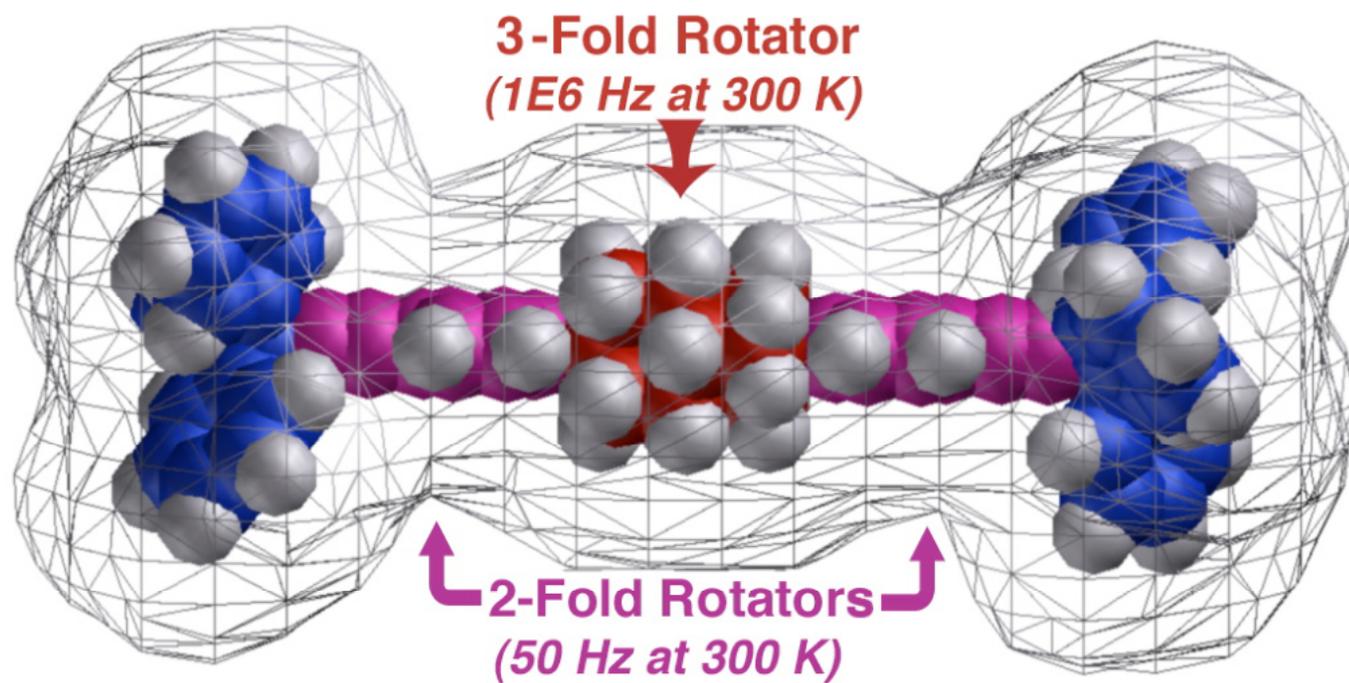






$E_a = 13.7 \pm 1.1 \text{ kcal/mol}$
 $A = 5.1 \pm 4.5 \times 10^{11} \text{ s}^{-1}$





Conclusions: *Internal Dynamics in Crystals can be “Engineered” with Information Contained in Molecular Structures. Artificial Molecular Machines Can’t be that Far*

