

# The Development of Organic Conductors, Including Semiconductors, Metals and Superconductors

## Lessons from History

# Organic Electronic(MOLECULAR) Materials

Design



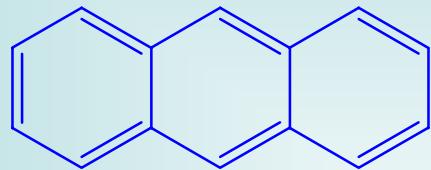
Ensemble



Properties

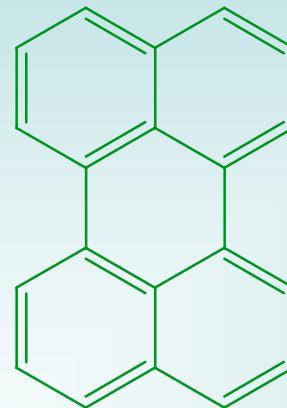
# Organic Conductors, Early History

## Charge Carrier Generation: Electron Donors



Pochettino, A., *Accad. Lincei Rend.*, **1906**,  
15, 171

Photogeneration



*facile oxidation*

Perylene (Per)

$$\text{Per(I}_2\text{)}_3 \quad \sigma_{\text{RT}} = 0.5 \text{ S/cm}$$

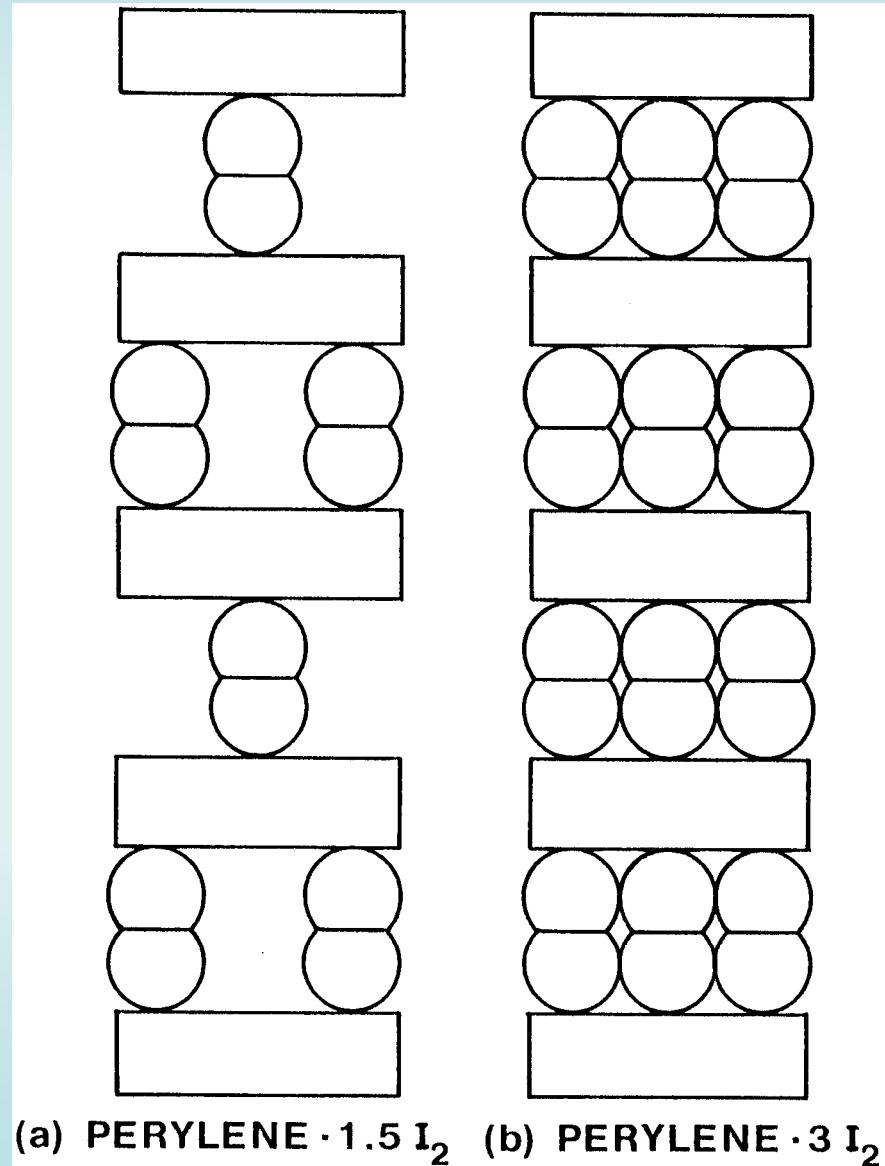
Akamatu, H.; Inokuchi, H.; Matsunaga, Y. *Nature*, **1954**, 173, 168

Labes, M. M., et al *Proc. Int. Conf. Semicon. Phys.* Prague **1960**, p 850

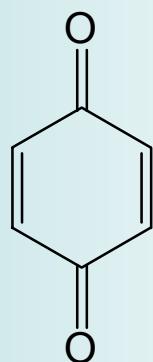
$$\text{Per(I}_2\text{)}_4 \quad \sigma_{\text{RT}} = 30 - 50 \text{ S/cm} \text{ (metal to ca 270K)}$$

Labes, M. M., et al *J. Chem. Soc. Chem. Commun.* **1979**, 329

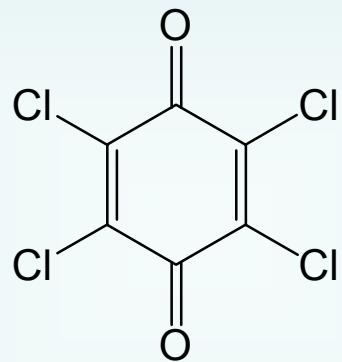
# Proposed Structure of Perylene Iodides



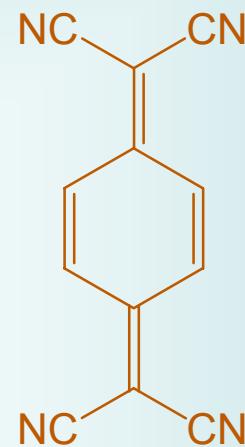
# Charge Generation: Electron Acceptors, Acceptor Repertoire



Quinone



Chloranil



TCNQ

facile reduction

Acker, D. S.; Blomstrom, D. C., *J. Am. Chem. Soc.*, **1962**, 84, 3370

## Types of C-T Complexes

Simple

$D_{(1)}T_{(1)}$

$$\sigma_{RT} = 10^{-12} - 10^{-6}$$

e.g.

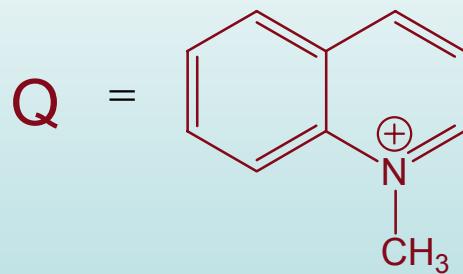
Li TCNQ

Complex

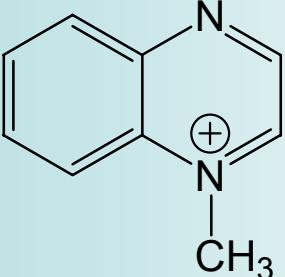
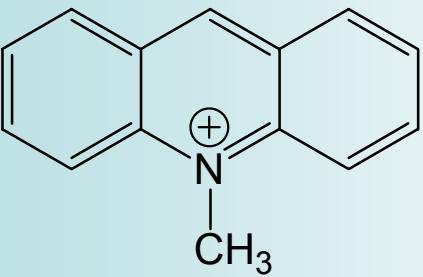
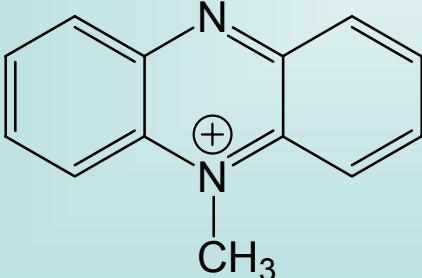
$D_n T_m$

$$\sigma_{RT} = 10 - 100$$

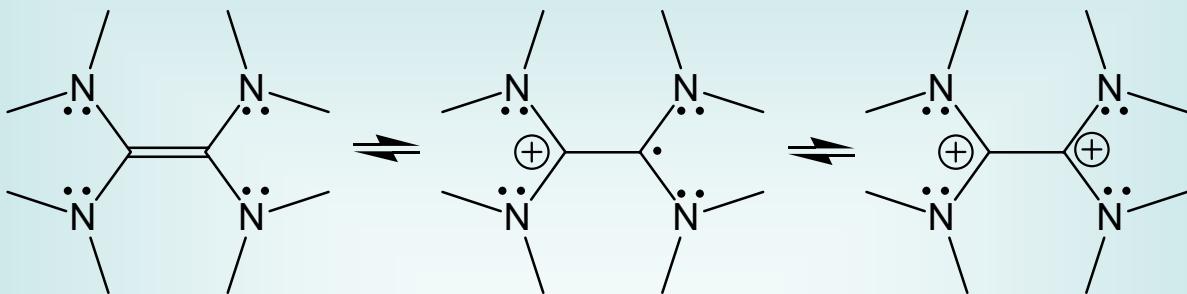
e.g.  $Q\text{TCNQ}_2$



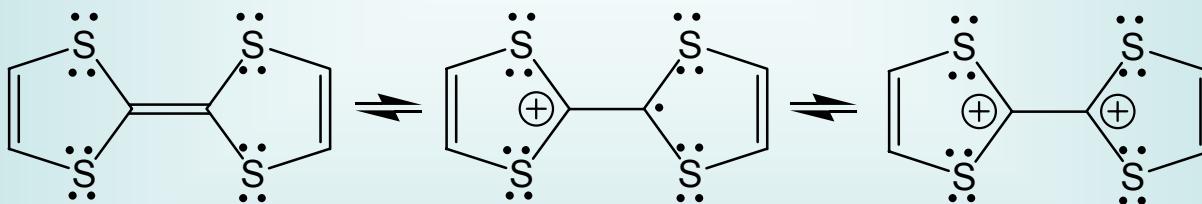
## Conductivities of Simple and Complex TCNQ Salts

Cation	Simple	Complex
	$10^{-8}$	$\approx 10^2$
	$10^{-14}$	$\approx 10^{-1}$
	$10^2$	$\approx 10^{-1}$

# The Birth of TTF



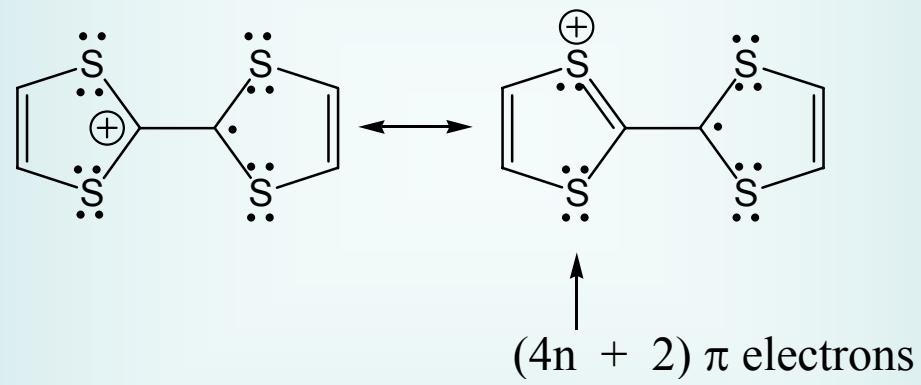
TDAE



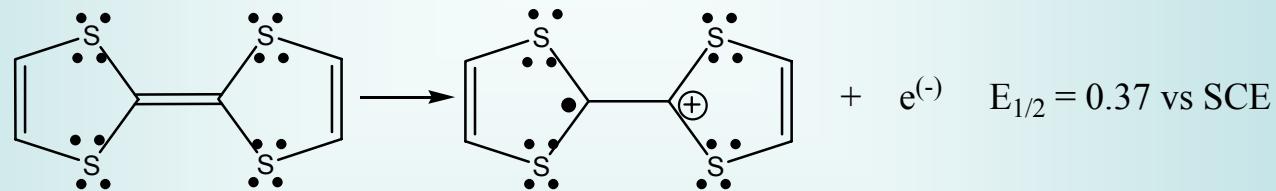
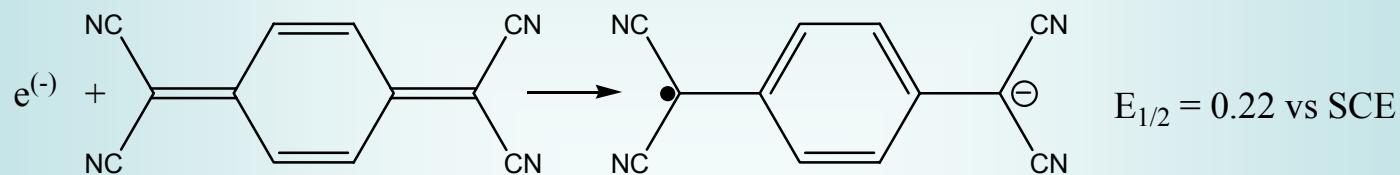
Wudl, F.; Smith, G. M.; Hufnagel, E. J. *Chem. Commun.* **1970**, 1453–1454.

Wudl, F.; Wobschall, D.; Hufnagel, E. J. *J. Am. Chem. Soc.* **1972**, 94, 670–672.

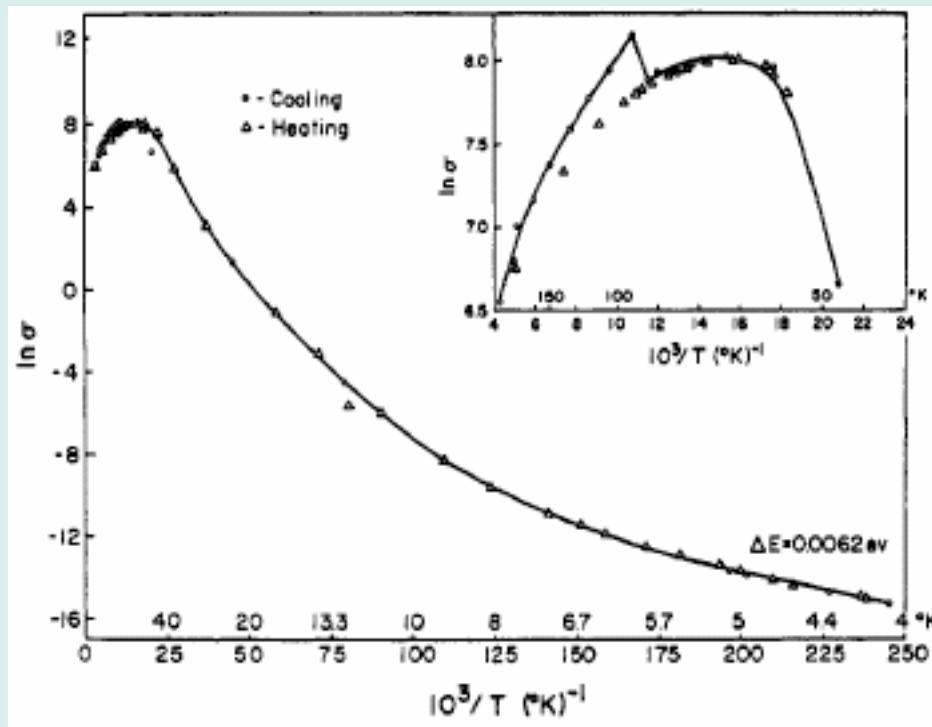
# An Important Design Feature



# The Marriage of TTF & TCNQ



# TTF-TCNQ, The First Organic Metal



John Ferraris, D. O. Cowan, V. Walatka, Jr., J. H. Perlstein *J. Am. Chem. Soc.* **1973**, *95*, 948.

# The “Giant Conductivity Peak” Phenomenon

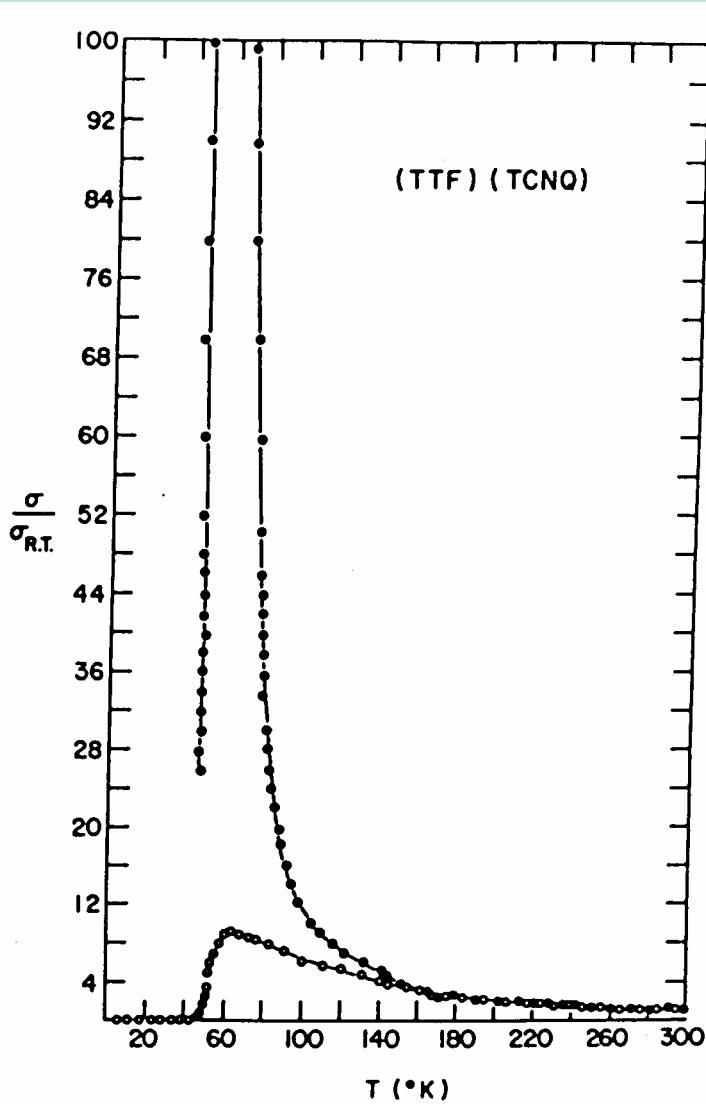
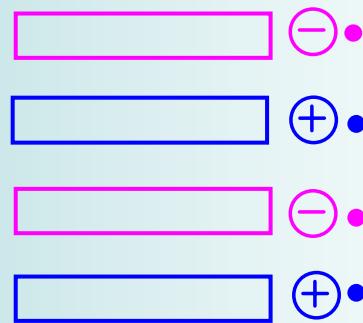


FIG. 3. Temperature dependence of the conductivity of (TTF)(TCNQ) single crystal ( $- \bullet - \bullet -$ ) and of (TTF)(TCNQ) typical crystals ( $- \circ - \circ -$ ).

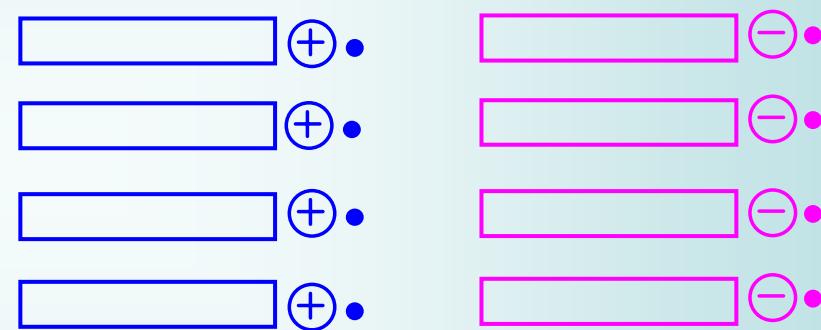
Coleman, et al, *Sol. State Commun.*  
1973, 12, 1125 - 1132

# Two Possibilities of Flat Molecule Ensembles

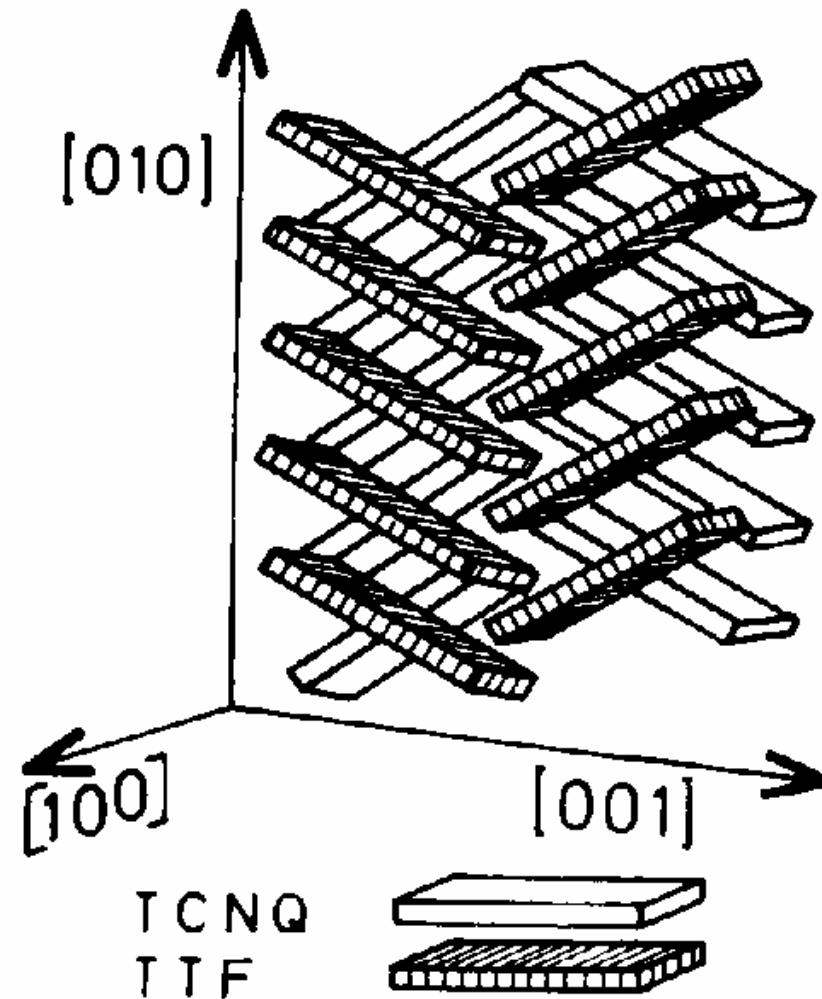
Alternating Stack



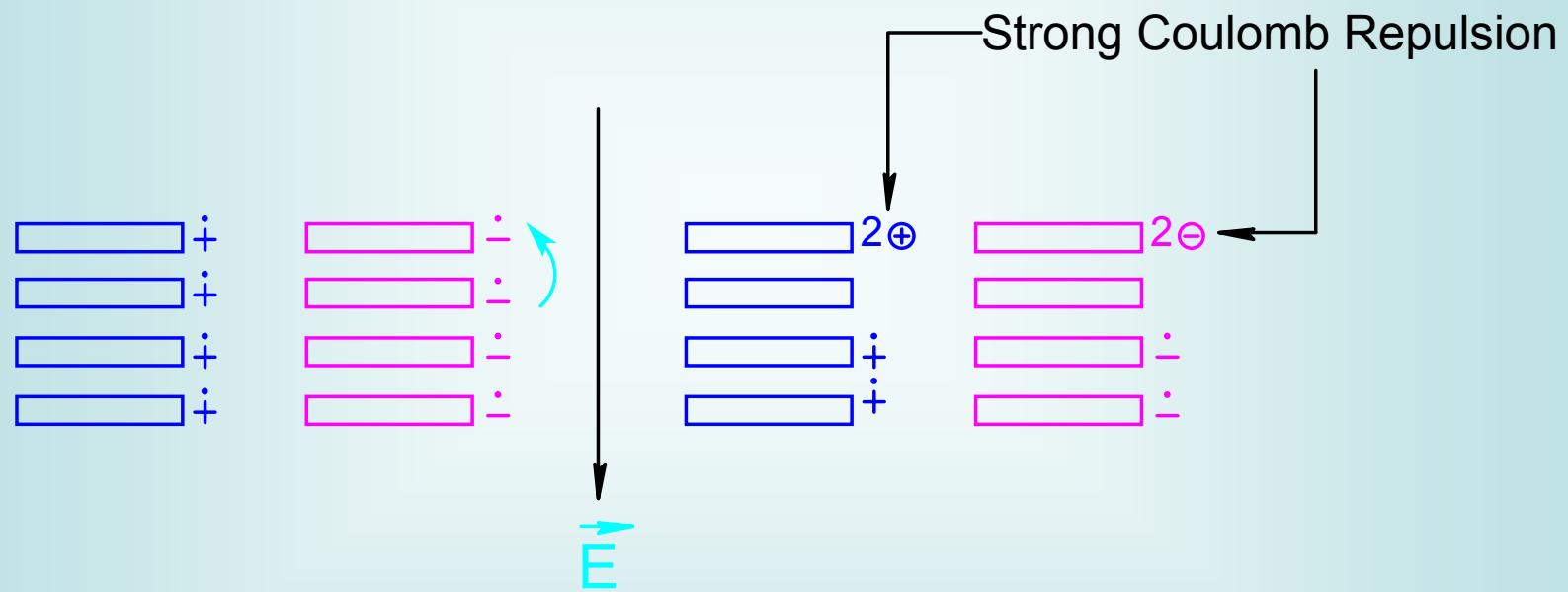
Segregated Stack



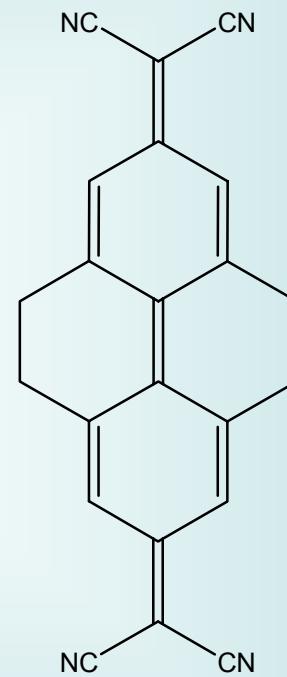
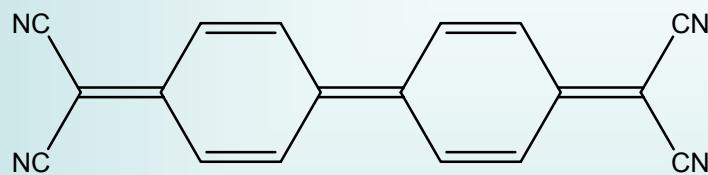
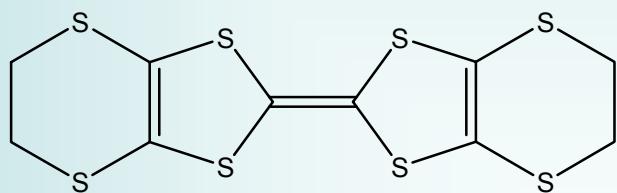
# The Structure of TTF-TCNQ



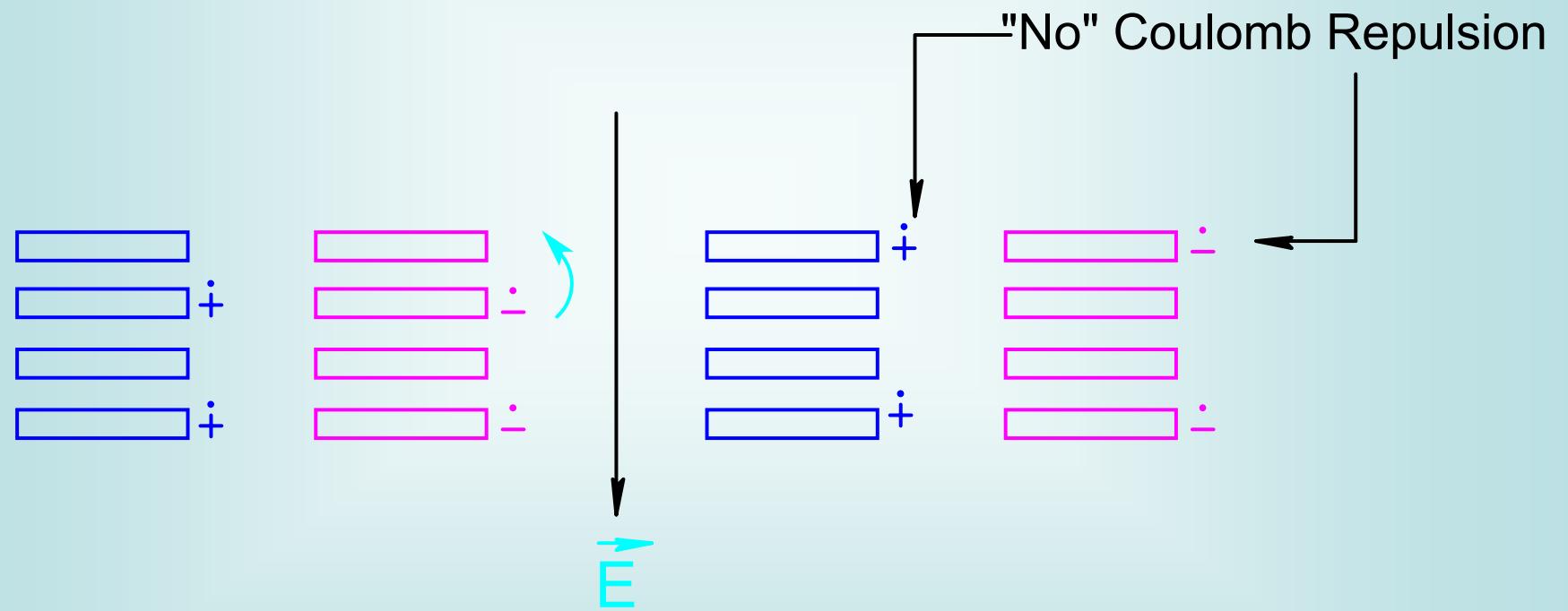
# A Problem with Transport in a $\frac{1}{2}$ Filled Stack (Band)



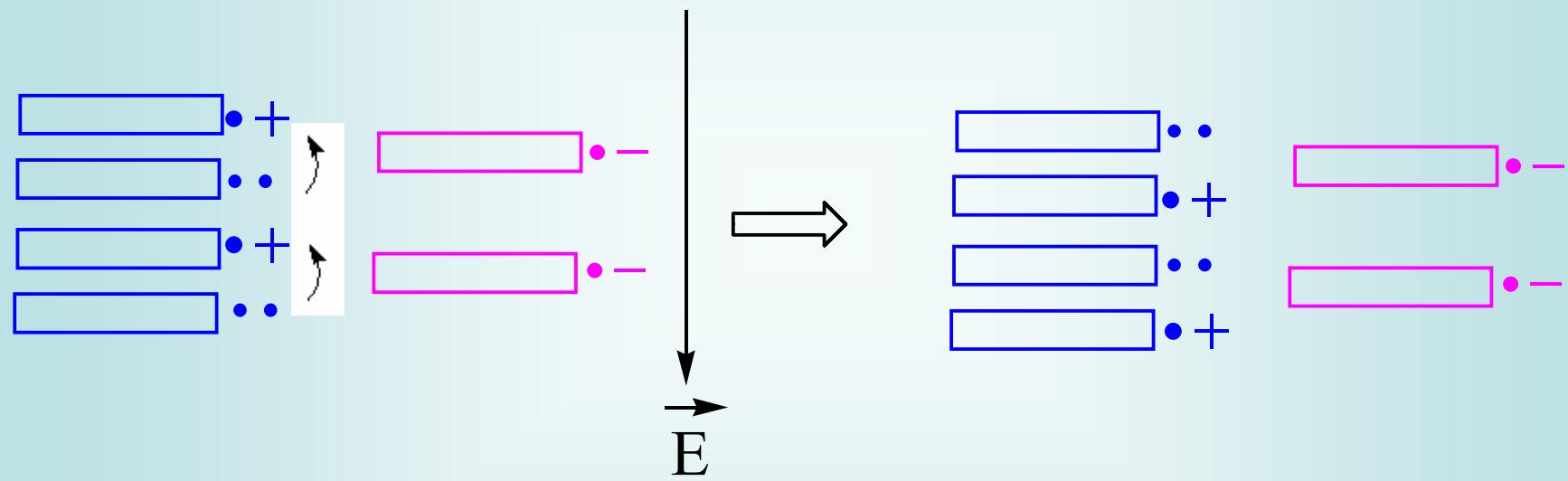
# Responding to Physicists Recipes



# Soos-Torrance Model (Mixed Valence)

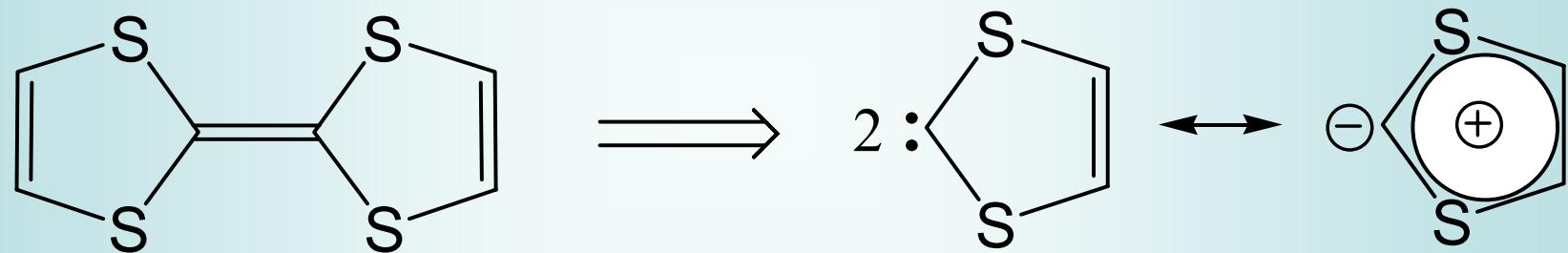


# Facile Conductivity in Complex Salts of TCNQ

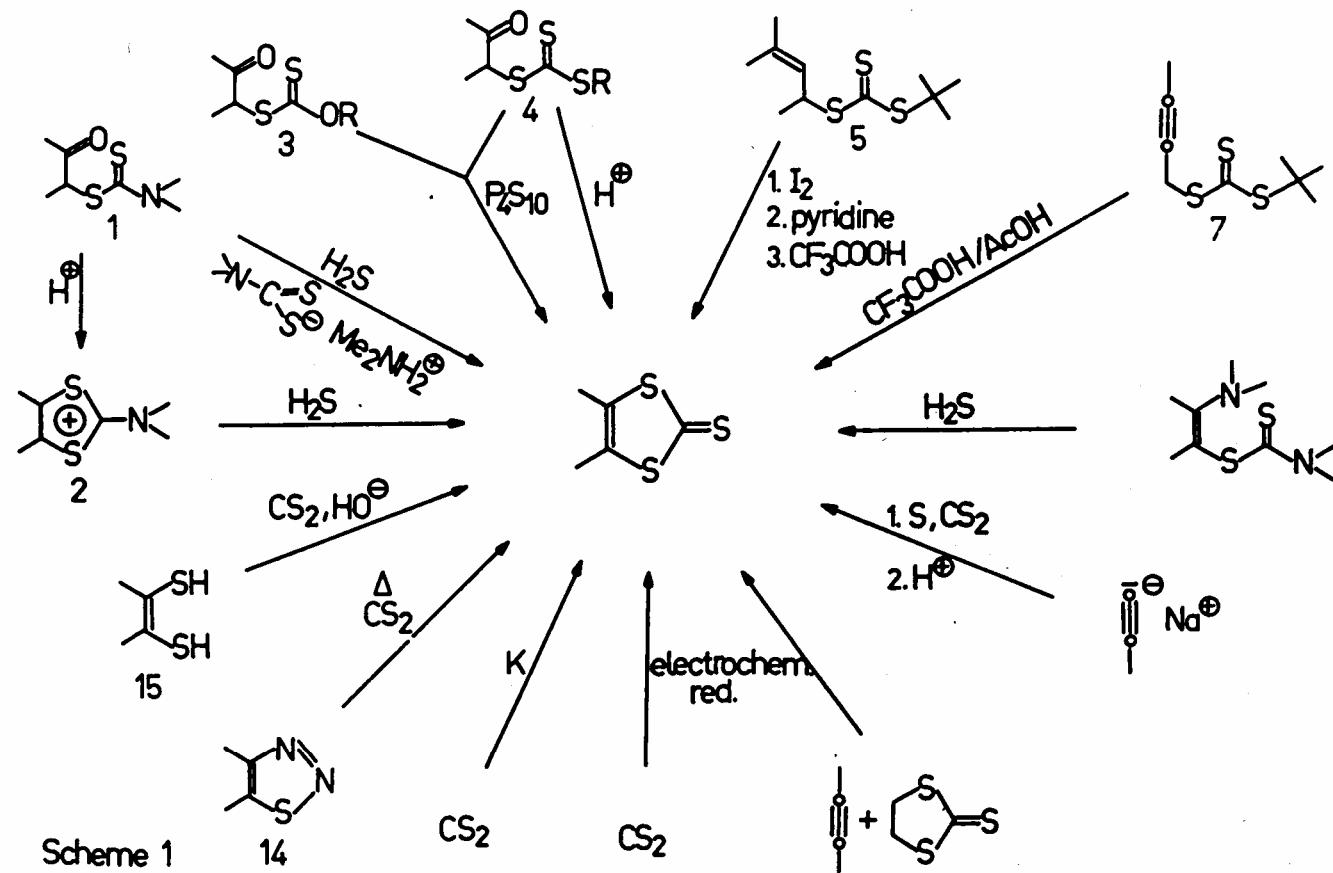


# Synthesis of TTF and Derivatives

## Strategy

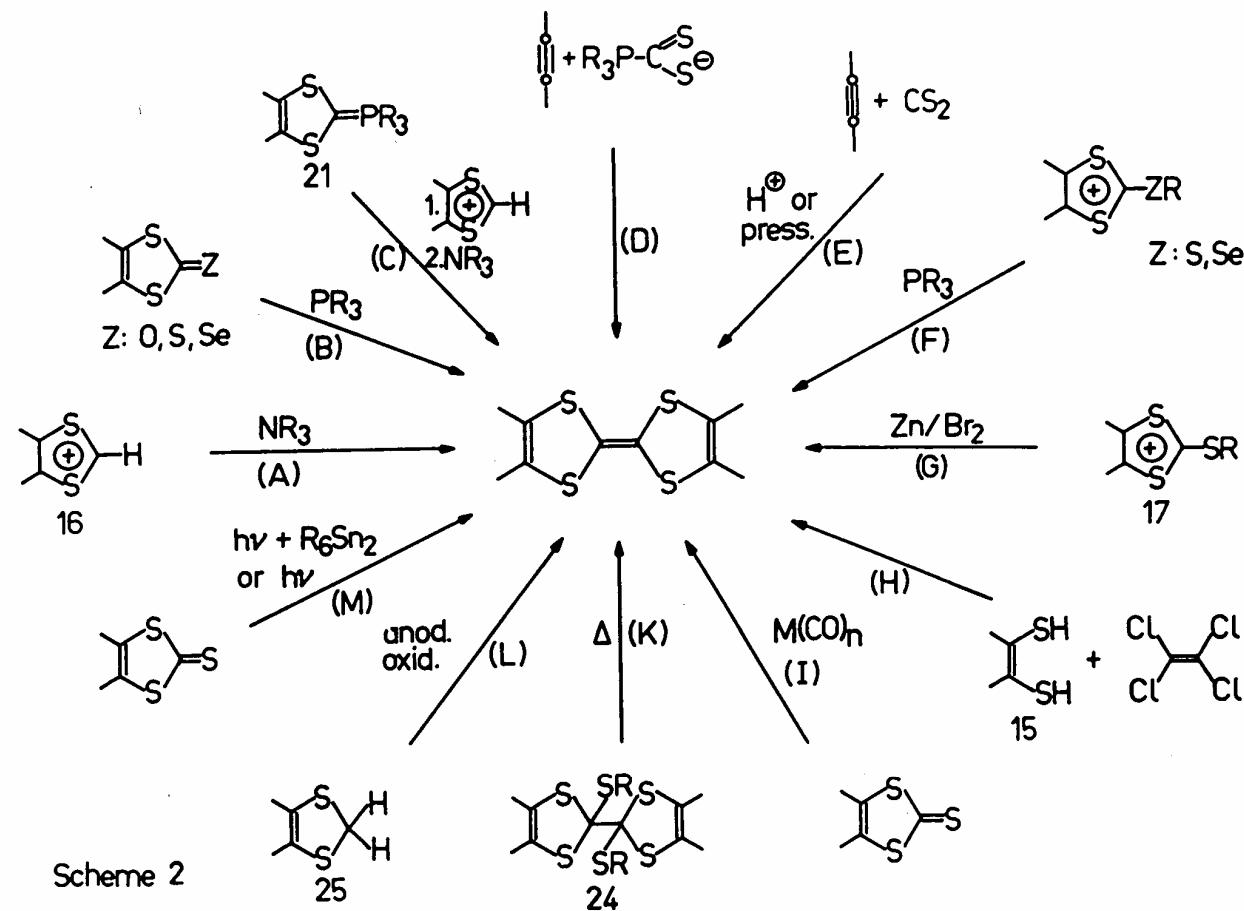


# Synthesis of 1,3-Dithiole-2-thiones

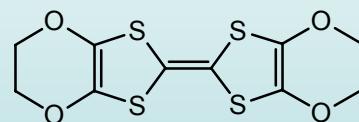
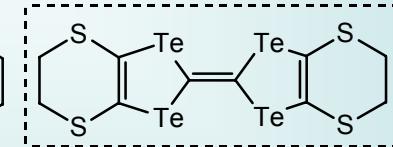
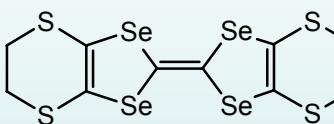
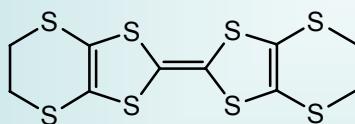
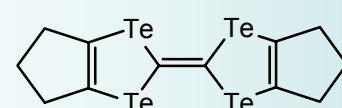
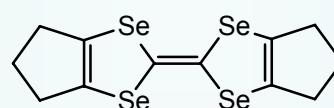
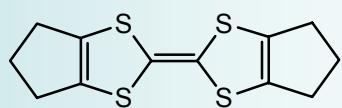
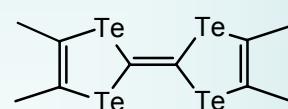
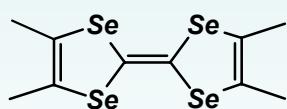
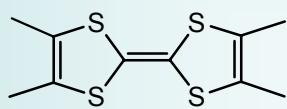
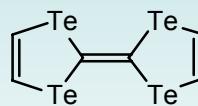
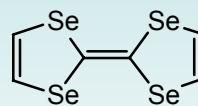
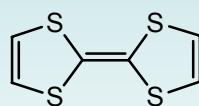


Schukat, G.; Richter, A.M.; Fanghänel, *Sulfur Reports*, 1987, 7, 155 - 240

# Synthesis of TTFs

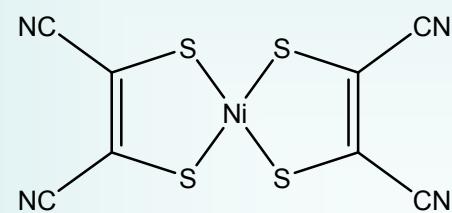
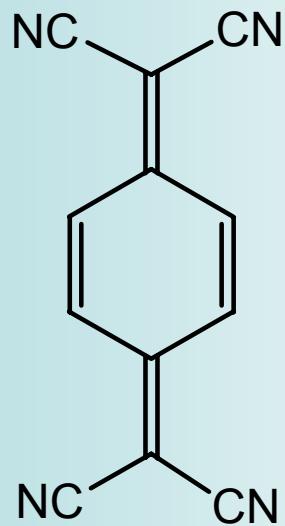


# The Most Popular and Unusual Tetrachalcogen Fulvalenes

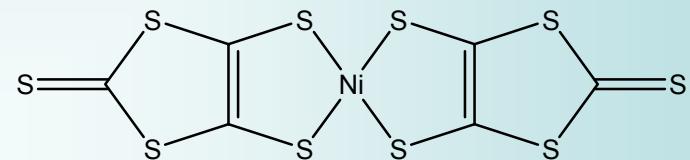


**S, Se** = Superconducting; **[ ]** = not yet prepared

# The Most Popular Acceptors



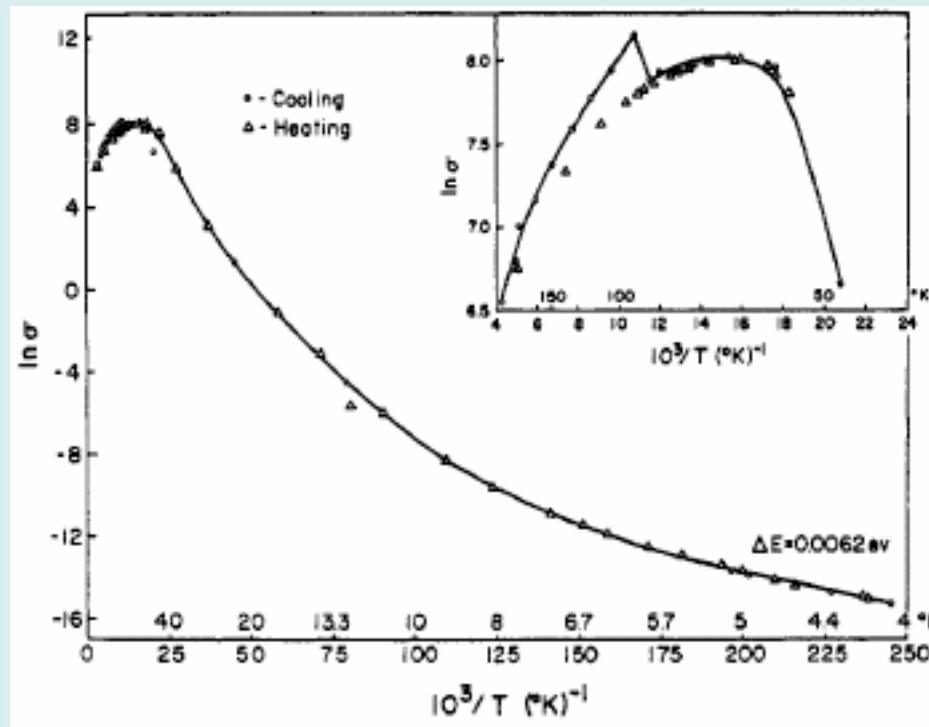
mnt



dmit

Metal Dithiolenes

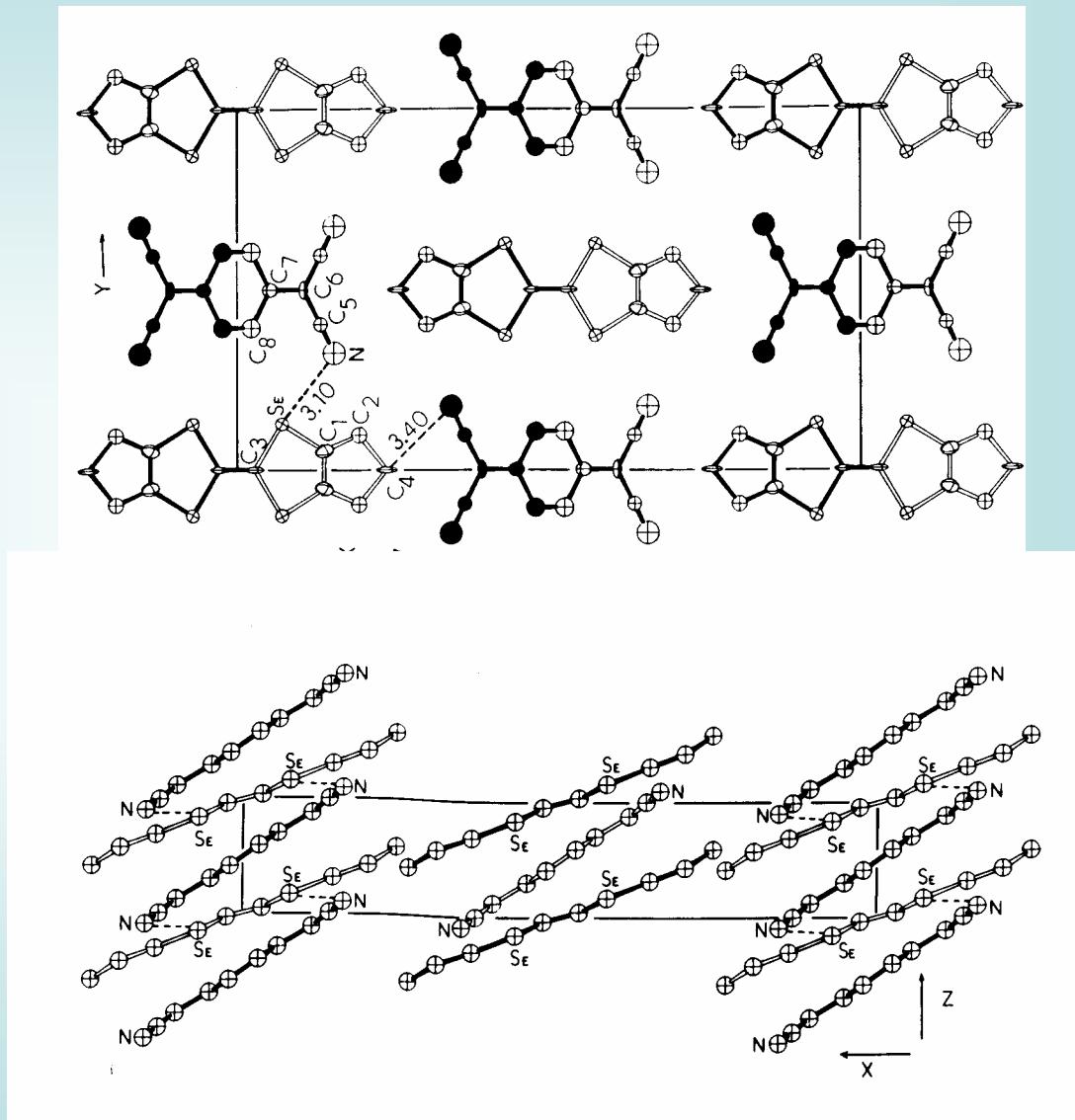
# TTF-TCNQ, The First Organic Metal ?



Why the loss in conductivity?

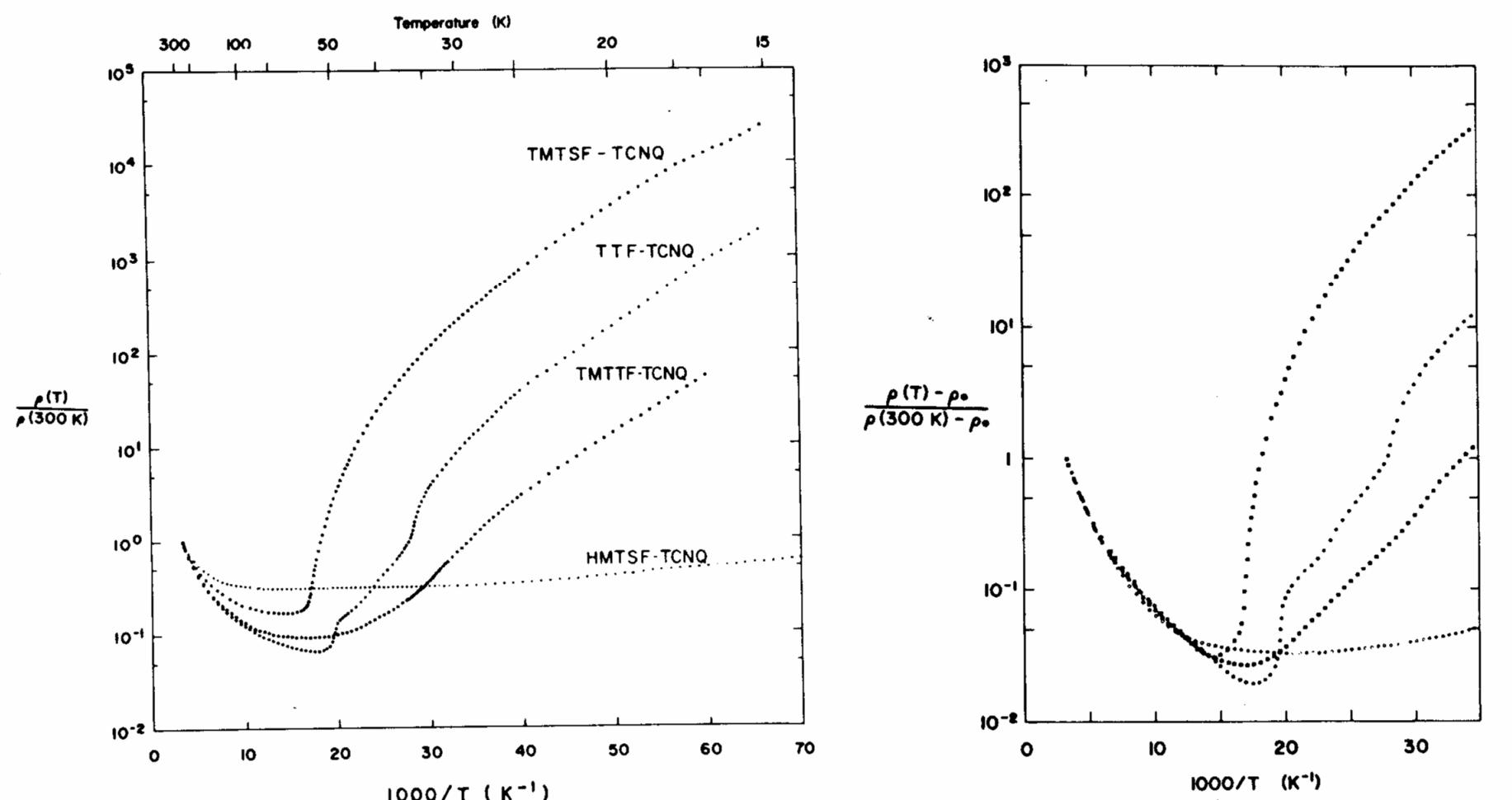
# Dealing with the Peierls Catastrophe: Increase in Dimensionality and Disorder

HMTSF-TCNQ



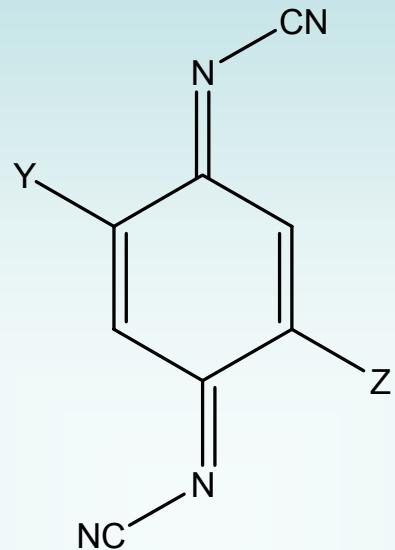
Bloch, A.N.; Carruthers, T.F.; Poehler, T.O.; Cowan, D.O. in "Chemistry and Physics of One-Dimensional Metals, Keller, H.J., Ed; Plenum, NY 1977; pp 47 -85

# Comparative Temperature-Dependent Resistivities



Bloch, A.N.; Carruthers, T.F.; Poehler, T.O.; Cowan, D.O. in "Chemistry and Physics of One-Dimensional Metals", Keller, H.J., Ed; Plenum, NY 1977; pp 47 -85

# The DCNQI Story: A Truly 3-D Molecular Metal



$Y = Z = \text{Me}$

$Y = \text{Br}, Z = \text{I}$

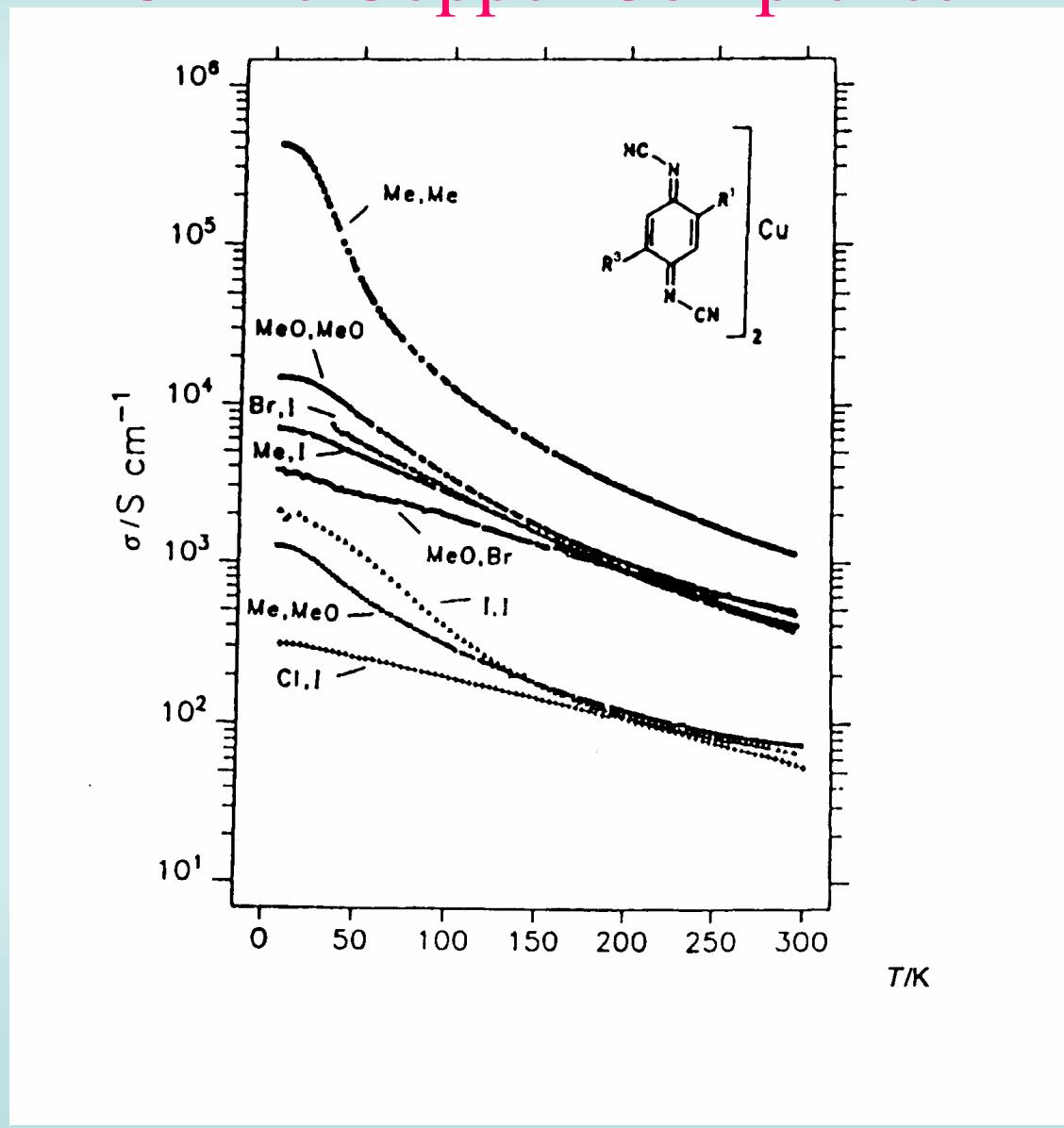
$Y = Z = \text{I}$

$Y = \text{Cl}, Z = \text{I}$

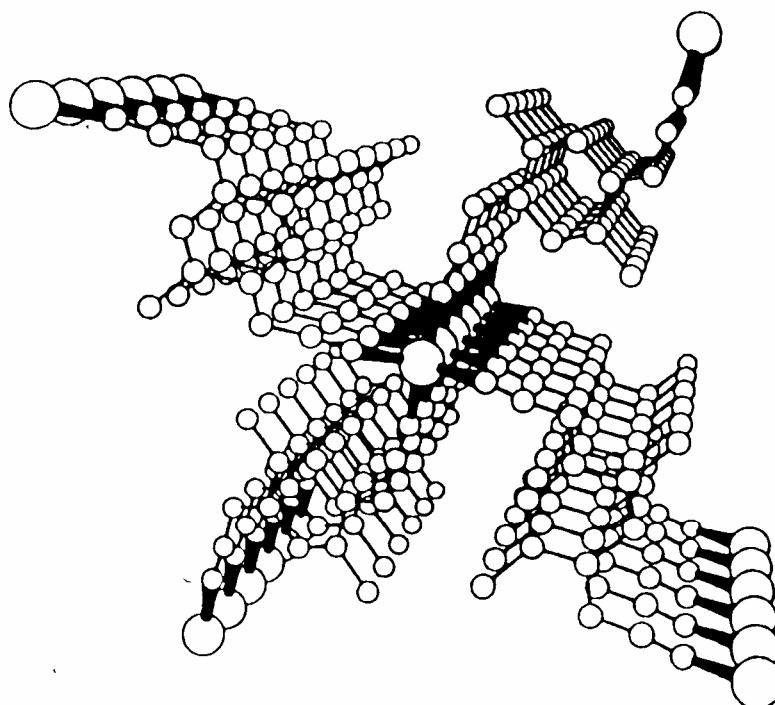
$Y = Z = \text{MeO}$

Etc, etc

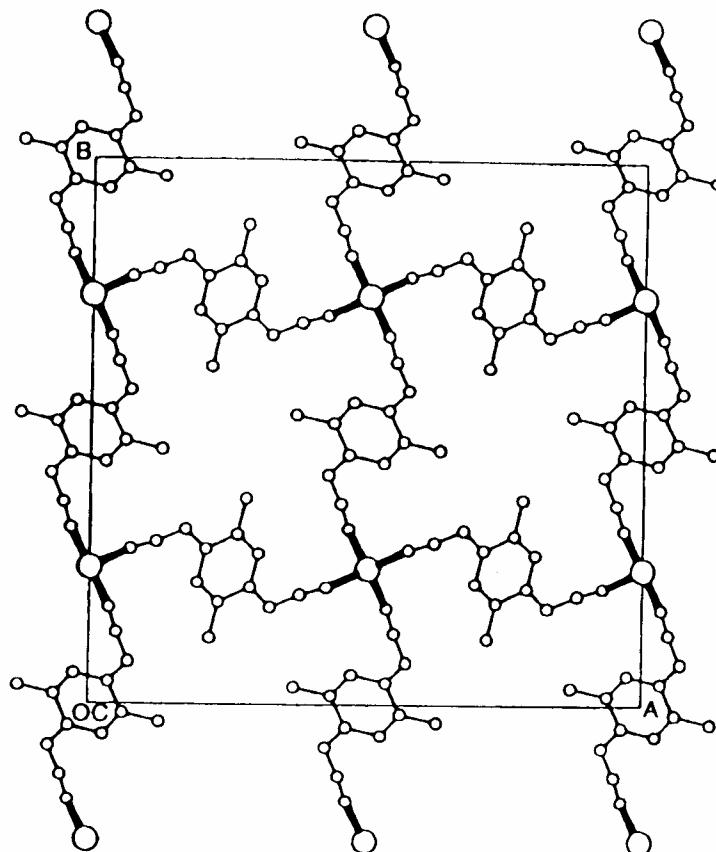
# Temperature Dependence of the Conductivity Of the Copper Complexes



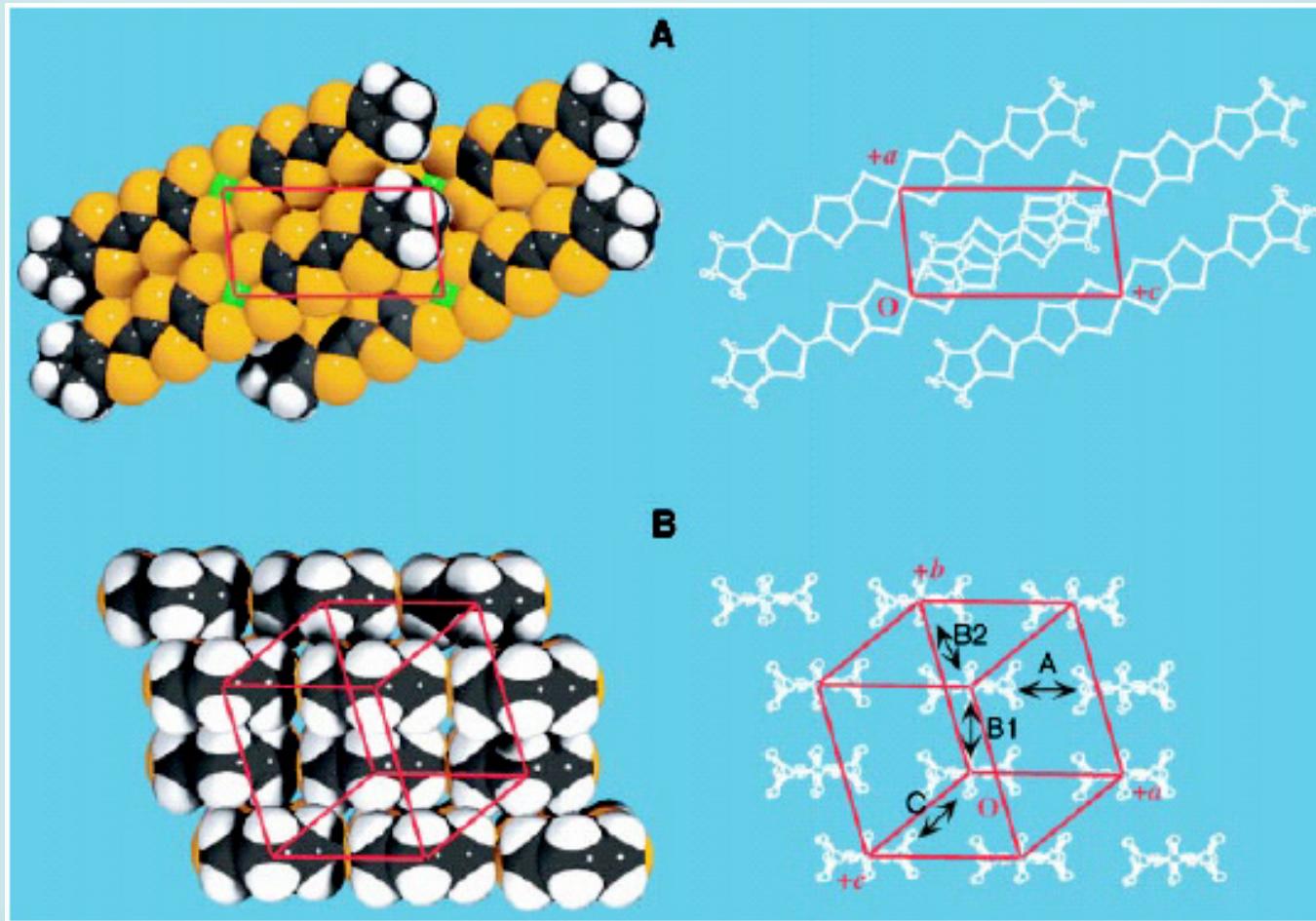
# Copper-Mediated 3-Dimensionality



# View of a Single Layer

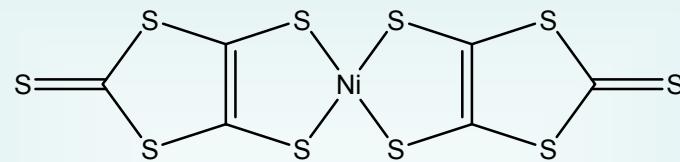


# A Molecular Metal Designer's Dream: The Single Component Metal



Hisashi Tanaka, Yoshinori Okano, Hayao Kobayashi, Wakako Suzuki, Akiko Kobayashi\* *Science*, **2001**, *291*, 285.

# The Key Building Block

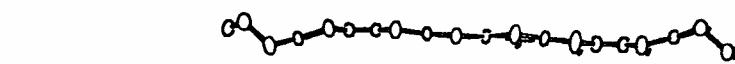
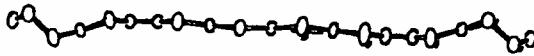


dmit

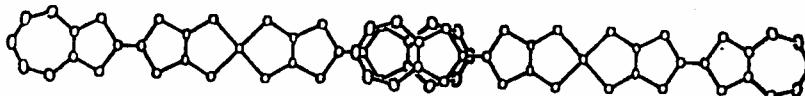
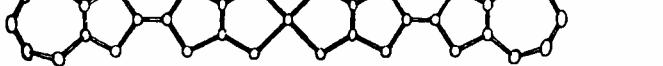
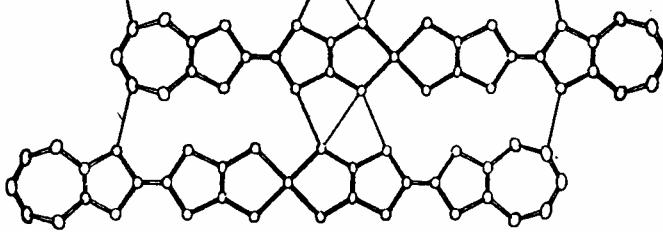
Metal Dithiolenes



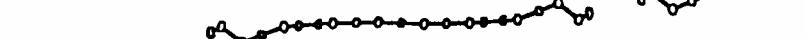
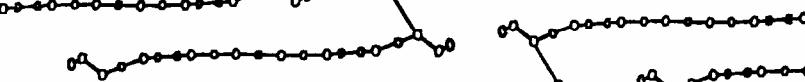
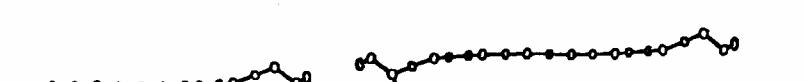
(a)



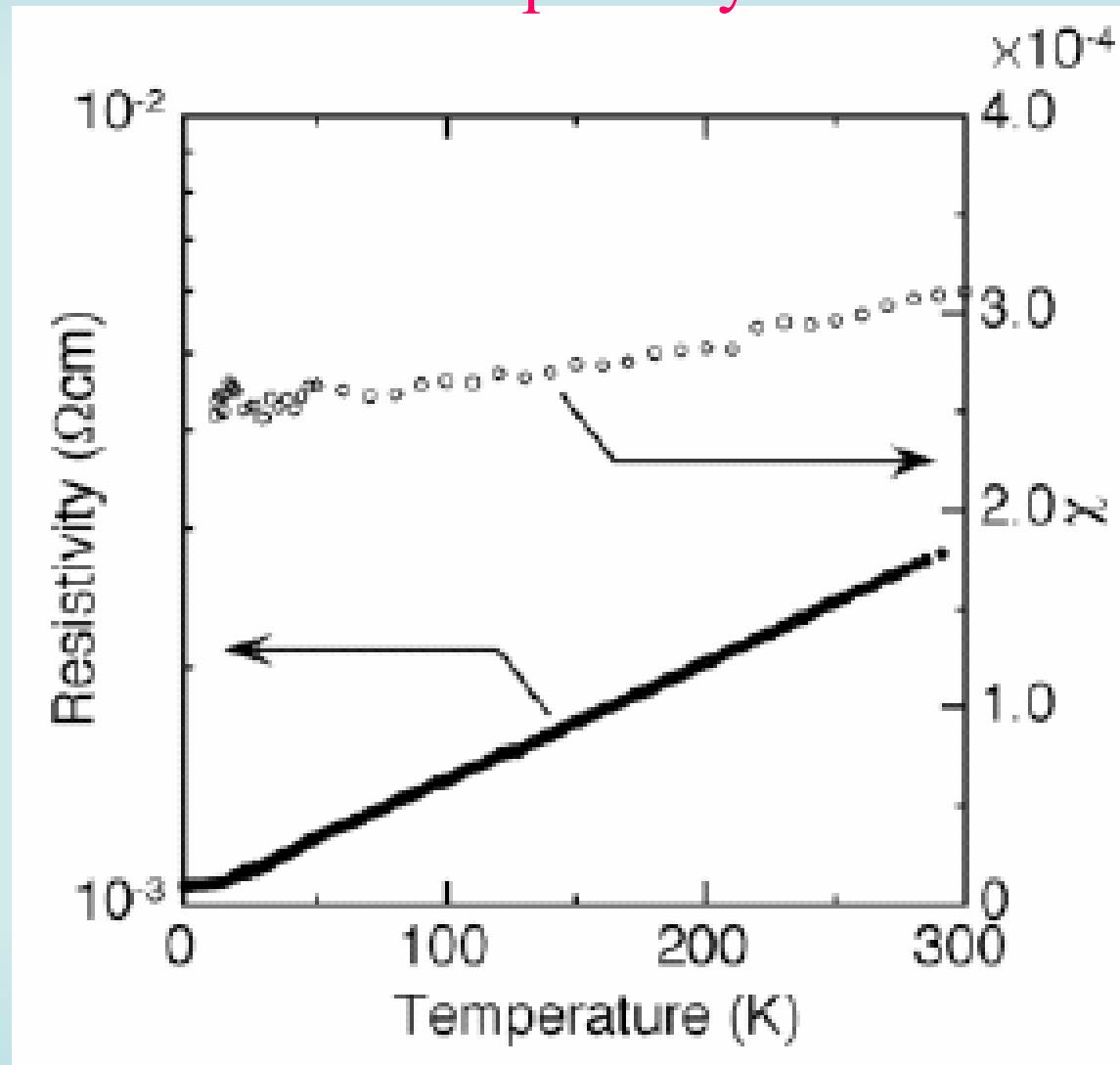
(b)



(c)

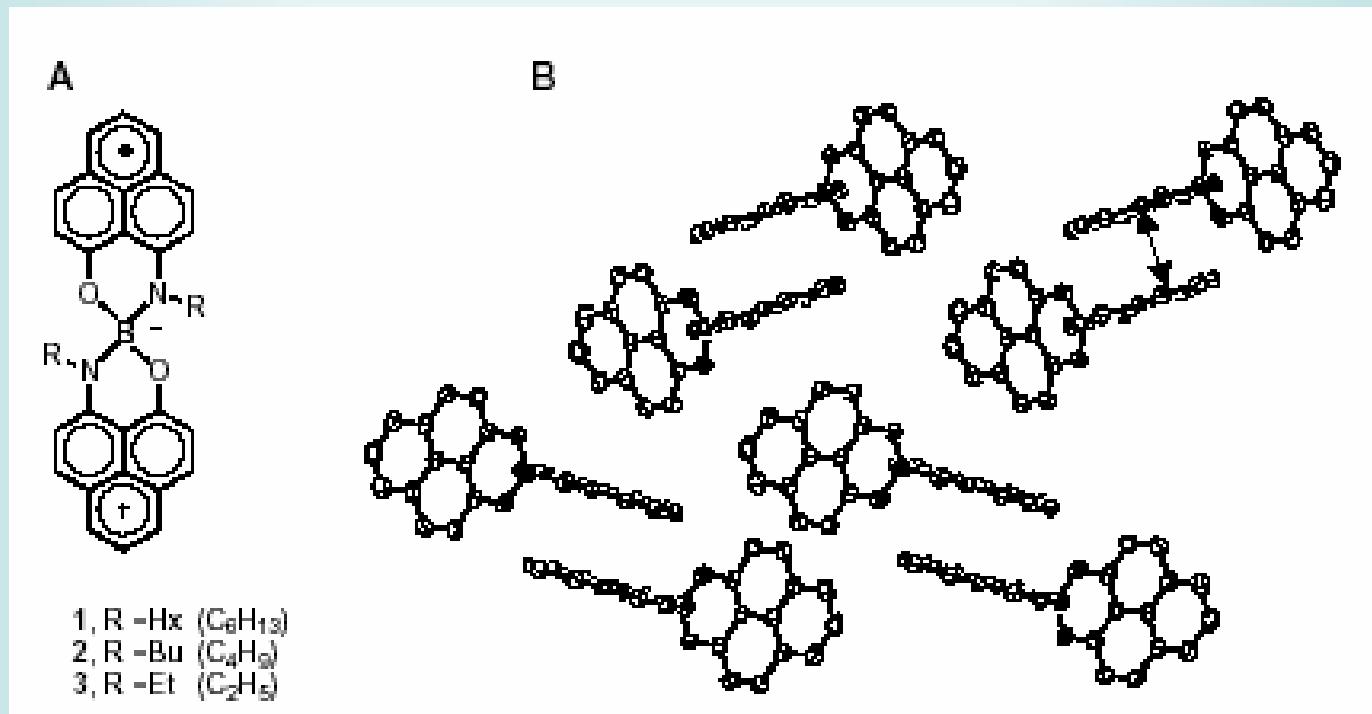


# Temperature Dependence of the Resistivity & Magnetic Susceptibility



Hisashi Tanaka,<sup>1</sup> Yoshinori Okano,<sup>1</sup> Hayao Kobayashi, Wakako Suzuki, Akiko Kobayashi\*  
*Science, 2001, 291, 285.*

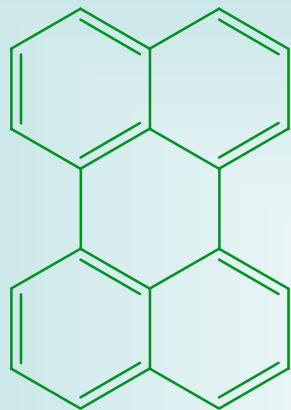
# Another Single-Component Conductor



M. E. Itkis, X. Chi, A. W. Cordes, R. C. Haddon, *Science*, **2002**, *291*, 1443.

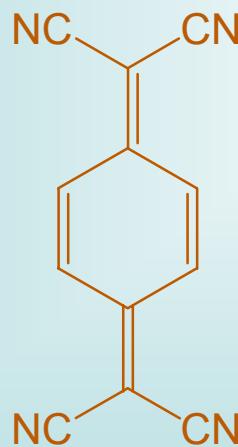
# Take Home Messages from this Lesson

1.



Perylene (Per)

facile oxidation = *p*-doping



TCNQ

facile reduction = *n*-doping

## Take Home Messages from this Lesson

2.

Need charge AND unpaired Spin to observe high conductivity

# Summary

Organic materials based on molecular solids, while showing conductivities as high as those of some traditional metals, also exhibit other very unusual properties.

The lessons learned from organic metals apply directly to conducting polymers

The End

Thanks!

# Discussion

# Coffee!



David Malloz

# The “Giant Conductivity Peak” Phenomenon

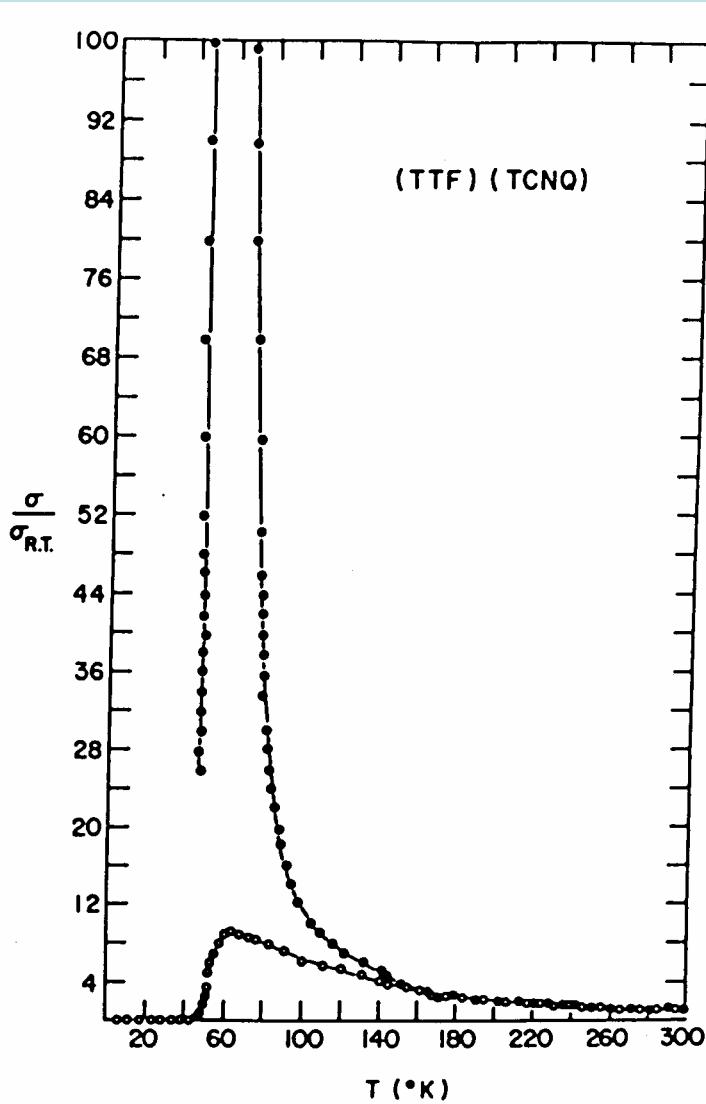


FIG. 3. Temperature dependence of the conductivity of (TTF)(TCNQ) single crystal ( $- \bullet - \bullet -$ ) and of (TTF)(TCNQ) typical crystals ( $- \circ - \circ -$ ).

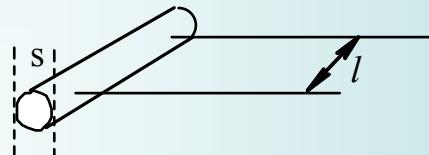
Coleman, et al, *Sol. State Commun.*  
1973, 12, 1125 - 1132

# The Measurement of Conductivity

$$R = \frac{V}{i}$$

$$R = \frac{\rho l}{s}$$

$$\sigma = \frac{1}{R}$$



R = resistance, V = Potential difference,  
i = current,  $\rho$  = resistivity,  $\sigma$  = conductivity,  
 $s$  = cross sectional area



